

**Syllabus for Bachelor of Technology**  
**(B.Tech. in Surface Coating Technology)**  
**(Under the New Education Policy-NEP 2020)**

**in**



**(2023-2024)**

**INSTITUTE OF CHEMICAL  
TECHNOLOGY**

(University Under Section-3 of UGC Act,  
1956)

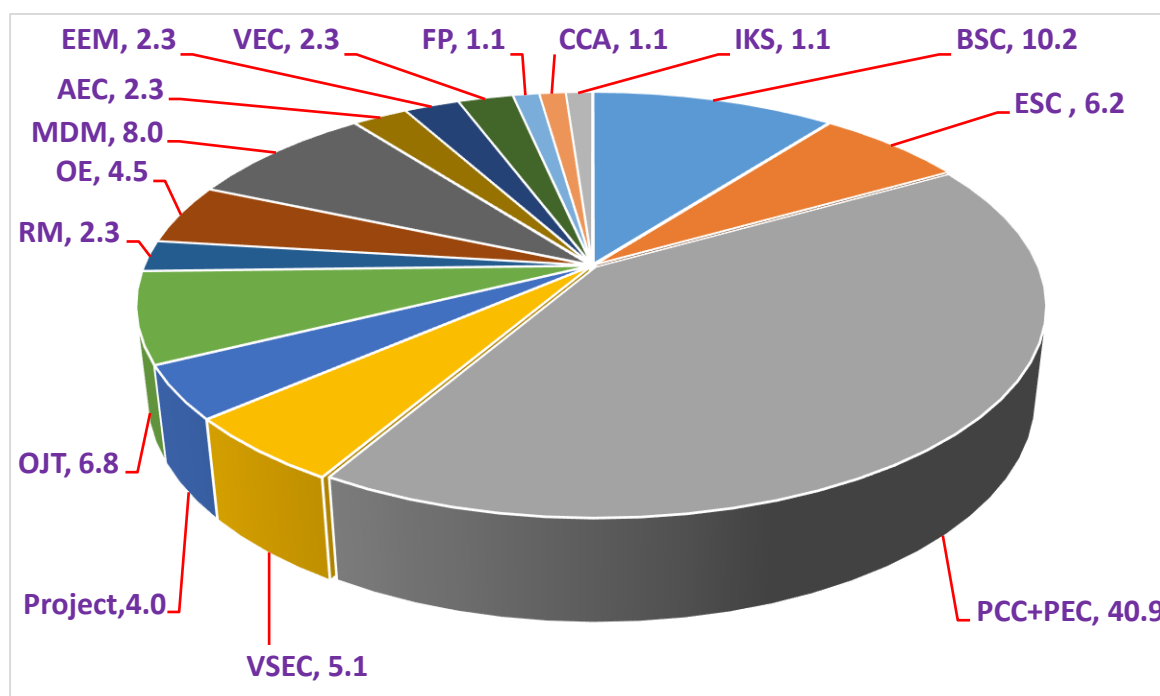
**Elite Status and Center for Excellence**  
**Government of Maharashtra**

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## Department of Polymer & Surface Engineering

### Preamble:

The undergraduate programmes at the Institute of Chemical Technology are reputed worldwide. Alumni from these programmes have found a place of pride in the Indian chemical industry including some top names and many as entrepreneurs, in Universities/ Institutes and Research Organisations throughout India and the world. The B.Tech. programmes in the then Department of Chemical Technology, University of Mumbai started in 1934 as post B.Sc., second graduation as B.Sc.(Tech.). Keeping national, societal needs in focus, post-independence, the programme grew into multiple branches keeping connection with chemical engineering content. Once the Institute became a University in 2009, these became independent B. Tech. Programmes retaining their dual core nature. The Institute of Chemical Technology is committed to keeping its syllabi updated and globally relevant for the industry. We have revamped the syllabi of all the B. Tech. programmes now in 2023 as per NEP 2020. The 176 credit programme each has following Credit Distribution



**This does not include Honors courses of 18 credits.**

All the courses are credit based and the evaluation are grade based. The credit system is a systematic way of describing an educational programme by attaching credits to its components. The definition of credits is based on student workload, learning outcomes and contact hours. This system is described in detail in Regulation No.9 of the Institute. Each theory course consists of Lectures and tutorials. During tutorial session, it is expected that the problem solving / case studies / relevant real life applications / student presentations / home assignments/individual or group projects are discussed in the presence of the teacher. Teacher can have the freedom to interchange lectures / tutorials depending upon the topic. Institute gives emphasis on continuous evaluation with considerable freedom to the teacher in deciding the mode of evaluation.

## **B. Tech. (Surface Coating Technology)**

### **PROGRAMME EDUCATIONAL OBJECTIVES for B. Tech. (Surface Coating Technology)**

- PEO-1: Graduate with in-depth knowledge in the field of polymer engineering science and technology applicable for successful career in Polymer and Surface coating Technology.
- PEO-2: Graduates with integrity, strong ethical values who are members and contribute to professional society.
- PEO-3: Graduates who engage in lifelong learning or continuous education opportunities.
- PEO-4: To prepare Graduates who contribute towards research and professional Development and who are entrepreneurial engineers.

**Programme Outcomes (POs) for B. Tech. (Surface Coating Technology)**

PO1	Surface coating technology knowledge: Apply the knowledge of chemistry, science, chemical engineering and paint technology fundamentals, and surface coating technology specialization to the solution of complex problems in coating technology.
PO2	Problem analysis: Identify, formulate, review research literature, and analyze complex surface coating technology problems reaching substantiated conclusions and designing of innovative coatings to fulfil the need of country using first principles of chemistry, polymer sciences, and surface engineering sciences.
PO3	Design/development of solutions: Design solutions for complex coating technology problems and design system components or processes that meet the specified needs with appropriate consideration for the expected service life of MOCs, aesthetic appearance, safety and efficacy of the product and environmental considerations.
PO4	Conduct investigations of complex problems: Use research-based knowledge and research methods including design of experiments, analysis and interpretation of data and using that information to provide valid conclusions.
PO5	Modern tool usage: Create, select, and apply appropriate techniques, resources, and modern engineering and IT tools including predictions and conclusions for complex surface coating technology activities with an understanding of the limitations.
PO6	The engineer and society: Apply reasoning informed by the contextual knowledge to assess societal, health, safety, legal and environmental issues and the consequent responsibilities relevant to the professional practice of surface coating technology.
PO7	Environment and sustainability: Understand the impact of the professional surface coating technology solutions in societal and environmental contexts, and demonstrate the knowledge of, and need for substantial development.
PO8	Ethics: Apply ethical principles and commit to professional ethics and responsibilities and norms of the practice of surface coating technology.
PO9	Individual and teamwork: Function effectively as an individual, and as a member or leader in diverse teams, and in multidisciplinary settings.
PO10	Communication: Communicate effectively surface coating technology activities with the coating community and with society at large, such as, being able to comprehend and write effective reports and design documentation, make effective presentations, and give and receive clear instructions.
PO11	Project management and finance: Demonstrate knowledge and understanding of the coating technology and management principles and apply these to one's own work, as a member and leader in a team, to manage projects and in multidisciplinary environments.
PO12	Life-long learning: Recognize the need for, and have the preparation and ability to engage in independent and life-long learning in the broad context of technological change.
<b>(B) Program Specific Outcomes (PSOs)</b>	
PO13	Higher studies: Able to have knowledge for higher studies related to Surface Coating Technology disciplines.
PO14	Pertinent with paint industry: Able to develop skills about paint manufacturing, application and testing with following paint industry safety and regulation norms with inculcating the thought of sustainable development

## Graduate Attributes

1. Problem analysis and solving skills
2. Familiar with usage of modern tools, techniques
3. Communication Skills
4. Capacity to analyze new concepts
5. Capacity to analyze and interpret experimental data Capacity to analyze business trends
6. Capacity to design, optimize and operate equipment and plants safely, economically and effectively
7. Design and Development of solutions to industrial and societal needs
8. Skills related to Project Management and Economics
9. Skills to analyze scientific literature including patents
10. Ethics

