INSTITUTE OF CHEMICAL TECHNOLOGY

Ordinances, Regulations and Syllabi relating to the Degree of Master of Technology in Bioprocess Technology (M. Tech. Bioprocess Technology) 2018-2019

1. Introduction

The Institute is revamping its academic structure especially for the masters courses by way of introducing the compulsory industrial training for a period of six months (to be taken in the third semester of the program). The number of credits in the first two semesters has also been increased and a research component has been included. The total credits in the first two semesters now stand at 27 each instead of earlier 21. All the courses will continue to be credit based and the evaluation will be grade based.

The Departmental administrative committee and academic program committee periodically proposed the program outcomes having consistency with the graduate attributes available with NBA. The committee critically analysed information obtained from graduated students, employers and immediately passed out students. The program outcomes are as follows:

SR. NO.	PROGRAM OUTCOMES (POS)
1	The graduates will be able to apply knowledge of basic sciences (Mathematics, Physics, Chemistry, Biochemistry, Microbiology, Biology and Chemical Engineering Sciences) and applied engineering courses in getting solutions to issues pertaining to biotechnology, biochemical, biopharmaceutical and allied industries.
2	The graduates should be able to systematically break up complex processing problems in realizable steps and solve them.
3	The graduates will be able to design and develop a process, a product or a component of a biotech system or provide an engineering and technological solution for a specific task within realistic constraints
4	The graduates will be able to design and conduct experiments as well as analyze and interpret data.
5	The graduate will be able to use modern tools, software, equipment etc. to analyze and obtain solution to the problems.
6	The graduates will be able to study the impact of bioprocess industry in the global, economic, and societal context
7	The graduates should practice their profession considering environmental protection and sustainability
8	Graduates are expected to practice professional skills in an ethical manner
9	The graduates should have competence to undertake designated task on individual or team basis as per the requirement.
10	The graduates will be able to communicate effectively their points of view
11	The graduates will acquire attitude for life- long learning
12	The graduates should actively participate in project and financial management

SR.	PROGRAM SPECIFIC OUTCOMES (PSOs)
NO.	
13	Graduates will be acquainted with the latest development in different fields of bioprocessing so as to enable them to take up higher studies, research & developmental work
14	Graduates will be introduced to industrial bioprocessing and technology managerial subjects, so as to enable them to take up further studies in technology development, technology translation & function effectively as managers

Credit system is a systematic way of describing an educational programme by attaching credits to its components. The definition of credits may be based on different parameters, such as student workload, learning outcomes and contact hours. It is a student-centric system based on the **student workload** required to achieve the objectives of a programme. It should facilitate academic recognition of the courses and mobility of the students. Credits assignment is based on the principle that Credits can only be obtained after successful completion of the work required and appropriate assessment of the learning outcomes achieved. As per the AICTE norms 2L/week of lectures are 2 credits, while 2h/week of practical//seminar/literature review/research work are 1 credit. This has been taken as the basis during the working of the proposed syllabus.

Student workload consists of the time required to complete all prescribed learning activities such as attendance at lectures/practical, seminars, projects, etc. Credits are allocated to all the educational components of a study programme and indicate the quantity of work each component requires to achieve its specific objectives.

Evaluation is an important component of any teaching-learning process. The Institute gives emphasis on continuous evaluation with considerable freedom to the teacher in deciding the mode of evaluation of the students. The performance of the student is documented by a **grade** at the end of the semester. The grading scale ranks the students on a statistical basis. Therefore, statistical data on student performance is a prerequisite for applying the grading system.

2. Course Credits

In general, a certain quantum of work measured in terms of **credits** is laid down as the requirement for a particular degree. The student acquires credits by passing courses every semester, the amount of credit associated with a course being dependent upon the number of hours of instruction per week in that course.

There are mainly two types of courses in the Institute - lecture courses and laboratory courses. Lecture courses consist of lecture (L) and tutorial (T) hours. Laboratory courses consist of practical (P) hours. The credit (C) for a course is dependent on the number of hours of instruction per week in that course, as given below:

- (1) 1h/week of lecture (L) or tutorial (T) = 1 credit
- (2) 2h/week of Practicals (P) = 1 credit
- (3) Credit (C) for a theory course = No. of hours of lectures per week +
 - No. of hours of tutorials per week = L + T
- (4) Credits (C) for a Laboratory course/Seminar/research work =

½ x No. of hours per week

Credits will be assigned to In-plant, Seminar, Projects and other mandatory course requirements also and these will be mentioned in the respective syllabi. There may be some non-credit requirements. A student is required to earn credits as mentioned in the syllabus.

3. Evaluation

3.1 The weightage of different modes of assessments shall be as under.

	In-Semester	evaluation		
	Continuous mode	Mid Semester- Exam	End- Semester- Exam	Components of continuous mode
Theory	20%	30%	50%	Quizzes, class tests (open or closed book), home assignments, group assignments, <i>viva-voce</i> assignments, discussions
Practical	50%	-	50%	Attendance, <i>viva -voce</i> , journal, assignments, project, experiments, tests
Seminar/ critical review/ Research work	-	-	100%	Continuous evaluation not applicable, End semester evaluation will be based on written report evaluation and presentation in front of the external examiner within the Department

3.2. In-Semester Evaluation:

- (a) It is expected that the teacher would conduct at least two assessments (in any form as quizzes, tests, home work, group work etc) under the continuous mode in a Semester.
- (b) The teacher will announce at the beginning of the respective course the method of conducting the tests under the continuous mode and the assignment of marks
- (c) In-semester performance of all students should be displayed and sent to the academic office by the teacher at least 15 days before the end-semester examination.
- (d) For the theory courses, there will be one mid-semester test for each course to be held as per the schedule fixed in the Academic Calendar.
- (e) For mid –semester examinations in theory papers, duration of examination will be 1 hour for 3 credit courses and 2 hours for 4 credit courses

3.3. End-Semester examination:

- a) The semester end examination will cover the full syllabus of the course and will be conducted as per the Institutional timetable at the end of each semester.
- b) For end –semester examinations in theory papers, duration of examination will be 1 hour for 3 credit courses and 2 hours for 4 credit courses
- c) For the end semester evaluation of seminar/research work, student will be expected to submit a written report and make a presentation. The evaluation will be based on the quality of the written report and presentation.

3.4 Passes and Fail

- (a) The candidates who obtain 40% and more marks of the total marks of a course head shall be deemed to have **passed** the respective course head.
- (b) The candidates who obtain marks less than 40% of the total marks of a course head shall be deemed to have **failed** in the respective course head (**Grade FF**).

3.5 Grades:

- (a) The performance of a student shall be documented by a **Letter grade**. Each letter grade has a **Grade point** associated with it. The Grades and Grade points shall be assigned to each head of passing and both will be indicated in the mark-list of the semester examination.
- (c) The total marks (in-semester + end-semester) of a candidate in a subject head are converted into a letter grade, based on the relative (and some times the absolute) performance of the student.

Letter	Grade
Grade	Point
AA	10
AB	9
BB	8
BC	7
CC	6.5
CD	6
DD	5.5
EE	5

- (d) For granting class, a grade point of 6.0 and above will be considered equivalent to First class.
- (c) The grades to be allotted in the case of students who fail or do not appear at the end-semester examination shall be as under.

Letter	Grade	Explanation
Grade	Point	
FF	0	The candidate fails in course head. The candidate will be allowed to take end-
		semester repeat or subsequent examinations as per rule.
XX		The candidate has not kept term for the course head due to attendance less than
		requisite.
		Further see 3.5(g) below.
		In the above cases, the candidate has to repeat the respective course by paying the
		fees.
I	0	The candidate has kept term for the course head, has taken all the internal
		examinations with satisfactory performance, but has failed to take the end-semester
		examination or repeat examination due to genuine reasons. The candidate will be
		allowed to take end-semester repeat or subsequent examinations as per rule.
FR	0	The candidate has exhausted all the permissible chances to clear the end-semester
		examinations.
		The candidate has to register for the respective semester again for all the subject
		heads or will be out of the respective degree course as per the rules.
DR	0	(i) The candidate hasn't participated in academic programme.
		(ii) The candidate has taken a drop for the subject head;
		- provided he/she intimates the same (i or ii) at least 7 days in advance of the
		commencement of the end-semester examination for the respective year.