## Syllabus Structure for B. Tech Course

### SEMESTER- I

Course Code	Subjects	Course Type	Credits	Hrs/Week			Marks for various Exams			
				L	T	P	C.A.	M.S.	E.S.	Total
CHT1405	Physical Chemistry	BSC	3	2	1	0				
CHT1406	Analytical Chemistry	BSC	3	2	1	0				
MAT1205	Engineering Mathematics	ESC	3	2	1	0				
PYT1205	Applied Physics	BSC	2	1	1	0				
GET1305	Engineering Graphics and Computer Aided Drawing	VSEC	3	1	0	4				
PST1101	<b>SPL-1:</b> Polymer science and technology I(Common)	ESC	2	1	1	0				
PYP1101	Physics Laboratory	BSC	2	0	0	4				
HUT1110B	Communication Skills(English)	AEC	2	0	0	4				
	OPEN Activity - Sports/ Fine arts/Yoga/ Music/NSS**	CCA	2	0	0	4				
	<b>TOTAL:</b>		<b>22</b>	<b>9</b>	<b>5</b>	<b>16</b>				

### SEMESTER- II

Subject Code	Subjects	Course Type	Credits	Hrs/week			Marks for various Exams			
				L	T	P	C.A.	M.S.	E.S.	Total
CHT1407	Organic Chemistry	BSC	3	2	1	0				
CHT1408	Industrial Chemistry	BSC	3	2	1	0				
SCT1201	<b>SPL-2:</b> Introduction to coating technology	PCC	2	1	1	0				
GET1306	Basic Mechanical Engineering	ESC	2	1	1	0				
GET1125	Electrical Engineering and Electronics	ESC	2	1	1	0				
CEP1720	Process Calculations	ESC	2	0	0	4				
CHP1343	Physical and Analytical Chemistry Laboratory	BSC	2	0	0	4				
CHP1132	Organic Chemistry Laboratory	VSEC	2	0	0	4				
	OPEN Activity- Sports/ Fine Arts/Yoga/ Music/NSS**	CCA	2	0	0	4				
	MOOC- Indian Knowledge System (NPTEL - Introduction to	IKS	2	0	0	4				

	Ancient Indian Technology)									
	<b>TOTAL:</b>		<b>22</b>	<b>7</b>	<b>5</b>	<b>20</b>				
<p>Note: Universal Human Values (UHV) an audit course to be taken in inter-semester break after Semester-II to be taken as MOOC course.</p> <p>** Students will undertake these co-curricular activities such as sports / Fine Arts / Yoga / Music / Literature etc administered through various clubs under Technological Association approved by Dean, Students Affairs.</p>										
<b>SEMESTER- III</b>										
Subject Code	Subjects	Course Type	Credits	Hrs/week			Marks for various Exams			
				L	T	P	C.A.	M.S.	E.S.	Total
PST1303	<b>SPL-3:</b> Polymer chemistry and technology (Common)	PCC	4	3	1	0				
PST1304	<b>SPL-4:</b> Polymer science and Technology II (Common)	PCC	2	1	1	0				
OE	From Basic Sciences (Chemistry/ Physics/Biology / Maths / Humanities)	OE	4	3	1	0				
	Communication Skills – (Marathi / Hindi or Any other language will be chosen using MOOCS)	AEC	2	1	1	0				
HUT1205	Basic Economics and Finance	EEM	2	1	1	0				
	Digital Computation in Emerging Areas (NPTEL course: <b>Introduction To Industry 4.0 And Industrial Internet Of Things</b> )	VEC	2	0	0	4				
	<b>MDM-I:</b> From Sciences and/or any other Engineering / Humanities Discipline	MDM	2	1	1	0				
PSP1301	<b>Pr 1:</b> Lab-1: Raw Material Analysis for Resins and Polymers (Common)	PCC	2	0	0	4				
PSP1302	<b>Pr 2:</b> Lab 2: Synthesis and Characterization of Resins and Polymers Lab I (Common)	PCC	2	0	0	4				
	<b>TOTAL:</b>		<b>22</b>	<b>11</b>	<b>7</b>	<b>8</b>				
<b>SEMESTER- IV</b>										
Subject Code	Subjects	Course Type	Credits	Hrs/week			Marks for various Exams			
				L	T	P	C. A.	M.S.	E. S.	Total
CET1105	Transport Phenomena	PCC	4	3	1	0				
PST1401	<b>SPL-5:</b> Technology of Thermoplastic Polymers (common)	PST	3	2	1	0				
PST1505	<b>SPL-6:</b> Technology of Thermoset polymers (common)	PCC	3	2	1	0				
OE	From Basic Sciences (Chemistry/ Physics/ Biology / Maths) or Humanities Discipline	OE	2	1	1	0				
CET1805	Chemical Process Economics	EEM	2	1	1	0				

HUT1206	Environmental Sciences and Technology	VEC	2	1	1	0				
	<b>MDM II:</b> From Sciences and/or any other Engineering /Humanities	MDM	2	1	1	0				
	Community Projects#	CEP/FP	2	0	0	4				
PSP1401	<b>Pr 3:</b> Lab-3 Synthesis and Characterization of Resins and Polymers Lab II (Common)	VSEC	2	0	0	4				
	<b>TOTAL:</b>		<b>22</b>	<b>11</b>	<b>7</b>	<b>8</b>				

# Students will undertake community projects as individual or group related to study of societal technological activities through various organization such as Lions club, Teach India, Marathi Vidnyan Parishad, CSR projects outsourced by various industries, ISR activities administered through Technological Association approved by the Dean, Student Affairs.

### SEMESTER- V

Subject Code	Subjects	Course Type	Credits	Hrs /week			Marks for various Exams			
				L	T	P	C. A.	M.S.	E. S.	Total
CET1806	Chemical Reaction Engineering	PCC	2	1	1	0				
CET1807	Chemical Engineering Operations	PCC	2	1	1	0				
SCT1501	<b>SPL-7:</b> Paint technology I	PCC	4	3	1	0				
	Offered by the department/MOOCs (One of the elective can be PST1609) <b>SPL-8:</b> Structure property Relationship (Common)	PEC	4	3	1	0				
OE	From Basic Sciences (Chemistry/ Physics/ Biology / Maths) or Humanities Discipline	OE	2	1	1	0				
PST1501	Honors Course-I (High polymer chemistry)	PCC	4	3	1	0				
	<b>MDM III:</b> From Sciences and/or any other Engineering / Humanities Discipline	MDM	4	2	0	4				
SCP1606	<b>Pr 4:</b> Lab 4: Processing of paints lab I	PCC	2	0	0	4				
PSP1504	<b>Pr 5:</b> Lab 5 : Analysis and Characterization of Resins and Polymers Lab (Common)	PCC	2	0	0	4				
	<b>TOTAL:</b>		<b>26</b>	<b>14</b>	<b>6</b>	<b>12</b>				

### SEMESTER- VI

Subject Code	Subjects	Course Type	Credits	Hrs/week			Marks for various Exams			
				L	T	P	C.A.	M.S.	E. S.	Total
SCT1502	<b>SPL-9:</b> Additives and processing of paint	PCC	3	2	1	0				
SCT1601	<b>SPL-10 :</b> Paint technology II	PCC	3	2	1	0				

	Offered by the department/MOOCs (One of the elective can be PST1712) SPL-11: Environmental health and Safety of Polymers and Coatings (Common)	PEC	4	3	1	0				
SCT1815	<b>SPL-12:</b> Advanced paint technology	PCC	4	3	1	0				
PST1610	Honors Course-II (Biopolymers)	PCC	4	3	1	0				
	<b>MDM IV:</b> From Sciences and/or any other Engineering / Humanities Discipline	MDM	2	1	1	0				
CEP1714	Chemical Engineering Laboratory	VSEC	2	0	0	4				
SCP1608	<b>Pr 6:</b> Lab-6 Synthesis, processing and characterization of colorants.	PCC	2	0	0	4				
SCP1609	<b>Pr 7:</b> Lab -7 : Processing of paints lab II	PEC	2	0	0	4				
	<b>TOTAL:</b>		<b>26</b>	<b>14</b>	<b>6</b>	<b>12</b>				
<b>SEMESTER- VII</b>										
Subject Code	Subjects	Course Type	Credits	Hrs/week			Marks for various Exams			
				L	T	P	C. A.	M.S.	E.S.	Total
PST1711	<b>SPL-13:</b> Evaluation and Testing of polymers and coatings (Common)	PCC	3	2	1	0				
SCT1701	<b>SPL-14-</b> Curing mechanism of coating	PCC	2	1	1	0				
	Offered by the department/MOOCs (One of the elective can be Printing inks - SCT1813)	PEC	3	2	1	0				
	Offered by the department/MOOCs (One of the elective can be High performance coating – SCT1703/ Intellectual Property Rights - PHT1440)	PEC	2	2	0	0				
PST1714	Honors-III (Nanomaterials and nanocomposites)	PCC	4	3	1	0				
	<b>MDM V:</b> From Sciences and/or any other Engineering / Humanities Discipline	MDM	2	1	1	0				
PHP1442	Literature Review (Research Methodology - I)	RM-1	2	1	0	2				
PHT1443	Design and Analysis of Experiments (Research Methodology - II)	RM-2	2	1	0	2				
	Project -I (Literature search + Expt)	Project	4	0	0	8				
SCP1701	<b>Pr 8:</b> Lab-8: Processing and characterization of paints	PCC	2	0	0	4				
	<b>TOTAL:</b>		<b>26</b>	<b>13</b>	<b>5</b>	<b>16</b>				



**SEMESTER- VIII**

**Semester-VIII (10 weeks)**

Subject Code	Subjects	Course Type	Credits	Hrs /week			Marks for various Exams			
				L	T	P	C.A.	M.S.	E. S.	Total
PST1801	<b>SPL-15:</b> Adhesion and adhesives	PCC	3	5	1	0				
SCT1816	Honors Course-IV ( <b>Corrosion Science and Corrosion prevention</b> )	PCC	3	5	1	0				
PST1713	Honors Course-V ( <b>Sustainability of polymers</b> )	PCC	3	5	1	0				
	<b>MDM VI:</b> From Sciences and/or any other Engineering / Humanities Discipline	MDM	2	2	1	0				
	Project-II (Experiments)	PCC	3	0	0	12				
SCP1801	<b>Pr 9:</b> Lab-9: Analysis and testing of paints	PEC	2	0	0	6				

**Semester-VIII (12-16 weeks)**

PHP1451	Internship with Industry	OJT	12	0	0	0				
	<b>Total</b>		<b>28</b>	<b>17</b>	<b>4</b>	<b>18</b>				

**Internship**

- In the Eighth semester, every student will have to undergo an internship and/or On Job Training. The Internship would be of 12 credits.
- The internship would be assigned to the student by the Departmental Internship Coordinator, with the approval of Head, Chemical Engineering Department.
- The total duration of the internship would be for a period equivalent to 12 Calendar weeks. The internship may be completed in one or more organizations as described below.
- The internship could be of the following forms:
- Industrial internship in a company (within India or Abroad) involved in R&D / design / manufacturing (QA/QC/Plant Engineering/Stores and Purchase) / marketing / finance / consultancy / Technical services / Engineering / Projects, etc.
- At the end of the internship, each student will submit a written report based on the work carried out during the Internship. The report will be countersigned by the Supervisor from Industry / Institute as the case may be.
- Performance of the student will be assessed based on the written report and a presentation to a committee consisting of two faculty members from the Chemical Engineering Department.
- Students will be assigned a grade based on the written report and a presentation; evaluated by a committee of faculty members.
- Feedback will be taken from Industry mentors and this will used while assigning the grades.

BSC: Basic Science Course,

ESC: Engineering Science Course

PCC: Program Core Course, PEC: Program Elective Course

MDM: Multi-disciplinary Minor: Different discipline of engineering or different faculty altogether

OE: Open Elective: To be chosen Compulsorily from faculty other than major discipline

VSEC: Vocational and Skill Enhancement Course: Hands on training corresponding to major/minor

AEC: Ability Enhancement Course: English 2 credit, Modern Indian Language 2 credit

IKS: Indian Knowledge System: Indian Architecture/Maths/Medicine

VEC: Value Education Course: e.g. Understanding India, Environmental Science / Education / Digital and Tech solutions

RM: Research Methodology

CCA: Co-curricular activities: Health and wellness / Yoga / Sports / Cultural activities / NSS/NCC/Applied visual performing arts

### **EXIT Policy**

Based on the National Education Policy guidelines, the students have an option of exiting at each level of their four year program. Student will get certificate after 1<sup>st</sup> year, diploma after second year and B.Sc (Tech/Engg) after third year.

<b>Sr. No.</b>	<b>Exit Year</b>	<b>Activity</b>	<b>Credits</b>	<b>Duration (No of Weeks)</b>
1	1 <sup>st</sup> Year (After Semester II)	8 credit course workshop/chemistry lab (after semester 2)	8	8 weeks
2	2 <sup>nd</sup> Year (After Semester IV)	Certificate Course in Practice of Chemical Technology (CCPCT)	8	8 weeks
3	3 <sup>rd</sup> Year (After Semester VI)	In-plant training	8	8 weeks

# Semester-I

BSC	Course Code: CHT1405	Course Title: Physical Chemistry	Credits = 3		
	Semester: I		Total Contact Hours: 45	L	T
<b>List of Prerequisite Courses</b>					
Standard XII Chemistry					
<b>List of Courses where this course will be Prerequisite</b>					
Physical and Analytical Chemistry laboratory , other multidisciplinary courses on Chemistry / Chemical Engineering.					
<b>Description of relevance of this course in the B. Tech. Programme</b>					
The course will enable the students to understand and apply the principles of thermodynamics to real-world systems. The students would be able to apply the insights to understand the stability of solutions, spontaneity of physical/chemical processes, effect of thermodynamics parameters on phase and chemical equilibria, etc.					
Sr. No.	Course Contents (Topics and Subtopics)				Required Hours
1	<b>Laws of thermodynamics –</b> a) Enthalpy and heat capacities, application of first law to gases, thermochemistry- Hess law b) Statements and applications of second law of thermodynamics, Clausius inequality, entropy as a state function, entropy changes for reversible and irreversible processes, entropy and probability c) Third law of thermodynamics, absolute entropies, verification of third law				6
2	<b>Spontaneous process and equilibrium</b> –Helmholtz and Gibbs free energy, spontaneity and free energy, Maxwell’s relations, effect of T and P on free energy,				3
3	<b>Multicomponent system</b> – free energy and entropy of mixing, partial molar quantities and chemical potential, Gibbs Duhem equation				6
4	<b>Equilibrium in solutions</b> – ideal and non ideal solutions, Henry’s law and Raoult’s law, colligative properties, activity and activity coefficients, thermodynamic properties of electrolytes in solution				7
5	<b>Solubility equilibria</b> – solubility constant, common ion effect, effect of added salts on solubility pH, weak and strong acids and bases, buffer solutions, ionic solutions <b>Chemical Equilibria</b> – le Chaterlier’s principle, Effect of temperature, pressure and composition on equilibrium				5
6	<b>Introduction</b> – concept of reaction rates and order, experimental methods in kinetic studies, differential and integral methods to formulate rate equations of zero, first and second order reactions Experimental methods of kinetic studies				3
7	<b>Kinetics and reaction mechanism</b> – rate determining step, steady state approximation <b>Complex reactions</b> - parallel, consecutive and reversible reactions Mechanism of thermal, photochemical chain reactions, polymerization reactions <b>Fast reactions</b> – experimental techniques				6
8	<b>Homogenous catalysis</b> – homogeneous acid / base catalysis (specific and general acid catalysis), enzyme catalysis (Michelis Menten kinetics)				6
9	<b>Reactions at interface</b> – Adsorption isotherms, kinetics of surface reactions- Hishelwood and Rideal models of surface reactions				3
<b>Total</b>					<b>45</b>
<b>List of Text Books/Reference Books</b>					
1	Atkins, Peter W.; Paula, Julio de; Keeler, James. Atkin’s Physical Chemistry; 11 <sup>th</sup> ed.; Oxford University Press (2018)				
2	Elements of Physical Chemistry (7th edition) by P. W. Atkins and J. de Paula, Oxford University Press, 2016.				