(d) Grades **FF** and **I** are place-holders only and do not enter into CPI/SPI calculations directly. These grades get converted to one of the regular grades after the end-semester examination.

- (e) A candidate with an **FR** grade is not eligible for any repeat examination in that course and has to re-register for that semester by paying the appropriate fees.
- (f) I grade will not be continued beyond the permissible number of end-semester/repeat examinations.
- (g) 'XX' Grade: The grade XX in a course is awarded if (i) candidate does not maintain the minimum 75% attendance in the Lecture/Tutorial/Practical classes, (ii) candidate receives less than 20% of the combined marks assigned for continuous assessment and mid-semester examination, and (iii) candidate indulges in a misconduct/uses unfair means in the examination, assignments, etc., of a nature serious enough to invite disciplinary action in the opinion of the teacher.
 - (**Note:** Award of the **XX** grade in the case of g(iii) above shall be done by Disciplinary Action Committee (DAC)).
- (h) The names/roll numbers of students to be awarded the **XX** grade should be communicated by the teacher to the Academic office as per academic calendar before the last date of submission of the application for end-semester examination.

3.6. Awarding the grades

The grading scale ranks the students on a statistical basis on the basis of the overall performance of the students of a given class in the given course head. Therefore, statistical data on students' performance is a prerequisite for applying the grading system. While assigning grades in a given course head, it is essential to know the **average marks (AM)** obtained by the students who have passed the subject head and the **highest marks (HM)** obtained in the same subject head.

- **3.6.1.** If the **average marks** (**AM**) obtained by the students *who have passed the subject head* is <60%, the interval AM shall be awarded grade CC and the other grades shall be decided as follows:
 - (i) AA, AB, BB, and BC grades shall be decided between the AM and HM by dividing the range in equal intervals.
 - (ii) CD, DD and EE grades shall be decided between the AM and minimum marks required for passing the head (i.e. 40%) by dividing the range in equal intervals.
- 3.6.2. If the average marks (AM) obtained by the students who have passed the subject head is such that 60% ≤ AM < 70%, the interval AM shall be awarded grade BC and the other grades shall be decided as follows:
- (i) AA, AB, BB grades shall be decided between the AM and HM by dividing the range in equal intervals.
- (ii) CC, CD, DD and EE grades shall be decided between the AM and minimum marks required for passing the head (i.e. 40%) by dividing the range in equal intervals.
- **3.6.3.** If the average marks (AM) obtained by the students who have passed the subject head is $\geq 70\%$, the interval AM shall be awarded grade BB and the other grades shall be decided as follows:
 - (i) AA and AB grades shall be decided between the AM and HM by dividing the range in equal intervals.
 - (ii) BC CC, CD, DD and EE grades shall be decided between the AM and minimum marks required for passing the head (i.e. 40%) by dividing the range in equal intervals.

4. SPI and CPI

(a) **Semester Performance Index (SPI):** The performance of a student in a semester is indicated by **Semester Performance Index (SPI),** which is a weighted average of the grade points obtained in all the courses taken by the student in the semester and scaled to a maximum of 10. (SPI is to be calculated upto two decimal places.) A Semester Grade Point Average (SGPA) will be computed for each semester as follows:

$$SOPA = \frac{\begin{pmatrix} n \\ \sum c_i g_i \\ i=1 \end{pmatrix}}{\begin{pmatrix} n \\ \sum c_i \\ i=1 \end{pmatrix}}$$

Where

'n' is the number of courses for the semester,

'ci' is the number of credits allotted to a particular course, and

'gi' is the grade-points awarded to the student for the course based on his performance as per the above table.

SGPA will be rounded off to the second place of decimal and recorded as such.

(b) **Cumulative Performance Index (CPI):** An up to date assessment of the overall performance of a student from the time he entered the Institute is obtained by calculating **Cumulative Performance Index (CPI)** of a student. The CPI is weighted average of the grade points obtained in all the courses registered by the student since he entered the Institute. CPI is also calculated at the end of every semester (upto two decimal places).

Starting from the first semester at the end of each semester (S), a Cumulative Grade Point Average (CGPA) will be computed as follows:

$$CGPA = \frac{\begin{pmatrix} m \\ \sum c_{i}g_{i} \\ i=1 \end{pmatrix}}{\begin{pmatrix} m \\ \sum c_{i} \\ i=1 \end{pmatrix}}$$

Where

'm' is the total number of courses from the first semester onwards up to and including the semester S,

'c_i' is the number of credits allotted to a particular course, and

'g_i' is the grade-points awarded to the student for the course based on his performance as per the above table. CGPA will be rounded off to the second place of decimal and recorded as such.

- (c) The CGPA, SGPA and the grades obtained in all the subjects in a semester will be communicated to every student at the end of every semester / beginning of the next semester.
- (d) When a student gets the grade 'FF', or I' in any subject head during a semester, the SGPA and CGPA from that semester onwards will be tentatively calculated, taking only 'zero' grade point for each such 'FF' or 'I' grade. When the 'FF' grade(s) has / have been substituted by better grades after the repeat examination or subsequent semester examination, the SGPA and CGPA will be recomputed and recorded.

5. Repeat End-Semester Examination

- **5.1.** For those candidates who fail in a subject head or are eligible for appearing at the repeat examination, **Repeat End-Semester Examination** will be conducted within one month from the declaration of the results of regular end-semester examination, as per **Regulation R.14**.
- **5.2.** The marks obtained by candidates in the in-semester examinations (continuous assessment and Mid-Semester Examination) will be carried forward in such cases.
- **5.3. Grading the performance in the Repeat Examination:** The grades will be assigned as per 3.5 and 3.6 above. However, for a candidate taking any repeat examination or subsequent regular semester examination or performance improvement examination shall be awarded **one grade lower** than that decided on the basis of the actual marks obtained; provided 'EE' grade obtained in such an examination shall remain 'EE'. For reference see the table below.

Grade obtained in repeat or subsequent end-semester examination	Grade to be assigned	Grade point
AA	AB	9.0
AB	BB	8.0
BB	BC	7.0
BC	CC	6.5
CC	CD	6.0
CD	DD	5.5
DD	EE	5.0
EE	EE	5.0

- 5.4. Revaluation of end-semester and repeat examination: Candidate's performance in these examinations will be displayed on proper notice board and after 3 days of such display the marks will be sent to the Academic Office. No revaluation of these examinations will be allowed.
- 6. Passing of a Semester examination

A candidate shall be declared as 'PASSED' any semester examination if he/she has

- (a) Cleared all heads of passing by securing grades EE or higher in all the heads;
- (b) Passed all the heads of passing such as project, seminar, training, etc as per the rules;
- (c) Satisfactorily completed all the mandatory requirements of the course;
- (d) paid all the Institute dues;
- (e) No case of indiscipline pending against him/her.

7. Eligibility for the Award of a Degree

A candidate shall be declared eligible for the award of a degree, if he/she has cleared all the semester examinations as given in (6) above.

8. Allowed to keep terms (ATKT)

- 8.1 A candidate who has I grade in one or more heads of passing of an odd semester of an academic year shall be allowed to keep terms for the respective even semester.
- 8.2. A candidate shall be allowed to keep terms for the subsequent academic year if he/she has FF or I grades in not more than two heads of passing from all the heads of passing of the two terms of the previous academic year taken together. Such a candidate shall be declared as **FAILED**, **ATKT**.

9. Repeating a course

- **9.1** A student is required to repeat the course under the following situations:
 - (a) A student who gets an **XX**, **FR**, or **DR** grade in a course; or
 - (b) A student has exhausted all permissible chances to clear the course.
- **9.2** A candidate from first year who remains absent for the regular end-semester examination of a semester and the corresponding repeat examination for **ALL SUBJECTS** shall have to take fresh admission for the corresponding year; unless the candidate has dropped out / terminated from the course.
- **9.3** If a candidate at the Second, fails to pass any semester examination in not more than 4 consecutive examinations, including the repeat examinations, from the date of registering for the respective year, the candidate shall have to take readmission for the corresponding year again in which the failure has occurred, provided the course is not changed.

10. Improvement of performance

A candidate will be allowed to appear at the **entire examination** after the regular end-semester examination as per the respective rules to improve the performance. In such a case if the result of the examination repeated –

- 1. Is better than the previous one, the previous result shall be declared null and void; and
- 2. Is worse than the previous one, the result of the subsequent examination shall not be declared.
- 3. However, awarding of final grade will be made under the provision of sub clause 5.3 above.