3	Chemical Kinetics (3rd edition) by Keith J. Laidler, New York : Harper & Row, 1987.
<b>Course Outcomes (Students will be able to.....)</b>	
CO1	<i>Elements of Physical Chemistry</i> (7 <sup>th</sup> edition) by P. W. Atkins and J. de Paula, Oxford University Press, 2016.
CO2	<i>Physical Chemistry</i> (6 <sup>th</sup> edition) by Ira Levine, McGraw-Hill Education, 2009
CO3	Elucidate the effect of thermodynamic quantities on chemical equilibria and relate it to properties of chemical systems
CO4	Comprehend fundamental knowledge in chemical kinetics with basics of order, molecularity and temperature effect
CO5	Examine kinetics for complex, fast as well as surface reactions and comprehend different theories in kinetics

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
<b>CO2</b>	K3	3	3	2	2	2	3	1	3	0	3	2	2	2	3
<b>CO3</b>	K3	3	3	1	2	2	0	3	3	2	3	3	2	3	3
<b>CO4</b>	K2	2	2	0	2	0	3	3	3	3	3	3	1	2	2
<b>Course</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

BSC	<b>Course Code:</b> CHT1406	<b>Course Title:</b> Analytical Chemistry	<b>Credits = 3</b>		
	<b>Semester: I</b>	<b>Total Contact Hours: 45</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>2</b>	<b>1</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
Standard XII Chemistry					
<b>List of Courses where this course will be prerequisite</b>					
Physical and Analytical Chemistry Laboratory , other Chemistry Courses					
<b>Description of relevance of this course in the B. Tech. Program</b>					
The course introduces the students to key concepts of chemical analysis – sampling, selection of analytical method and data analysis. It presents basic techniques like spectroscopy and chromatography. The students should be able to select an appropriate analytical technique and apply it in accordance with its strengths and limitations.					
<b>Sr. No.</b>	<b>Course Contents (Topics and Subtopics)</b>				<b>Required Hours</b>
1	Introduction to chemical analysis, terminology (technique / method / procedure / protocol), broad classification of analytical techniques, good laboratory practices				5
2	Criteria for selecting analytical methods – accuracy, precision, sensitivity, selectivity, and detection limit Calibration and validation				8
3	<b>Data analysis:</b> errors – systematic and random errors, statistical treatment of experimental results (F, Q and t tests, rejection of data, and confidence intervals), least square method, correlation coefficients				6
4	<b>Spectroscopic methods:</b> General principle, instrumentation and applications of - UV-visible spectroscopy - Infrared spectroscopy - fluorescence spectroscopy				8
5	<b>Electrochemical methods:</b> General principle, instrumentation and applications of - Conductometry - Potentiometry				8
6	<b>Chromatographic methods:</b> General principle, instrumentation and applications of - Gas chromatography (GC) - HPLC				10
	<b>Total</b>				<b>45</b>
<b>List of Textbooks/Reference Books</b>					
1	David Harvey. Modern Analytical Chemistry; McGraw-Hill (1999)				
2	R. A. Day and A. L. Underwood. Quantitative Analysis, Prentice Hall of India (2001)				
3	H. H. Willard, L. L. Merritt, J. A. Dean and F. A. Settle. Instrumental Methods of Analysis, 7 <sup>th</sup> ed.; Wadsworth Publishing, USA (2004)				
4	D. A. Skoog, D. M. West, F. James Holler and S. R. Crouch. Fundamentals of Analytical Chemistry; 9 <sup>th</sup> ed.; Cengage Learning (2013)				
5	D. A. Skoog, F. James Holler and S. R. Crouch. Principles of Instrumental Analysis; 6 <sup>th</sup> ed.; Cengage Learning (2016)				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	apply the knowledge of sampling, data analysis and select proper analytical method. (K3)				
CO2	explain the principles of UV Visible and Fluorescence spectroscopic methods. (K2)				
CO3	explain the principles of electrochemical methods. (K2)				
CO4	Understand the principles of chromatographic separations. (K2)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	0	3	3	0	2	3	3
<b>CO2</b>	K2	3	1	0	1	1	0	3	3	2	3	3	0	2	2
<b>CO3</b>	K2	3	2	1	2	0	3	3	3	3	2	3	1	3	2
<b>CO4</b>	K2	3	2	1	1	1	3	2	3	3	3	3	1	1	2
<b>Course</b>	K3	3	2	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge-level from cognitive domain; A, Affective domain; P, Psychomotor domain

ESC	Course Code: MAT 1301	Course Title: Engineering Mathematics	Credits = 3		
	Semester: I		L	T	P
		Total contact hours: 45	2	1	0
<b>List of Prerequisite Courses</b>					
HSC Standard Mathematics					
<b>List of Courses where this course will be prerequisite</b>					
This is a basic Mathematics course. This knowledge will be required in almost all subjects later.					
<b>Description of relevance of this course in the B. Tech. Program</b>					
This is a basic Mathematics course which will give the students the required foundations of mathematics to understand engineering concepts in the later part of the technology programs in ICT Mumbai. This course will also introduce probability distributions and basic statistics will be helpful to understand various data science studies in different engineering disciplines.					
<b>Course Contents (Topics and subtopics)</b>					<b>Required Hours</b>
1	<b>Linear Algebra:</b> Vectors in $\mathbb{R}^n$ , notion of linear independence and dependence. $\mathbb{R}^n$ as a vector space, vector subspaces of $\mathbb{R}^n$ , basis of a vector subspace, row space, null space, and column space, rank of a matrix. Determinants and rank of matrices. Linear transformations in $\mathbb{R}^n$ , Matrix of a linear transformation, change of basis and similarity, rank-nullity theorem, and its applications. Inner product spaces, orthonormal bases, Gram-Schmidt orthogonalization process, Eigenvalues and eigenvectors, characteristic polynomials, eigenvalues of special Orthogonal projection and its application to least square methods, Diagonalization of matrices and its applications to stochastic matrices				15
2	<b>Differential Calculus:</b> Higher order differentiation and Leibnitz Rule for the derivative, Taylor's and Maclaurin's theorems, Maxima/Minima, convexity of functions and applications. Functions of two or more variables, Limit and continuity, Partial differentiation, Total derivatives, Taylor's theorem for multivariable functions and its application to error calculations, Maxima/Minima, Method of Lagrange Multipliers, Introduction to double and triple integrals.				15
3	<b>Probability &amp; Statistics:</b> Random variables and cumulative distribution function; probability mass function and probability density function; Some common univariate distributions: Binomial, Poisson, Uniform, exponential, Normal; Expectation and Moments; Moment generating function, Multiple random variables, and Joint distribution; marginal distributions, Covariance and Correlation. Concept of parameter estimation: maximum likelihood estimation; method of least squares and simple linear regression; nonlinear regression				15
<b>Total</b>					<b>45</b>
<b>List of Textbooks/ Reference Books</b>					
1	G. Strang, Linear Algebra and its Applications (4th Edition), Thomson (2006).				
2	Howard Anton, Elementary Linear Algebra, John Wiley & Sons (2016)				
3	Stewart, James, Single Variable Calculus, 6th Edition, Cengage learning (2016)				
4	Hughes-Hallett et al., Calculus - Single and Multivariable (3rd Edition), John-Wiley and Sons (2003).				
5	E. Kreyszig, Advanced Engineering Mathematics (8th Edition), John Wiley (1999). (Officially prescribed)				
6	S. R. K. Iyengar, R. K. Jain, Advanced Engineering Mathematics Narosa, (2020)				
7	A First Course in Probability, Sheldon Ross, Pearson Prentice Hall, 9 <sup>th</sup> Edition (2018)				
8	W.W. Hines, D. C. Montgomery, D.M. Goldsman, John-Wiely, Probability and Statistics in Engineering, John Wiley & Sons (2008)				
9	Alexander M. Mood, Duane C. Boes, and Franklin A. Graybill, Introduction to the Theory of Statistics, Mc GrawHill, (1973)				
<b>Course Outcomes (students will be able to....)</b>					
CO1	Understand the notion of differentiability and be able to find maxima and minima of functions of one and several variables(K2, K3)				
CO2	Understand the computational and geometrical concepts related to linear transformations, eigenvalues and eigenvectors and apply them to solve computational problems(K1, K2, K3)				
CO3	Demonstrate understanding of different concepts in linear algebra in solving computational problems related to vectors and matrices and apply them to solve problems arising the Engineering especially in AI and ML.( K2, K3, K5)				
CO4	Understand the concepts of various probability distributions and apply them to analyze various engineering problems and make inference about the system (K2, K3, K4)				

CO5	Understand the method of linear and nonlinear least squares method and apply it to choose appropriate mathematical functions for modelling real data sets, arising from engineering disciplines (K3, K4, K5)
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Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	0	2	3	3	2	3	3	3	2	3	3
<b>CO2</b>	K3	3	3	2	2	2	3	1	1	3	3	2	1	3	3
<b>CO3</b>	K2	3	2	1	2	1	2	3	3	3	3	3	0	3	2
<b>CO4</b>	K3	3	3	2	1	2	3	2	0	0	0	3	2	3	3
<b>CO5</b>	K3	3	3	1	2	2	3	3	2	3	3	1	2	3	3
<b>Course</b>	K3	3	3	2	2	2	3	3	2	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain



BSC	<b>Course Code:</b> PYT1205	<b>Course Title: Applied Physics</b>	<b>Credits = 2</b>		
	<b>Semester: I</b>	<b>Total contact hours: 30</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>2</b>	<b>0</b>	<b>0</b>
<b>Course Outcomes (students will be able to...)</b>					
1	Assign Miller indices to various crystallographic planes and directions in a crystal lattice, thereby understand periodicity in the crystal lattice.				
2	Analyze a given x-ray diffraction pattern to deduce the crystal structure of the material and calculate the values of the basic structural parameters.				
3	Classify solids, and in turn semiconductors, based on electron occupancy and calculate basic quantities related to charge transport in them.				
4	Analyze simple ideal fluid flows by applying the continuity equation and Bernoulli's equation.				
5	Describe the basic behaviour of viscous flows and the relationships between various flow parameters.				
6	Understand simple models that are used to describe viscoelastic flows.				
<b>List of Prerequisite Courses</b>					
1	Standard XI and XII Physics course				
2	Standard XII Chemistry course				
<b>List of Courses where this course will be prerequisite</b>					
1	Applied Physics Laboratory (Sem-II)				
2	Materials Science Minor program courses (Sem-III, IV, V, VI, VII, VIII)				
3	Open Elective courses from Physics Department (Sem-II, IV, V)				
<b>Description of relevance of this course in the B. Chem.Tech. Program</b>					
The physics of solids and fluids play a key role in the various areas of chemical technology. The Applied Physics course will provide the students with the necessary fundamentals to develop a broad understanding of various aspects related to solids and fluids, and thereby equip them with the ability to apply it wherever required in their course of study.					
	<b>Course Contents (Topics and subtopics)</b>				<b>Reqd. hours</b>
<i>Solid State Physics</i>					
1	Crystal Structure of Solids: A revision of concepts of a lattice, a basis, unit cell, different crystal systems (SC, BCC, FCC, HCP), co-ordination number and packing fractions. Single crystalline, Polycrystalline, and Amorphous materials.				3
2	Crystallographic planes and directions: concept of Miller indices and its determination, examples; calculation of inter-planar spacing in terms of Miller indices.				3
3	Determination of crystal structure using X-rays: Bragg's law of X-ray diffraction, types of diffractometers, Indexing diffraction peaks and calculation of various lattice parameters and crystallite size				4
4	Energy band in solids and classification of solids, the concept of Fermi level and Fermi distribution function, Intrinsic and extrinsic semiconductors, Transport properties of semiconductors: Conductivity in semiconductors and its dependence of carrier concentration and mobility.				5
<i>Physics of Fluids</i>					
5	A revision of the basic concepts of hydrostatics and ideal fluid flow: Equation of continuity and Bernoulli's equation.				4
6	The concept of viscosity, Newton's law of viscosity, Reynold's number, Poiseuille's equation for streamline flows				4
6	An introduction to Rheology: Parameters of viscous flows, Newtonian and non-Newtonian behaviour, Variation of viscosity with shear rate, shear time, temperature, and pressure (qualitative ideas with illustrative examples), measuring properties of viscous flows. The concept of viscoelasticity, Maxwell and Kelvin models of relaxation, relaxation spectrum, creep testing.				7
				<b>Total</b>	<b>30</b>
<b>List of Textbooks/Reference books</b>					
1	Fundamentals of Physics – Halliday, Resnick, Walker – 6 <sup>th</sup> Edition – John Wiley				
2	Sears and Zeemansky's University Physics – Young and Freedman – 12 <sup>th</sup> Edition – Pearson Education				
3	A Textbook of Engineering Physics – M N Avadhanulu, P G Kshirsagar, TVS Arun Murthy – 11 <sup>th</sup> Edition – S. Chand Publishers				
4	Solid State Physics – S. O. Pillai – 10 <sup>th</sup> Edition – New Age Publishers				
5	Solid State Physics – A. J. Dekker – MacMillan India				
6	Engineering Physics – V Rajendran – 6 <sup>th</sup> Edition – McGraw Hill Publishers				
7	Introduction to Rheology – H. A. Barnes, J. F. Hutton and K. Walters – 4 <sup>th</sup> Edition – Elsevier Science.				
8	Viscoelastic Properties of Polymers – J. D. Ferry – 3 <sup>rd</sup> Edition – Wiley				

<b>Course Outcomes (Students will be able to.....)</b>	
CO1	Apply acoustic cavitation of Chemical Engineering Processes. (K3)
CO2	Apply Bernoulli equation in simple pipe flows. (K3)
CO3	Introduced to the principles of lasers, types of lasers and applications. (K2)
CO4	Calculate resolving power of instruments.(K3)
CO5	Describe principles of optical fibre communication.(K2)

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	1	1	3	3	3	3	2	3	3
<b>CO2</b>	K3	3	1	2	1	2	3	3	3	3	3	0	2	1	3
<b>CO3</b>	K2	3	2	1	2	0	3	3	3	3	2	3	1	3	2
<b>CO4</b>	K3	2	3	2	1	2	2	0	2	3	3	3	2	0	3
<b>CO5</b>	K2	3	2	1	2	0	0	3	3	1	3	1	1	3	2
<b>Course</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
 K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