11. Exit rules for poorly performing students

A candidate shall be excluded from a course under the following conditions:

- (a) If he/she fails to pass any semester examination of the any year of the course in not more than four consecutive attempts (Examination conducted by Institute) from the date of joining the course.
- (b) If he/she does not keep two consecutive terms without giving any reasonable justification (as prescribed by the institute) for doing so.
- (c) If a candidate fails to fulfill all the requirements of his/her respective degree within the prescribed period from the date of taking admission to the course, the candidate shall be excluded from the course.

12. Miscellaneous

- (a) Although CPI will be given in the Semester grade report, the final degree certificate will not mention any **Class** whatsoever.
- (b) Not withstanding anything said above if a course is revised /restructured then transient provisions applicable at the time of revision /restructuring shall be applicable.

Syllabus Details for the degree of Master of Technology (Bioprocess Technology) Program

C-1-14	whicat Hr/Week Marks								
Subject code	Subject	Credit	L	Т	P	Continuous Assessment	Mid-semester Examination	Final Examination	Total
					SEN	MESTER I			
BST	Core I: Bioreaction	3	2	1	0	10	15	25	50
2101	Engineering			1	U	10	13	23	30
BST	Core II: Unit	2		1	0	10	1.5	25	50
2102	Operations in Bioprocessing	3	2	1	0	10	15	25	50
BST	Core III: Industrial								
2103	Biocatalysis	3	2	1	0	10	15	25	50
	Elective I	3	2	1	0	10	15	25	50
	Elective II	3	2	1	0	10	15	25	50
BSP2101	Bioprocess Engineering Laboratory	3			6	25	-	25	50
BSP2102	Seminar and Critical Review	3			6	-	-	30 (Report) 20 (Presentation)	50
BSP2103	Research Project-I	6			12	-	-	60 (Report) 40 (Presentation)	100
	TOTAL:	27	10	5	24				450
			1						
				5	SEM	IESTER II	T		1
BST2104	Core IV: Bioprocess and Biosystem Engineering	3	2	1	0	10	15	25	50
BST2105	Core V:Bioreactor Design and	3	2	1	0	10	15	25	50
BST2106	Core VI: Adsorptive, Chromatographic and Membrane separations	3	2	1	0	10	15	25	50
	Elective III	3	2	1	0	10	15	25	50
	Elective IV	3	2	1	0	10	15	25	50
BSP2104	Biosciences and Bioprocess Technology Laboratory	3			6	25		25	50
BSP2105	Research Project-II	9			18	-	-	90 (Report) 60 (Presentation)	150
	TOTAL:	27	10	5	24	-	-	-	450
BSP2106	Industrial Training (15 weeks to maximum of 6 months)	30	-	-	40	ESTERS III		270 (Report) 180 (Presentation)	450
	Research Project-III			S		ESTER IV	T	270 (Report)	
BSP2107	Research Project-III	30	-	-	40	-	-	180 (Presentation)	450

Note: Semester III and Semester IV Evaluation will be conducted be at end of IV semester.

SEMESTER - I

	Course Code: BST 2101	Course Title: BST 2101 Bioreaction Engineering (Marks 50)	Cred	dits = 3		
	(Core subject)		L	T	P	
	Semester: I	Total contact hours: 30	2	1	0	
		List of Prerequisite Courses				
		and chemical kinetics, Simple design methods, graphicarison of capabilities of the major reactor types.	1			
	List of C	Courses where this course will be prerequisite				
		chnology, Biotechnology, Biochemical Engineering	,,			
Desci		g, Chemical Technology this course in the M. Tech. (Bioprocess Technology)	ov) P	rograr	nme	
application biochemic processing	ons in biotechnology (fe cal, agroprocessing, nat	es, material and energy balance etc. They will be abstraction, biocatalysis, biotransformations etc.), bioural product, nutraceutical, edible oil, flavor and an industries according to the biochemical reaction involves. Course Contents (Topics and subtopics)	ophar fragra	maceut ance, f	ical,	
Module						
1	Material and Energy E			5		
2	Basic Biochemistry, Basic Microbiology and Basic Molecular Biology, Principles of biochemical reactions and kinetics					
3	<u> </u>	ioreactions and biotransformations		5		
4	Unstructured and si morphologically structured	mple structured models, Mechanistic models tured models	and	5		
5	Phase equilibria in systems Calculation COSMO-RS (exercise	multicomponent systems Partioning in biorele of phase equilibria in colloidal systems: UNIF ses in computer pool) Calculation of partitionical membranes: COSMO-RS (exercises in comp	AC, ning	5		
6		ions of state (vapour pressure, phase equilibria, er pool) Intermolecular forces, interaction Potenical thermodynamics		5		
	Li	st of Text Books/ Reference Books				
	 2.Chemical Engineering 3.Elements of Chemical 4.Basic Biotechnology University Press 2 5.Biochemical Engine 6.Bioreacation Engine 7.Bioprocess computation 8.Advanced Biochemical 	ral Reaction Engineering: H.Scott, Fogler. y, edited by Colin Ratledge and Bjorn Kristiansen, C 003. rering Fundamentals, Bailey, and Ollis, McGraw Hil rering, K. Schergeri, Vols 1 & 2, John Wiley. 1985. rtions in Biotechnology, T.K. Ghosh, Ellis Horwood rical Engg., ' Henry R. Bugay Georgs Belforj, John V	ll Boo	ok Co.1	s, 1988.	
	9. Lehninger, Biocher	mistry, 4 th edition, 2005 S. Chase and N.R. Kreigh, "Microbiology", 4 th Edit	•			

	McGrawhill, India.
	11. P.A. Ketchum, "Microbiology", John Wiley and Sons, New York, 1984.
	Freifelder D., "Molecular Biology", Jones and Bartlett Publishers Inc., 1987
Course (Outcomes
1	Able to understand the basics of basic knowledge on enzymatic and microbial kinetics
2	Able to understand the application of thermodynamics in bioprocesses
3	Understand the mechanistic involved in biochemistry, microbiology and molecular biology
4	Understand phase equilibria and intermolecular forces
5	To find out the applications of thermodynamics and intermolecular forces in bioprocessing e.g. protein stability, conformations, affinity interactions etc.

	Course Code:	Course Title: BST 2102	Cre	Credits = 3	
	BST 2102	Unit operations in Bioprocessing			
	(Core subject)	(Marks 50)	L	T	P
	Semester: I	Total contact hours: 30	2	1	0
	l	List of Prerequisite Courses		1	
	Physicochemical Prop	erties of biochemical's, Transport phenomenon, b	iochemistry		
	List of (Courses where this course will be prerequis	site		
	PhD in Bioprocess To	echnology, Biotechnology, Biochemical Engir	neering,		
	Chemical Engineerin	g, Chemical Engineering operations, Pharmac	eutical		
	Biotechnology, Bioar	nalyticals			
Desci	rintion of relevance of	this course in the M. Tech. (Bioprocess Tec	chnology) P	rogran	ıme

Description of relevance of this course in the M. Tech. (Bioprocess Technology) Programme

Students will understand the unit operations, specifically downstream processing operations like cell harvesting, cell lysis, primary product recovery and purification, polishing etc. They will be able to explain its applications of different unit operation in bioprocessing for purification of biomolecules and their scale up in industries according to the separation methods involved.