VSEC	Course Code: GET1305	Course Title: Engineering Graphics and Computer Aided Drawing	Credits = 3		
	Semester: I	Total Contact Hours: 75	L	T	P
<b>List of Prerequisite Courses</b>					
Mathematics, Geometry, basic drawing and visualization					
<b>List of Courses where this course will be prerequisite</b>					
Industrial drawing, Equipment Design, Manufacturing and designing of any component, industrial 3D product modelling etc.					
<b>Description of relevance of this course in the B. Tech. Program</b>					
Drawing is a language used by engineers and technologists. A student is required to know the various processes and the equipment used to carry out the processes. Some of the elementary areas like product sizing, manufacturing etc., are very common to all the branches of technology. These and many other processes require machines and equipment's. One should be familiar with the design, manufacturing, working, maintenance of such machines and equipments. The subject of "drawing" is a medium through which, one can learn all such matter, because the "drawings" are used to represent objects and various processes on the paper. Through the drawings, a lot of accurate information is conveyed which will not be practicable through a spoken word or a written text. This course is required in many subjects as well as later in the professional career.					
<b>Course Contents (Topics and Subtopics)</b>					<b>Required Hours</b>
1	<b>Orthographic projections:</b> Introduction, Principles of Projection, Methods of Projection, Planes of projection, Quadrants, First-angle method of projection, Third-angle method of projection, and concept of orthographic projections.				20
2	<b>Sectional Projections and Missing Views:</b> Need for the drawing sectional views, concept of sectioning and section lines, Sectional drawings of different solids and machine components, Auxiliary planes, and views. <b>Missing Views:</b> Concept of recognizing missing views and their interpretation, drawing of missing views from given orthographic drawings.				15
3	<b>Isometric projections:</b> Concept of isometric views, isometric projections and isometric scale, Iso metric projections of different solids and machine components				15
4	<b>Computer Aided Drafting and Assembly drawing:</b> Basic introduction to CAD softwares, Design and Development of new products, Application of CAD, 2D, 3D part modelling on softwares, drawing modification and dimensioning, modelling of different machine components. Basics of Assembly drawing, preparation of 2D, 3D components and assembling on CAD software, conversions, labelling and table creation for bill of materials.				25
<b>Total</b>					<b>75</b>
<b>List of Textbooks/Reference Books</b>					
1	Engineering Drawing by N.D.Bhat				
2	Engineering Drawing by N.H.Dubey				
3	CAD/CAM: Theory and Practice by Ibrahim Zeid and R Sivasubramanian				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Draw Orthographic and Sectional Orthographic Views from Pictorial View 2 Draw isometric view when Front View and either top view or side view is given. 3 Understand basics of Assembly Drawing 4 Understand basics of CAD and Prepare 2D,3D drawings using CAD.				
CO2	Draw Orthographic and Sectional Orthographic Views from Pictorial View 2 Draw isometric view when Front View and either top view or side view is given. 3 Understand basics of Assembly Drawing 4 Understand basics of CAD and Prepare 2D,3D drawings using CAD.				
CO3	Draw Orthographic and Sectional Orthographic Views from Pictorial View 2 Draw isometric view when Front View and either top view or side view is given. 3 Understand basics of Assembly Drawing 4 Understand basics of CAD and Prepare 2D,3D drawings using CAD.				
CO4	Draw Orthographic and Sectional Orthographic Views from Pictorial View 2 Draw isometric view when Front View and either top view or side view is given. 3 Understand basics of Assembly Drawing 4 Understand basics of CAD and Prepare 2D,3D drawings using CAD.				

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
<b>CO2</b>	K3	3	2	2	2	2	3	3	3	1	2	3	2	3	1
<b>CO3</b>	K3	3	3	2	2	1	1	3	3	3	3	3	2	2	3
<b>CO4</b>	K3	3	3	2	2	2	3	0	2	3	3	3	2	3	1
<b>Course</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

ESC	<b>Course Code:</b> PST1101	<b>Course Title</b> <b>Spl 1 - Polymer Science &amp; Technology I</b>	<b>Credits = 2</b>		
	<b>Semester: I</b>	<b>Total Contact Hours: 30</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>1</b>	<b>1</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
HSC (Science)					
<b>List of Courses where this course will be Prerequisite</b>					
Raw materials Analysis & Characterization for Resin and Polymers, Analysis & Characterization of Resin and Polymers, Technology of Thermoset, Technology of Thermoset Polymers					
<b>Description of relevance of this course in the B. Tech. (Surface coating Tech.) Program</b>					
To train the students with respect to basics of polymers, Overview of Polymer and Coating Industry Manufacturing Chemistry, properties applications of monomers for synthetic and natural polymers and their handling hazards.					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	Overview of Polymer and Coating Industry, Historical developments in polymeric materials with introduction and classification of polymers				5
2	Basic concepts & definitions: monomer & functionality, oligomer, polymer, repeating units, degree of polymerization, molecular weight & molecular weight distribution commodity engineering polymers specialty polymer definitions				15
3	Manufacturing Chemistry, properties applications of raw material for synthetic polymers like Ethylene, propylene, butadiene, vinyl chloride, vinylidene dichloride, styrene etc.				10
	<b>Total</b>				<b>30</b>
<b>List of Text Books/ Reference Books</b>					
1	Raw Materials for Industrial Polymers by H Ulrich, Hanser Publication 1989.				
2	Principles of Polymer Science, by Bahadur and Sastry, Narosa Publishing House 2002.				
3	Polymer Science by Gowariker, Johan Wiley and Sons 1986.				
4	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc 1965.				
5	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc 1988.				
6	Petrochemicals: The Rise of an Industry by Peter H. Spitz, Johan Wiley and sons 1988.				
7	Polymer Chemistry by Malcolm P. Stevens, Oxford University Press, Inc, 1990				
<b>Course Outcomes (Students will be able to.....)</b>					
CO2	Interpret the physical and chemical properties of raw materials (K3)				
CO3	Analyze the manufacturing routes and impurities in monomers and raw materials (K4)				
CO4	Discuss about the environmental concerns handling Safety and Hazards of Monomers (K2)				
CO5	Propose plan about evaluation of raw materials and reactants for synthesis & manufacturing of resins and polymers. (K5)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K2	3	3	2	3	3	3	2	1	1	1	1	2	2	2
CO2	K3	3	3	1	1	1	3	3	1	1	2	2	3	3	3
CO3	K4	3	3	3	3	2	3	3	2	2	1	1	3	3	3
CO4	K5	2	3	3	2	3	1	3	3	2	2	3	2	2	1
Course	K5	3	3	3	3	3	3	3	3	2	2	3	3	2	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

BSC	<b>Course Code:</b> PYP1101	<b>Course Title:</b> Physics Laboratory	<b>Credits = 2</b>		
	<b>Semester: I</b>	<b>Total Contact Hours: 60</b>	<b>L</b> 0	<b>T</b> 0	<b>P</b> 4
<b>List of Prerequisite Courses</b>					
Applied Physics					
<b>List of Courses where this course will be prerequisite</b>					
Independently set up, handle, and use basic setups to measure and obtain various physical quantities. Use basic instruments like vernier-caliper, screw-gauge, travelling microscope, thermometer, etc. to make accurate measurements. Correlate and use directly measured quantities to obtain the relevant parameters through appropriate formulae, calculations, and/or graphical plotting, thereby understand the measurement principle involved in the experimental setups. Preliminarily treat the obtained datasets statistically to obtain errors in the experiments.					
<b>Description of relevance of this course in the B. Tech. Program</b>					
The hands-on experience gained by the students in the Applied Physics laboratory course will equip them with basic experimental skills related to measurement of various important physical quantities. These skills will act as a useful foundation for other laboratory and theory courses in their area of specialization.					
<b>Sr. No.</b>	<b>Course Contents (Topics and Subtopics)</b>				<b>Required Hours</b>
1	Determination of Co-efficient of Viscosity by Poiseuille's method				5
2	Thermistor characteristics: Determination of Bandgap of a semiconductor				6
3	Determination of compressibility of liquids using an Ultrasonic Interferometer				5
4	Measurement of thermal conductivity of a solid: Lee's disc method				6
5	Photoelectric effect: Determination of h/e				5
6	Hall effect: Determination of carrier type and concentration in a semiconductor				6
7	Newton's rings: Determination of wavelength of light				5
8	Laser Diffraction: Determination of particle size				8
9	Determination of Co-efficient of Viscosity by Poiseuille's method				8
10	Thermistor characteristics: Determination of Bandgap of a semiconductor				6
<b>Total</b>					<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Fundamentals of Physics - Halliday, Resnick, Walker - 6 <sup>th</sup> Edition - John Wiley				
2	Sears and Zeemansky's University Physics - Young and Freedman - 12 <sup>th</sup> Edition - Pearson Education				
3	A Textbook of Engineering Physics - M N Avadhanulu, P G Kshirsagar, TVS Arun Murthy - 11 <sup>th</sup> Edition - S. Chand Publishers				
4	Engineering Physics - V Rajendran - 6 <sup>th</sup> Edition - McGraw Hill Publishers				
5	Concepts of Modern Physics - A. Beiser, McGraw-Hill.				
6	Ultrasonics: Methods and Applications - J. Blitz, Butterworth.				
7	Optics - Ajoy Ghatak - 7 <sup>th</sup> Edition - McGraw Hill				
8	Fundamentals of Optics - F. Jenkins and H. White - 4 <sup>th</sup> Edition McGraw Hill				
9	ICT Physics Laboratory Manual (supplied to students)				
<b>Course Outcomes (students will be able to.....)</b>					
<b>CO1</b>	Apply various laws which they have studied through experiments (K3)				
<b>CO2</b>	Measure transport properties like viscosity, conductivity, etc.(K4)				
<b>CO3</b>	Explain the application of acoustic cavitation (K2)				

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	1	3	3	3	3	3	3	2	3	3
<b>CO2</b>	K4	3	3	2	3	2	3	3	2	3	3	3	0	2	3
<b>CO3</b>	K2	3	2	1	2	0	3	3	3	3	1	3	1	3	2
<b>Course</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
 K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

AEC	<b>Course Code:</b> HUT1110B	<b>Course Title:</b> Communication Skills-English	<b>Credits = 2</b>		
	<b>Semester: I</b>	<b>Total Contact Hours: 60</b>	<b>L</b> 0	<b>T</b> 0	<b>P</b> 4
<b>List of Prerequisite Courses</b>					
Standard XII <sup>th</sup> English					
<b>List of Courses where this course will be prerequisite</b>					
All courses in this and subsequent semesters					
<b>Description of relevance of this course in the B. Tech. Program</b>					
This is an important course for the effective functioning of an Engineer and a Technologist. Communication skills are required in all courses and professional career.					
<b>Sr. No.</b>	<b>Course Contents (Topics and Subtopics)</b>				<b>Required Hours</b>
1	Development of communication skills in oral as well as writing				10
2	The writing skills should emphasize technical report writing, scientific paper writing, letter drafting, etc.				14
3	The oral communication skills should emphasize presentation skills.				10
4	Use of audio-visual facilities like powerpoint, LCD. for making effective oral Presentation				14
5	Group Discussions				12
<b>Total</b>				<b>60</b>	
<b>List of Text Books/ Reference Books</b>					
1	Elements of Style – Strunk and White				
<b>Course Outcomes (students will be able to.....)</b>					
CO1	write grammar error free technical reports in MS Word or equivalent software.(K3)				
CO2	make power point slides in MS PowerPoint or equivalent software.(K3)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	3	3	1	2	3	3
<b>CO2</b>	K3	3	3	2	0	2	3	1	3	3	2	3	2	3	3
<b>Course</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain



# Semester-II

BSC	Course Code: CHT1407	Course Title: Organic Chemistry	Credits = 3		
	Semester: II		Total Contact Hours: 45	L	T
			2	1	0
<b>List of Prerequisite Courses</b>					
This is a Basic Organic Chemistry course. The Organic Chemistry studied at HSC is the basis for building up Advanced Organic Chemistry knowledge.					
<b>List of Courses where this course will be Prerequisite</b>					
Organic Chemistry, Biochemistry and several Special Subjects of Chemical Technology Departments					
<b>Description of relevance of this course in the B. Tech. (Pharm. Chem. Tech.) Programme</b>					
To acquaint the students with IUPAC and other types of Nomenclature of organic compounds, fundamentals of Organic Chemistry including reaction mechanisms, organic transformations, types of reactions, selectivity of chemical transformations, etc., stereochemical implications of organic reactions, functional group identification and reactions					
Sr. No.	Course Contents (Topics and Subtopics)				Required Hours
1	<b>Chemistry of Carbonyl Compounds</b> Concept of acidity and tautomerism of carbonyl compounds, General methods of preparation and Nucleophilic Addition reactions Enolate chemistry, Aldol and related condensation reactions, Michael reaction, Robinson annulation, Claisen condensation, Dieckmann condensation, Mannich reaction.				9
2	<b>Aromatic Substitution Reactions</b> <b>A) Electrophilic Substitution Reactions</b> Nitration, Halogenation, Alkylation, Acylation and Sulfonation Activating, deactivating and orienting effects of functional groups in mono- and poly-substituted benzenes Friedel-Crafts alkylation, Acylation, Gattermann, Gattermann-Koch, Riemer-Tiemann reactions. <b>B) Nucleophilic Substitution Reactions</b> Addition and elimination mechanism, Benzyne mechanism, Sandmeyer reaction.				10
3	<b>Heteroaromatic Compounds</b> IUPAC nomenclature, structures and common names, comparison with benzenoid compounds, reactivity and synthesis – pyrroles, furans, thiophenes and pyridines				8
5	<b>Named Organic Reactions</b> Perkin reaction (Mauvine synthesis-dyes), Fischer indole synthesis, (dyes), Jacobson Corey epoxide synthesis (Pharmaceutical), Ziegler Natta polymerisation (polymer), Multicomponent reactions, Mailard reaction (foods), Strecker amino acid synthesis (Pharmaceuticals & Food), Wittig reactions, Prilezhaev reaction				10
6	<b>Stereochemistry of Organic Compounds</b> Containing one and two asymmetric carbon atoms, Stereo descriptors – R/S, E/Z, erythro and thero, Conformation – Ethane and butane. Enantiomers and Diastereomers, meso compounds, different representations of stereoisomers – Saw-horse, Newmann, Wedge and dash and Fischer and their interconversions				8
<b>Total</b>					<b>45</b>
<b>List of Text Books/Reference Books</b>					
1	Clayden, J., Greeves, N., Warren, S.; Organic Chemistry; 2 <sup>nd</sup> ed.; Oxford University Press (2012)				
2	Graham Solomons, T. W.; Fryhle, Craig B.; Snyder, Scott A. Organic Chemistry; 12 <sup>th</sup> Ed.; John Wiley & Sons. Inc. (2016)				
3	Smith, M. B.; March's Advanced Organic Chemistry: Reactions, Mechanisms and Structure; 7 <sup>th</sup> ed.; Wiley, India (2015)				
4	Carey F. A., Sundberg, R. J. Advanced Organic Chemistry: Part A: Structure and Mechanisms; 5 <sup>th</sup> ed.; Springer (2005)				

5	Carey F. A., Sundberg, R. J.; Advanced Organic Chemistry: Part B: Reaction and Synthesis; 5 <sup>th</sup> ed.; Springer (2007)
6	Wade, L. G.; Simek, J. W.; Singh, M. S. Organic Chemistry; 9 <sup>th</sup> ed.; Pearson Education (2019)
7	Eliel, E. L. Stereochemistry of Carbon Compounds; Mcgraw-Hill (2001)
8	Bruice, Paula, Y. Organic Chemistry; 8 <sup>th</sup> Ed.; Pearson Education (2020)

<b>Course Outcomes (Students will be able to.....)</b>	
CO1	Draw structures of organic compounds and write their IUPAC names correctly (K2).
CO2	be well versed with aromatic chemistry and interpret the outcome of general transformations (K3).
CO3	Understand the importance of heterocycles, learn the properties and synthetic routes, interpret the IUPAC of compounds and decipher outcomes of various transformations involving heterocycles (K3).
CO4	Apply the knowledge obtained through the course to predict the outcome of reactions and devise solutions to unknown problems (K3).
CO5	Appreciate the stereo-chemical implications of organic compounds and visualize and appreciate the chirality concept (K2).
CO6	Understand organic chemistry reactions related to aliphatic as well as aromatic compounds as well as decipher the outcome of a given organic transformation (K3).
CO7	Interpret and analyze reactions having different functionalities, deduce and solve problems related to the reactions as well as apply them, if need be (K4).