Module	Course Contents (Topics and subtopics)	Reqd. hours
1	Downstream Processing in Biotechnology, Selection of unit operation with due consideration of physical, chemical and biochemical aspect of biomolecules, basic review of bioprocess designing.	5
2	Primary separation and recovery processes: Cell disruption methods for intracellular products, removal of insolubles, biomass (and particulate debris) separation techniques, flocculation and sedimentation, centrifugation and filtration methods.	5
3	Enrichment operations: Membrane – based separations (micro and ultrafiltration, precipitation methods, extractive separation, aqueous two-phase extraction, supercritical extraction, insitu product removal, integrated bioprocessing.	6
4	Product resolution / fractionation: Introduction to adsorptive chromatographic separations processes, electrophoretic separations, hybrid separation technologies (electrochromatography).	5

	Product finishing: precipitation/crystallization, mixing, dialysis, distillation	5
5	and drying. Ultracentrifugation as a separation technique for fractionation of	J
	cells and proteins.	
	Introduction to Process Analytical Technology (PAT) and Quality by Design	4
6	(QbD). Scale down, monitoring and Validation of bioprocesses	
	List of Text Books/ Reference Books	
1	Encyclopedia of Bioprocess Technology, Vol. 1-5, 1999	
2	Scopes Ak, Protein Purification, IRL Press, 1993	
3	Biotechnology: Bioprocessing, Rhem and Reed, Vol. 3, 1993	
4	Separation and purification techniques in biotechnology, Fredreich Dechow, 19	89
5	Coulson J.M. and Richardson, J.F. "Chemical Engineering, Vol.2 Unit Opergamon Press (1978).	erations, Ed.3,
Course (Dutcomes	
1	Able to understand the basic physicochemical properties of various biomolecule	es
2	Able to understand the basics of various unit operations in bioprocessing of bio	molecules
3	Able to understand the process integration of with various unit operations	
4	Able to understand the process optimization with modern strategies	
	To find out the applications of various unit operations in bioprocessing specific	
5	downstream processing of proteins, enzymes, antibiotics, vitamins, amino acids	
3	acids, agro-products, biochemicals, glycerides, beverages, food and nutrition, n products, biofuels, organic acids etc.	atural

	Course Code:	Course Title: BST 2103	Cred	dits = 3	3
	BST 2103	Industrial Biocatalysis-I			
	(Core subject)	(Marks 50)	L	T	P
	Semester: I	Total contact hours: 30	2	1	0
L		List of Prerequisite Courses		I	
		<u>-</u>			
	Bioreaction Engineering				
	List of Cour	ses where this course will be prerequisite			
	PhD in Bioprocess Techno	ology, Biotechnology, Biochemical Engineerin	g,		
	Chemical Engineering, Cl	nemical Engineering operations	-		
Descrip	ption of relevance of this	course in the M. Tech. (Bioprocess Technological Course)	ogy) P	rogran	nme

etc. in the development of biocatalysis based industries will be covered in this course.

Students will understand the state of art and future prospects of industrial biocatalysis. They will be able to explain its applications for synthesis of bio-based materials, and compounds as intermediates for bioproducts and biofuels. The role of recombinant technology, synthetic biology, protein engineering, pathway engineering, metabolic engineering, systems biology, fermentation biology, algal biotechnology

Module	Course Contents (Topics and subtopics)	Reqd. hours
1	Biocatalysts, aspects of biocatalytic process design and development, steps from laboratory to industrial scale, applications, global market and societal challenges	2
2	Biocatalysis using natural enzymes, microorganisms (bacteria, yeast, fungi), eukaryotic cells (insect, CHO, algae, plant cell cultures), plants, and recombinant enzymes engineered for specific applications, Catalytic activity of biomolecules – enzymes and ribozymes; Enzyme applications: Hydrolase enzymes – lipases, esterases, proteases etc. with specific examples and mechanism, Lyases – e.g. Aspartase, tyrosine-phenol lyase; Isomerases – e.g. glucose isomerise; Transferases – e.g. aminotransferases, PLP as cofactor; Ligases; Oxidoreductases – dehydrogenases, oxidases, oxygenases, peroxidases.	4
3	Enzyme structure-function relationships, Thermodynamics of protein folding and substrate binding, , Enzyme kinetics and modes of inhibition; Regulation mechanisms; Mechanism of enzyme action; Multienzyme systems; Selection and screening of biocatalysts for activity, stability and substrate or product selectivity; Extremozymes – biocatalysts at extremes of temperature, pressure and pH.	6
4	Biocatalysis versus chemical catalysis; Understanding when to use a biocatalyst for a chemical problem; Advantages/disadvantages of biocatalysts compared to traditional chemical reactions and heterogeneous/ homogeneous catalysis; Mild reaction conditions, excellent stereo- chemo- and regio- selectivity versus substrate specificity, product inhibition, lack of catalysts robustness, cofactor recycling; Isolated enzyme systems and whole cell systems. Free and immobilized enzymes for biocatalysis. Water versus organic solvent; Reactor and process technology: types, mass balances and their modes of operation; Biocatalyst recycling and recovery; Enzyme immobilization.	6
5	Enzymes in organic synthesis, Enzymes in novel media, Green chemistry, Oxidation catalysis, Catalysis in water, Homogeneous catalysis, Heterogeneous catalysis, Asymmetric catalysis	6
6	Modern branches of biotechnology as the workhorse for biocatalysts design improvement, Synthetic biology and biocatalyst engineering, Enzyme discovery and metagenomics, enzyme engineering strategies Chassis selection and host cell engineering, Practical enzyme characterisation, Industrial availability in enzyme production, Scale-up challenges, Process economics and sustainability	6
	List of Text Books/ Reference Books	
1	Nelson, D. L. And M.M. Cox (2005), Lehninger, Principles Of Biochemistry. W. H. Freeman And Company.	4th Edition,
2	Conn, E.E., P. K. Stumpf, G. Bruening and R. Y. Doi. (1987) Outlines Of Bioch Edition, 1987. John Wiley & Sons. New York.	hemistry, 5 th
3	Grunwald P, Biocatalysis- Biochemical Fundamentals And Applications (2017) IS 1783269082	SBN-13: 978-
4	K. Buchholz, V. Kasche and U. T. Bornscheuer (2005) Biocatalysts and Enzym Wiley VCH Verlag GmbH and Co.	ne technology
5	Gaikar V G, ()Biotransformations And Bioprocesses (Biotechnology and Bioproce ISBN-13: 978-0824757137	essing Series)
6	Grunwald P, (2014) Industrial Biocatalysis, Edited by Peter Grunwald, Pan Sta ISBN 9789814463881	nford
7	Copeland R A, (2000) Enzymes: A Practical Introduction to Structure, Mechani Analysis, Second Edition, Wiley VCH, ISBN: 978-0-471-35929-6	ism and Data
Course O	Outcomes	
1	Able to understand the fundamentals of Biocatalysis, principles of designing bio	ocatalysts

2	Basic biochemistry of enzymes and their mechanisms, types of reactions catalysed
3	Understand the challenges and solutions associated with conducting biocatalysis and different modes of carrying out these reactions
4	Understand the methods, tools and strategies for modifying biocatalysts
5	Applications for biocatalysis in different areas

Course Code:	Course Title: (Elective I) (Marks 50)	Cre	dits = 3	3
		L	T	P
Semester: I	Total contact hours: 30	2	1	0

Elective-I (from the list appended)

Candidate will have to choose one of the elective subjects offered for that semester from the elective subjects. A consolidated list of all the elective subjects is given at the end.

Course Code:	Course Title: (Elective II) (Marks 50)	Cre	dits = 3	3
		L	T	P
Semester: I	Total contact hours: 30	2	1	0

Elective-II (from the list appended)

Candidate will have to choose one of the elective subjects offered for that semester from the elective subjects. A consolidated list of all the elective subjects is given at the end.

	Course Code: BSP 2101	Course Title: BSP 2101 Bioprocess Engineering Laboratory	Cred	lits = 3	
		(Marks 50)	L	T	P
	Semester: I	Total contact hours: 30	-	-	6
	1	List of Prerequisite Courses			
	Bioreaction Engineeri	ng, unit operations in bioprocessing			
	List of	Courses where this course will be prerequisite	,		
	PhD in Bioprocess T	echnology, Biotechnology, Biochemical Enginee	ering,		
	Chemical Engineerin	g, Chemical Engineering operations	_		
Desci	ription of relevance of	this course in the M. Tech. (Bioprocess Tech	nology) P	rogran	ıme
		plain the application of various chemical and bioroducts (both small and large biomolecules).	chemical e	engineer	ring
Module		Course Contents (Topics and subtopics)		Reqd	. hou

1	Flow through pipes, coils and fittings. Flow meters, orifice, venturi, rotameter and turbine meter. Flow through packed beds. Two phase flow. Sedimentation. Fluidization. Solid-liquid separation. Mixing. Evaporators. Absorption in a packed column. Adsorption isotherms. Drying characteristics. Study of spray nozzles, impellers, tower packings, dryers, filters, evaporators. Demonstration of some phenomena, particularly in mixing, fluid mechanics, etc.	
2	Absorption with and without chemical reactions in packed columns. Distillation in packed and/or plate column. Spray, packed and mechanically agitated extraction columns. Absorption/ion exchange in fixed beds. Separation by membranes. Flow of non-Newtonian fluids. Dynamics of feedback control systems. Level and pH control. Demonstration of some important phenomena in bioprocess Engineering, notably coalescence, foaming, internal circulations in drops and bubbles, two and three phase fluidization, aggregative and particulate fluidization, mixing, crystallization etc.	
3	Suitable number of experiments from the above list will be performed. In addition to these experiments, students will also undertake demonstration experiments related to advanced analytical instruments such as GC, HPLC, GC-MS, LC-MS, SEM, FTIR, UV-Vis Spectrophotometry, NMR, TEM, ICP, particle size analyzer etc. In this student will work in groups on these instruments to make a report on theory, working principle, standard operating procedure and one case study as well as live demonstration at the end of laboratory session.	
Course C	Outcomes	
1	To be able to perform experiments to ascertain the governing bioprocesses and engineering principles	biochemical
2	To design experimental configurations/loops for demonstrating the basic gover fundamentals	
3	To perform calculations and predict the trends in design variables as a function operating/geometric parameters	of
4	To effectively operate the analytical facilities	

Course Code: BSP 2102	Course Title: BSP 2102 Seminar and Critical Review	Cred	lits = 3	
	(Marks 50)	L	T	P
Semester: I	Total contact hours:	-	-	6
Bioprocess Technolog characterization, Molecular be evaluated based on the (a) Introduction: 2 pages of (b) Exhaustive review of 1 (c) Critical analysis of the should also contain quamongst the various paramongst the various para	maximum, literature (including figures): $10 - 12$ pages: 50% the literature and comments on the analysis Critical comparison of observations, results, and	Weightantical analytical analytic	ge alysis usion to be railed 1/2 page. hard Each	

- figures must be sufficiently clear and hand drawn figures will be acceptable. Particular care must be taken if a figure is photocopied from source. Each figure must have a sequence number and caption below. Each table must have a sequence number and title at the top.
- **4.** Name of the student, title of the problem and year of examination must be indicated on the top cover. THE NAME OF THE SUPERVISOR (ONLY INITIALS) MUST APPEAR ON THE BOTTOM RIGHT CORNER OF THE TOP COVER.
- 5. The report must be precise. All important aspects of the topic should be considered and reported. The total number of pages, including tables, figures, and references should not exceed 30. Chapters or subsections need not be started on new pages, while getting the report typed.
- **6.** Typographical errors in the report must be corrected by the student. The student will be discredited for any omission in the report. All the symbols used in the text should be arranged in an alphabetical order and given separately after conclusions.
- 7. The list of references should be arranged in alphabetical order of the names of authors. In the text, the reference should be cited with author's name and year. (author date style) For example:
- (i) The flow pattern in gas-liquid-solid fluidized bed has been reported in the published literature (Murooka et al., 1982).

OR

- (ii) Murooka et al. (1982) have measured flow patterns in gas-liquid-solid fluidized beds. The title of the article should also be included. The references must be given in the following standard format.
- (a) Format for listing references of articles from periodicals: Murooka S., Uchida K. And Kato Y., Recirculation Turbulent Flow of Liquid in Gas-Liquid-Solid Fluidised Bed", J. Chem. Engg. Japan, 15, 29-34 (1982).
- (b) Format for listing references of Books: Constant R.F.,"Crystallization, Academic Press, New York, pp. 89-90, 1968.
- (c) Format for listing Thesis:
 Niranjan K., "Hydrodynamic and Mass Transfer Characteristics of Packed Columns", Ph.D. (Tech.) Thesis, University of Mumbai, 1983.
- (d) Format for listing references of Patents in Chemical Abstracts: Cananaush R.M., U.S.Patent 2,647,141, Cf. C.A. 48, 82636 (1954).
- (e) Format for listing Handbooks, Tables, Symposia etc.: Kumar R and Kuloor N.R., "Formation of Drops and Bubbles", in Advances in Chemical Engineering, Vol.8, T.B. Drew et.al. (Eds.) New York, Academic Press, pp.256-364 (1970).
- (f) Format for listing Private Communications and other categories: Sharma, M.M., Private Communication (1984).
- **8.** Consistency of units should be maintained in the written report. SI systems should be used. [For SI system Ref: Ind. Chem. Engr., 24, 32, 3 (1983)]. Units used in the literature (if not SI) should be correctly converted.
- **9.** The time allotted for the oral presentation of seminar is 20 minutes: additional 10 minutes are provided for questions and answers.
- **10.** INCOMPLETE AND CARELESSLY WRITTEN REPORT IS LIABLE TO BE REJECTED.
- **11.** The last date for submission will NOT be extended on any grounds whatsoever.
- **12.** There must not be any acknowledgment about the guidance by the faculty in the Seminar.
- 13. The Seminar will be evaluated on the basis of (i) rational approach to the problem, ii) correctness and completeness of the written text and iii) performance in the oral presentation.
- **14.** Word-to-word copying from the published article is not permitted. Flowery language is not to be used.

The submitted report will be evaluated by the research guide and an external examiner

	from the Department/Industry based on the presentation made by the candidate. A suitable combination of the marks for report and presentation will be considered for the final evaluation.
Course O	utcomes
1	Survey literature related to the given topic
2	Analyze the reported outcomes and classify the work under key categories
3	Write a technically correct report as per the suggested guidelines and present the seminar work

	Course Code: BSP 2103	Course Title: BSP 2103 Research Project -I	Cree	dits = 3	3
		(Marks 100)	L	T	P
	Semester: I	Total contact hours:	-	-	12
Course	area in consultation with experimental/analytical the outcomes of the camprovided for BSP2102 are examiner from the Department of the consultation with the outcomes of the camprovided for BSP2102 are assumed to the consultation with the outcomes of the camprovided for BSP2102 are assumed to the consultation with the outcomes of the camprovided for BSP2102 are assumed to the consultation with the camprovided for BSP2102 are assumed to the camprovided f	is concerned with detailed literature review of the the research supervisor, developing an simulation protocol and initiate the actual research didate is expected to submit a report as per simulation which will be evaluated by the research grattment/Industry based on the presentation made of the marks for report and presentation will be	arch work. Base ilar guidelines guide and an ext e by the candid	ed on ernal ate.	
1	Analyze existing literate	ure for research topic and develop detailed plan	of experiments	/ simula	ations
2	Systematically perform	experiments/modeling activity to accomplish th	he set objectives	s	
3	Critically analyse the re present the work	sults and write a technically correct report as pe	er the suggested	guidel	ines and

SEMESTER II

	Course Code: BST 2104	Course Title: BST 2104 Bioprocess and Biosystem Engineering	Cred	lits = 3	3
	(Core subject)	(Marks 50)	L	T	P
	Semester: II	Total contact hours: 30	2	1	0
		List of Prerequisite Courses		•	
	Biochemical reactions, e	nzyme catalysis, nucleic acid, proteins, and metabolite	es		
	List of Co	ourses where this course will be prerequisite			
		hnology, Biotechnology, Biochemical Engineerin	g,		
Dogg	Chemical Engineering,	Chemical Technology nis course in the M. Tech. (Bioprocess Technology)	ogy) D	rogran	nmo
native probiotechnology	oducts, bioinformatics and blogy (fermentation, biocessing, natural product, n	anistic of cellular systems, pathway engineering for modeling of bioreactors. They will be able to explain atalysis, biotransformations etc.), biopharmaceuti utraceutical, edible oil, flavor and fragrance, foo- rding to the biochemical reaction involved.	its app cal, bi	lication ochemi	ns in ical,
Module		Course Contents (Topics and subtopics)		Reqd	l. hours
1	Thermodynamics of Bios	ystems		6	
2		Metabolism and Principles of Metabolic flux and ing, Cybernetic principles of optimal growth modeling		6	
3	Biochemical pathway er metabolic and gene biocatalysts/bioproducts/v		rough new	6	
4		esign of cellular systems: Integration of recomb design, as well as bioinformatics and process sy		6	
5	Basic principles of Syster	m and Synthetic biology and modeling of bioreactors		6	
	List	t of Text Books/ Reference Books			
		tems Engineering, 1 edition, McGraw-Hill, Inc, 20	009		
<u> </u>	2. Biosystem Engineeri	ng Journal, Elsevier			
Course C	Jutcomes				
1	Able to understand the improvement of biosyst	basics of basic knowledge on genetic and metabo	lic eng	ineerin	ig for
2		application of thermodynamics in molecular aspe	cts in o	designi	ng of
3		istic involved in biochemistry, microbiology and	molec	ular bio	ology
4	Understand phase equil	ibria and intermolecular forces			
5		ions of thermodynamics and intermolecular force informations, affinity interactions etc.	s in bio	oproces	ssing

	Course Code: BST 2105	Course Title: BST 2105 Bioreactor Design and Industrial	Cred	lits = 3	3
	(Core subject)	Bioprocess Automation (Marks 50)	L	T	P
	Semester: II	Total contact hours: 30	2	1	0
	1	List of Prerequisite Courses		l	
	Basic knowledge of biokinetics, and basic math	ological processes, cellular metabolism, enzyme and micr hematics	robial		
	List of C	Courses where this course will be prerequisite			
		chnology, Biotechnology, Biochemical Engineering	<u>,</u>		
Descr		g, Chemical Technology this course in the M. Tech. (Bioprocess Technolog	gv) Pr	rogran	nme
biopharm fragrance	l also learn bioreactor mod naceuticals, biochemicals, c, food processing and p	ioreactors, their design and operation for production of be deling, process optimization as well as continuous Bioman agroprocessing, natural product, nutraceutical, edible othermaceuticals etc. industries according to the bioch	nanufa e oil,	cturing flavor	g for and
biopharm	l also learn bioreactor mod naceuticals, biochemicals, c, food processing and p	deling, process optimization as well as continuous Biom, agroprocessing, natural product, nutraceutical, edible	nanufa e oil,	cturing flavor al reac	g for and tion
biopharm fragrance involved.	Background of bioreactor, and continuous bioreactors, Loop b bioreactor, Packed-be	deling, process optimization as well as continuous Biom, agroprocessing, natural product, nutraceutical, edible pharmaceuticals etc. industries according to the bioch	fed- bed umn	cturing flavor al reac	g for and
biopharm fragrance involved. Module	Background of biorea batch, and continuous bioreactors, Loop bioreactor, Packed-be fermenter, Multiphas bioreactor). Design of Stirrers as bioreactors and phot	deling, process optimization as well as continuous Bioman agroprocessing, natural product, nutraceutical, edible charmaceuticals etc. industries according to the biocher Course Contents (Topics and subtopics) actors, Modeling and Design of bioreactors: batch, as flow types (Airlift bioreactors, Airlift pressure concreactor, Stirred tank bioreactors, Fluidized ed reactors, Trickle bed bioreactor, Bubble coluse bioreactors, Disposable bioreactors and World impellers. Design, development and scale up obioreactors for production of antibiotics, enzyriproducts and biofuels. Reactors with non ideal mix	fed- ycle bed umn Vave	cturing flavor al reac Reqd	g for and tion

List of Text Books/ Reference Books

automatic processes). Continuous Biomanufacturing processes

sludge and fixed-film systems. Solid state, Surface, submerged and anaerobic

Principles and Strategies for Control of Bioreactors (feedback, feedforward,

adaptive and statistical control, fuzzy logic control), of bioreactors and

Industrial bioprocess and their process automation (batch, semiautomatic and

fermentation, Sterilization and asepsis.

ancillary equipment.

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- 1. Najafpour, G. D., "Biochemical Engineering and biotechnology", Elsevier, 2007.
- 2. Doran, P.M., "Bioprocess Engineering Principles", Academic Press, 2005.
- 3. Walker, J.M. and Rapley, R., "Molecular Biology and Biotechnology", 4th Edition, Royal Society of Chemistry, 2000.

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4. Blanch, H. W. and Clark, D. S., "Biochemical Engineering", Marcel Dekker, Inc., 1999.

	Dunn, I.J., Heinzle, E., Ingham J. and Prenosil, J.E., "Biological Reaction Engineering: Dynamic Modeling Fundamentals with Simulation Examples", 2 nd Edition, Wiley-VCH, 2003.
Course (Outcomes
1	Able to design and analyze batch, continuous flow, and fed batch reactors with specific instrumentation required for the efficient monitoring and control of simple bioreactor, ancillary equipment required for the aseptic feeding, sampling and processing of bioreactor fluids
2	Able to understand the application of design in biological reactors with cell recycle streams
3	Students should be able to apply the reactor optimization principles for the design of bioreactors for industrially important biological products, primary and secondary metabolites
4	Understand usefulness of automatic and continuous biomanufacturing
5	To find out the applications of bioreactor design and continuous processing for various biotech products

Course Code: BST 2106		Credits = 3			
(Core subject)	Membrane Separations (Marks 50)	L	T	P	
Semester: II	Total contact hours: 30	2	1	0	
1	List of Prerequisite Courses				
	, unit operation in bioprocesses, plug flow and fluidiz aration and purification	zed bed			
	-				
List of (Courses where this course will be prerequisite	:			
PhD in Bioprocess To	echnology, Biotechnology, Biochemical Enginee	ering,			
Chemical Engineerin	g, Chemical Technology				
	4hia aayyaa iy 4ha M. Taab (Diammaaaa Taaby	. 1 \ D			

Description of relevance of this course in the M. Tech. (Bioprocess Technology) Programme

Students will understand various adsorption of biomolecules and preparative chromatographic separations. They will also learn membrane filtration operations for various biomolecules including process optimization and integration for manufacturing for biopharmaceuticals, biochemicals, agroprocessing, natural product, nutraceutical, edible oil, flavor and fragrance, food processing and pharmaceuticals etc. industries according to the biochemical reaction involved.

Module	Course Contents (Topics and subtopics)	Reqd. hours
1	Introduction, Theory and chemistry of adsorption. Chromatographic Fundamentals: Retention, Band Spreading, Resolution; Dynamics of Chromatography: Basic mass transfer equations, Method of moments, Linear dispersion model, Linear staged models for chromatography; Instrument Requirements for Chromatography: System design, Column packing techniques; Fundamentals of Adsorption: Gibbs adsorption isotherm, Adsorption isotherm models, Local equilibrium theory and solute movement plots;	
2	Preparative Chromatography: Preparative elution, Frontal, Gradient, Displacement chromatography, Optimization; Hydrodynamic design of adsorbent: Particle size, pore size, surface area and pore volume etc. Thermodynamic design of adsorbent: Ligand design through Molecular modeling, retention mechanisms;	6

3	Modes of Chromatography: Reversed phase and hydrophobic interaction, Ion exchange and Ion exclusion, Size-exclusion, Group specific and biospecific affinity, IMAC, Supercritical fluid chromatography; Isocratic and Gradient Elution preparative chromatography; Mode of contacting solids with liquid: Packed bed, expanded bed, fluidized bed, moving bed (Simulated moving bed, True moving bed, Liquid-solid circulating fluidized bed, Fluidized moving bed);	6			
4	Novel Chromatographic Morphologies: Continuous annular systems, Radial flow, centrifugal chromatography, SMB, ISMB, Continuous chromatography (PCC and varicol systems), Perfusion chromatography, Membrane chromatography and Monoliths; Chromatographic Applications in Biotechnology: Applications of various modes of operation, sequencing of chromatographic operations, Multidimensional separations for proteomics.	5			
5	Principles of membrane separation, Membrane Materials, Transport phenomena of species, molecular and ionic, in porous or dense, charged or not, membranes. Membrane separation processes: Reverse Osmosis, Ultrafiltration, Microfiltration, Nanofiltration, Dialysis, Electrodialysis, Gas Permeation, Pervaporation, Liquid membranes, Membrane modules and design, cost estimation.	5			
6	Process optimization using advanced strategies, QbD, DOE, Multivariate data analysis (MVDA), Process integration and intensification for biotech products	3			
	List of Text Books/ Reference Books				
Course O	 Anurag Rathore and Ajoy Velyudhan, Scale-up and optimization in preparative chromatography, 2003 Sewell P.A. Clarke B, Chromatographic separations. John Wiley & Sons, 1991 Lindsay B., High performance Liquid Chromatography, John Wiley & Sons, Lecture Notes on short course on Enantiomeric separations, April 28-29,1995. Handbook of membrane separations: chemical, pharmaceutical, food and biotechnological applications by Anil K. Pabby, Syed Rezvi and Anna Satre, CRC press, 2009 Filtration and purification in biopharmaceutical industry, second edition by Miak Jornitz and Theodore Meltzer, Informa Healthcare, Vol. 174 				
Course O					
1	Able to understand high resolution techniques in bioseparation, purification of slarge biomolecules by chromatography, polishing and concentration steps in biomolecules by chromatography, polishing and concentration steps in biomolecules by chromatography.				
2	Able to understand column packing, designing of separation and its scale-up,				
3	Students should be able to understand of nature of membranes; membrane transmechanism; design of membrane modules and plant; membrane fouling				
4	The ability to classify adsorbents, membrane processes; determine the nature of and membranes; formulate the theory of membrane transport and apply the gen membrane theory in specific cases				
5	To find out the applications of process chromatography and membrane filtration processing for various biotech products	n for			
	processing for various biotech products				

Course Code:	Course Title: (Elective III) (Marks 50)	Cred	lits = 3	,
		L	T	P

	Semester: I	Total contact hours: 30	2	1	0
Candidate that seme		f the elective subjects offered for ets. A consolidated list of all the			

Course Code:	Course Title: (Elective IV) (Marks 50)	Credits = 3		3
		L	T	P
Semester: I	Total contact hours: 30	2	1	0

Elective-II (from the list appended)

Candidate will have to choose one of the elective subjects offered for that semester from the elective subjects. A consolidated list of all the elective subjects is given at the end.

Course Code:	Course Title: BSP 2104	Credits = 3		1
BSP 2104	Biosciences and Bioprocess Technology			
	Laboratory	L	T	P
	(Marks 50)		_	
Semester: I	Total contact hours: 30	•	-	6
	List of Prerequisite Courses			
Bioreaction Engineeri Biochemistry and Ger	ng, unit operations in bioprocessing, Microbiology, netics			
List of	Courses where this course will be prerequisite			
PhD in Bioprocess T	echnology, Biotechnology, Biochemical Engineeri	ng,		
Chemical Engineerin	g, Chemical Engineering operations			
Description of relevance of	this course in the M. Tech. (Bioprocess Techno	logy) P	rogran	ıme

Students will understand and explain the application of concepts in basic biology and life sciences as well as biochemical engineering in bioprocessing of bioproducts (both small and large biomolecules), Technical Microbiology, fermentation, biochemistry and molecular biology as well as downstream processing operations.

Module	Course Contents (Topics and subtopics)	Reqd. hours
1	Technical Microbiology pertaining to strain isolation for pure culture and its maintenance	
2	Technical Biochemistry and molecular biology pertaining to proteins and enzymes, enzyme activity and kinetics, nucleic acid isolation, protein quantification etc.	
3	Fermentation and Bioreactions: fermentation of primary and secondary metabolite on shake flask and at fermentor level with control parameters	
4	Biocatalysis: enzyme immobilization, enzymatic and whole call biocatalysis	
5	Downstream processing consisting membrane filtration MF/UF/NF/RO, column packing, column qualification, precipitation, solid-liquid extraction, thermodynamic and kinetic chromatographic separations, resolution of biomolecules,	
6	Characterization of biotech and biobased products using various analytical techniques e.g.UV/Vis, HPLC, FTIR, LC-MS/MS, DLS, Electrophoresis etc.	

Course	Course Outcomes				
1	To be able to perform experiments to ascertain the biosciences and life biosciences governing bioprocesses and biochemical engineering principles				
2	To design experimental configurations/loops for demonstrating the basic governing fundamentals				
3	To perform calculations and predict the trends in design variables as a function of operating/geometric parameters				
4	To effectively operate the analytical facilities				

	Course Code:	Course Title: BSP 2105	Cred	dits = 3	3
	BSP 2105	Research Project -II			
		(Marks 150)	L	T	P
	Semester: I	Total contact hours:	-	-	18
Course (area in consultation with the experimental/analytical/simu the outcomes of the candidat provided for BSP2102 above examiner from the Departme	acerned with detailed literature review of the assignment of the supervisor, developing an lation protocol and initiate the actual research work is expected to submit a report as per similar guide which will be evaluated by the research guide and nt/Industry based on the presentation made by the marks for report and presentation will be considered.	k. Base lelines d an ext candida	ed on ernal	
1	Systematically perform expe	riments/modeling activity to accomplish the set ob	jectives	S	
2	Critically analyse the results	and present them in coherent manner in the form	of graph	ıs, table	s etc
3	Write a technically correct re	port as per the suggested guidelines and present the	ne work		

FOLLOWING IS THE LIST OF ELECTIVE SUBJECTS

	Course Code: BST 2107	Course Title: BST 2107 Analytical Techniques in Bioprocessing	Cred	dits = 3	3
	(Elective Subject)	(Marks 50)	L	T	P
	Semester: I	Total contact hours: 30	2	1	0
		List of Prerequisite Courses			
	Basic principles and inst chromatography.	rumentation techniques like spectroscopy and liquid			
	List of Co	ourses where this course will be prerequisite			
	Chemical Engineering, Biotechnology, Bioanal		al		
		nis course in the M. Tech. (Bioprocess Technol			
products antibiotic usefulnes	including proteins, amino s, biopharmaceuticals, biolo s of various analytical tech	analytical techniques and methods of analysis for acids, vitamins, sugars, nucleic acids, organic acids gicals, biosimilars etc. They will be able to explain an industries according to the processes involved for a processe involved for a processes in a processe in a	s, polys its appli otech an	accharications d biob	ides, and
Module		Course Contents (Topics and subtopics)		Requ	l. hours
1	polysaccharides and sm products etc.	ntive analysis of proteins, nucleic acids, all molecules such as antibiotics, vitamins, nature cation of modern analytical instrumentation.	al	2	
2	Chromatography: HPLO	C (including ELSD, CAD and DLS detectors), Unatography and 2D techniques etc.	PLC,	4	
3	Mass spectrometry: Fra Derivatisation techniqu Recent developments in	gmentation patterns for molecular analysis. es. Sample introduction features for large molecular applications to proteomics and metabolomics OF, Triple Quad and Ion trap mass analyzers).	ıles.	6	
4	Immunoassay: radioimi technique (EMIT); fluo enzyme donor immunoi in solution (KIMS); enz	munoassay (RIA); enzyme-multiplied immunoas rescence polarization immunoassay (FPIA); clos assay (CEDIA); kinetic interaction of micropartic zyme-linked immunosorbent assay (ELISA).	ed	6	
5	Hybrid techniques: Gas spectroscopic detection spectrometric detection spectrometric detection plasma with mass spect proteomics, metabolom Electrophoresis: PAGE electrophoresis, 2-D tech	chromatography with Fourier transforms infrare (GC-FTIR), gas chromatography with mass (GC-MS), liquid chromatography with mass (LC-MS and LC-MS/MS), and inductively couprometric detection (ICP-MS). Applications to ics, Impurity identification and profiling., SDS-PAGE, Zone electrophoresis, Capillary chniques, laser ablation, Qualitative and quantitativalyzers. PCR, Q-PCR and RT-PCR techniques.	led	6	

6	Particle size analysis, SEM, TEM and their application in bioprocessing and bioproduct characterizations Application of IR and NMR spectroscopy, FT-IR, FT-NMR, X-ray diffraction (XRD, XRPD) and differential scanning calorimetry, Microcalorimetry in bioproducts. Synchrotron radiation and their application in bioprocessing Advanced analytical techniques like automated electrophoresis and lab on chip.	
	List of Text Books/ Reference Books	
1	Handbook of analytical separations, vol. 4, by Ian Wilson, 2003	
2	Encyclopedia of spectroscopy and spectrometry, vol. 1-3, 2000	
3	Methods of biochemical Analysis, Vol. 35, Clarence Suelter, 1991	
4	Methods of biochemical Analysis, Vol. 36, Clarence Suelter, 1992	
	Course Outcomes (students will be)	
1	Able to understand the basics of bioanalytical instrumentation and its applicability in bioprocessing be able to use these techniques carefully during their research work	
2	Quality control for biopharmaceuticals and biochemical	
3	Understand the effect of physicochemical properties of analysis of biotech and biobased products	
4	Understand the raw material, in process and finished product quality control for biotech and biobased products	
5	To find out the applications of various analytical techniques in process automation, process control and various biotech products as well as their formulations/stability etc.	

	Course Code: BST 2108 (Elective)	Course Title: BST 2108 Applied Molecular and Synthetic Biology (Marks 50) Total contact hours: 30	Credits = 3		
(Elec			L	T	P
Sem	ester: I		2	1	0
		List of Prerequisite Courses			
Basi	c biochemistry, n	nicrobiology and genetics			
	List of C	ourses where this course will be prerequisite			
Chen		chnology, Biotechnology, Biochemical Engineering, Chemical Engineering operations, Pharmaceutical alyticals			
		his course in the M. Tech. (Bioprocess Technolog	gy) Pr	ogran	ıme

Students will understand (a) basic genetic toolbox/components to manipulate a cellular system to produce a range of natural and engineered products; (b) Design of efficient and robust toolbox for genetic manipulation of cellular platforms like *E. coli*, *Saccharomyces cerevisiae* and (c) Metabolic engineering of the cellular system to enhance the product titer. They will be able to explain its applications in production

and designing of Biopharmaceutical and industrial biotechnology products such as recombinant therapeutic proteins and peptides, sugar based molecules like xylitiol, designer biocatalysts etc. .

Module	Course Contents (Topics and subtopics)	Reqd. hours	
1	Concept and History of Synthetic Biology	1	
2	Basic expression and regulation in a model organism: <i>E. coli</i> 2.1 Gene expression and regulation 2.2 Metabolic pathways and its regulation	4	
3	Natural and advanced Genetic Tool Box for Manipulation of Pathways and Gene Expression 3. 1 Extra chromosomal tools: Plasmids, Cosmids 3.2 Genomic tools: Homologous recombination, CRISPR-Cas systems 3.3 Synthetic expression elements: Promoter, ribosome binding sites 3.4 Advance tools for assembly of genetic elements: Gibson assembly 3.5 Genome engineering and synthetic cells	8	
4	Synthetic Biology approach for production of Biopharmaceuticals 4.1 Anti malarial Drug: artemisinin production in <i>Saccharomyces cerevisiae</i> 4.2 Glycoengineered microbial strain for glycosylated proteins 4.3 Production of recombinant proteins: Single protein production (SPP) system in <i>Escherichia coli</i> 4.4 Production of sugar based biotechnological molecules: Xylitol, GOS (galacto-oligosaccharides)	8	
5	Synthetic Biology approach for production of Biochemical and Biocatalyst 5.1 Cell free SyPaB (Synthetic Pathway for Biotransformation) 5.2 Cellular platforms for Microbial Engineering	5	
6	Critically analyse scientific literature in the area 6.1 The International Genetically Engineered Machine (iGEM) foundation and 6.2 Case study of iGEM competition	4	
_	List of Text Books/Reference Books		
1	Synthetic Biology: Tools and Applications Edited by Huimin Zhao; ISBN: 978-0-12	2-394430-6	
2	Bioengineering: A conceptual Approach by Mirjana Pavlovic; ISBN 978-3-319	-10798-1	
	Course Outcomes (students will be)		
1	Able to understand the basics of gene expression and pathway regulation		
2	Genetic tool boxes to manipulate the cellular platform for production of Biopharmaceuticals		
3	Production and application of designer biotechnological products		
4	Understand the concept of synthetic cell		
5	Application of synthetic biology for discovery of new set of biotechnological P	roducts	

	Course Code: BST 2109	Course Title: BST 2109 Cro Transport Phenomenon in Bioprocessing	Credits = 3		
	(Elective)	(Marks 50)	T	P	
	Semester: I	Total contact hours: 30 2	1	0	
	I	List of Prerequisite Courses			
	Basic mathematics and	d Algebra			
	List of (Courses where this course will be prerequisite			
	PhD in Bioprocess T	echnology, Biotechnology, Biochemical Engineering,			
	Chemical Engineerin	g, Chemical Engineering operations, Pharmaceutical			
	Biotechnology, Bioan		_		
Desci	ription of relevance of	this course in the M. Tech. (Bioprocess Technology)	Prograi	nme	
		aid flow and its types in various biotech processes i.e. for will be able to explain its applications in bioprocess development.			
Module		Course Contents (Topics and subtopics)	Requ	d. hours	
1		mensional diffusive transport: momentum, heat and mass logies; characteristics of transport processes;	ss 4		
2		ole shear flow and developing flows, entrance effect,	6		
	-	s. Multiphase systems and transport coefficients; ; Transport in turbulent condition	4		
3	•	•			
4	Non-steady state transport; Transport phenomena in bioprocesses and biosystem: interphase, diffusion in biofilm-floc, determination of transport coefficients, agitation power, and evaluation of oxygen transport rate as a function of operating variables.				
5	Introduction to micro	ofluidics in bioprocessing unit operations	4		
6		oort phenomenon in various bioprocesses, case studies in armaceuticals, bioanalysis	6		
	L	ist of Text Books/ Reference Books			
1	Biron R. Bird, Warre	n E. Stewart, and Edwin Lightfoot, "Transport Phenome	na"		
2	Bennet C.O. and Mey	yer J.E., "Momentum and mass Transfer"			
3	Sission and Pitts "Int	roduction to Transprot Phenomena"			
4	Christie J. Geankoplis, "Transport Processes and Unit Operations", Prentice hall of India, 1997				
5	J.C.Slattery, " Mom company	entum, Energy and Mass Transfer in continuum, Kr	iger Pu	ıblishin	
	(Course Outcomes (students will be)			
1	Able to understand the development of biopro	ne basics of transport processes and its applicability in decrocesses	sign and	i	
2	Able to understand d	ifferent behaviour of fluid based on biotech system and p	rocesse	S	
3	Understand the use o	f transport phenomenon in scale up of biotech industries			
4	To find out the applic	cations in various biotech products i.e. small and large bio	omoleci	ıles etc.	

	Course Code: BST 2109 (Elective)	Course Title: BST 2109 Fermentation and Cell Culture Engineering (Marks 50)	Credits = 3		
			L	T	P
	Semester: II	Total contact hours: 30	2	1	0
	List of Prerequisite Courses				
	Biological sciences, life sciengineering	ences, microbiology, biochemistry, bioreaction			
	List of Cour	ses where this course will be prerequisite			
		ology, Biotechnology, Biochemical Engineering nemical Engineering operations, Pharmaceutical icals			
Descr	ription of relevance of this	course in the M. Tech. (Bioprocess Technolo	gy) P	rogran	nme
	ic products	al and animal cell culture for production of ther	apeutio		
Module		Course Contents (Topics and subtopics)		Reqd	l. hours
1	Nature of fermentation processes, Nutritional requirements in fermentation process, Strain Construction and Strain Improvement		6		
2	Modern Experimental Techniques: Batch, Fed-Batch, Continuous and extractive Fermentation, High cell-density and High-Performance Bioreactors, Quantitative Physiological Studies		6		
3	Aerobic and anaerobic fermentation, surface, submerged and solid state fermentation technology, Statistical methods for fermentation optimization, Instrumentation and Control Systems, Improving the production of recombinant DNA proteins through fermentation development,		6		
4	Automation, optimization and Control of fermentation processes, Fermentation design and Cost, Design considerations for aseptic fermentation, Case studies with respect to antibiotic, enzymes and therapeutics.				
5	1	and technology: Plant and mammalian cell cu	lture	6	
	List of	f Text Books/ Reference Books			
1	Wang D. I. C., Cooney C. I and Enzyme Technology, Jo	L., Demain A. L., Dunnil P., Humphrey A. E., Lilly hn Wiles and Sons., 1980.	у М. Г	D., Fern	nentation
2		A., Principles of Fermentation Technology, Pergan	non Pro	ess, 198	34.
3	Zubay G., Biochemistry, Ma	cmillan Publishers, 1989.			

Course Outcomes

- knowledge of microbial fermentation, growth kinetic and product formation knowledge of

1	Able to understand microbial fermentation, growth kinetic and product formation
2	Able to understand strain improvement, cell culture and its application for production of various bioproducts

3	Understand the use of fermentation and bioreaction for production of therapeutic and non-therapeutic products
4	To find out the applications in various biotech products i.e. small and large biomolecules etc.