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K2	3	2	0	2	1	3	3	2	3	3	3	1	3	2
<b>CO2</b>	K2	3	2	0	1	0	3	3	1	2	3	2	0	3	2
<b>CO3</b>	K3	3	3	1	2	2	3	1	3	3	2	3	2	3	3
<b>CO4</b>	K4	3	3	1	3	2	3	2	3	2	3	3	2	3	3
<b>CO5</b>	K4	3	3	1	3	2	3	2	3	2	3	3	2	3	3
<b>CO6</b>	K4	3	3	1	3	2	3	2	3	2	3	3	2	3	3
<b>CO7</b>	K4	3	3	1	3	2	3	2	3	2	3	3	2	3	3
<b>Course</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
 K, Knowledge-level from cognitive domain; A, Affective domain; P, Psychomotor domain

BSC	Course Code: CHT1408	Course Title: Industrial Chemistry	Credits = 3		
	Semester: II	Total Contact Hours: 45	L	T	P
<b>List of Prerequisite Courses</b>					
Standard XII Inorganic Chemistry					
<b>List of Courses where this course will be Prerequisite</b>					
Material Technology, Environment Science and Technology					
<b>Description of relevance of this course in the B. Tech. Programme</b>					
To acquaint the students with synthesis, properties and applications of various industrial inorganic chemicals					
Sr. No.	Course Contents (Topics and Subtopics)				Required Hours
1	Introduction to Chemical Industry: Bulk chemicals, fine chemicals, intermediates, active pharmaceutical ingredients (API), etc.				3
2	Petrochemical Industry: operations and processes in manufacture of ethers, hydrocarbons, aromatic compounds, etc.				6
3	PRIMARY INORGANIC MATERIALS: Water, Hydrogen, Hydrogen Peroxide and Inorganic Peroxo Compounds, Nitrogen and Nitrogen Compounds, Phosphorus and its Compounds, Sulfur and Sulfur Compounds, Halogens and Halogen Compounds,				8
4	MINERAL FERTILIZERS: Phosphorus-Containing Fertilizers, Nitrogen-Containing Fertilizers, Potassium-Containing Fertilizers				4
5	METALS AND THEIR COMPOUNDS: Alkali and Alkaline Earth Metals and their Compounds Aluminum and its Compounds, Chromium Compounds and Chromium, Silicon and its Inorganic Compounds, Manganese Compounds and Manganese				8
6	ORGANIC BULK CHEMICALS: Manufacture of methanol, acetic acid, ethanol, ethylene, propylene, butadiene, acetaldehyde, acetylene, BTX, alkyl benzenes, acetone, phenol, styrene, esters, ethylene oxide, phthalic acid, Vinyl-Halogen and Vinyl-Oxygen Compounds, azo dyes, Polyamides, Propene Conversion Products, Aromatics - Production and Oxidation Products of Xylene and Naphthalene				8
7	Important pharmaceutically active ingredients, agrochemicals, insecticides, pesticides, perfumery chemicals.				8
<b>Total</b>					<b>45</b>
<b>List of Text Books/ Reference Books</b>					
1	Industrial Organic Chemistry, 3rd, Completely Revised Edition, Klaus Weissermel, Hans-Jürgen Arpe ISBN: 978-3-527-61459-2 July 2008.				
2	Industrial Inorganic Chemistry, 2nd Completely Revised Edition, Karl Heinz Buchel, Hans-Heinrich Moretto, Dietmar Werner, ISBN: 978-3-527-61333-5, 667 pages, November 2008, Wiley-VCH.				
3	Inorganic Chemistry – an industrial and environmental perspective, T.W. Swaddle, ISBN 0-12-678550-3, 482 pages, Academic Press				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Understand the important of chemical principles applied to various industrial processes				
CO2	Describe the fundamental processes underlying manufacture of important organic and inorganic chemicals				
CO3	Review and assess the impact of the chemical factors on the efficiency of industries and feedstock manufacturing				
CO4	Modify existing applications for improving the efficiencies in terms of yields, energy requirement and				

	environmental impact
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Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K2	3	2	1	2	0	3	2	3	3	3	3	1	3	2
<b>CO2</b>	K3	3	3	2	2	2	3	3	1	3	3	2	2	3	3
<b>CO3</b>	K2	3	2	0	2	1	3	3	3	3	0	3	1	2	1
<b>CO4</b>	K2	3	2	1	2	1	2	3	3	3	3	1	1	3	2
<b>Course</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

PCC	Course Code:	Course Title	Credits = 2		
	SCT1201		L	T	P
	Semester: II	Total Contact Hours: 30	1	1	0
<b>List of Prerequisite Courses</b>					
HSC (Science), Polymer science and technology I					
<b>List of Courses where this course will be Prerequisite</b>					
Introduction to coating technology, Polymer chemistry and technology, Polymer Science and Technology II, Raw Material Analysis for Resins and Polymers, Synthesis and Characterization of Resins and Polymers Lab I					
<b>Description of relevance of this course in the B. Tech. (Surface coating Tech.) Programme</b>					
To teach students basic concepts of Polymer chemistry & Technology so that they can have good base to learn other subjects					
Sr. No.	Course Contents (Topics and Subtopics)				Required Hours
1	Introduction to the coating of materials and various substract				6
2	Various types of resin materials				8
3	Various techniques for the application of paint				8
4	Types of paint anticorrosive, decorative, flame-retardant, antimicrobial, hydrophobic				8
<b>Total</b>				<b>30</b>	
<b>List of Text Books/ Reference Books</b>					
1	Surface coating (Volume 1 )Oils and color coating association Australia				
2	Basic of paint technology by V.C. malshe				
3	Outlines of paint technology by morgan (hard cover)				
4	Resins for surface coating by P.K.T.olding				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Describe the fundamental principles of coating technology and its significance in protecting and enhancing material properties (K2)				
CO2	Interpret the characteristics and applications of various types of resin materials used in coating technology (K3)				
CO3	Analyze the different techniques employed for the application of paint, evaluating their advantages and limitations in specific contexts (K4)				
CO4	Explain the distinct properties and purposes of anticorrosive, decorative, flame-retardant, antimicrobial, and hydrophobic paints, and compare their effectiveness in diverse settings (K2)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K1	3	1	3	2	1	2	2	1	2	2	2	3	3	3
CO2	K2	3	2	2	2	1	2	3	1	2	2	2	3	3	3
CO3	K4	3	3	3	3	2	3	3	1	1	1	2	3	3	3
CO4	K2	3	3	3	3	1	2	1	1	1	2	2	3	3	2
Course	K4	3	3	3	3	2	3	3	1	2	2	2	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

ESC	Course Code: GET1306	Course Title: Basic Mechanical Engineering	Credits = 2		
	Semester: II	Total Contact Hours: 30	L	T	P
			1	1	0
<b>List of Prerequisite Courses</b>					
Physics, Basic Mathematics					
<b>List of Courses where this course will be Prerequisite</b>					
Energy Engineering, Unit Operations, Mechanical design of chemical equipments					
<b>Description of relevance of this course in the B. Tech. Programme</b>					
Students will be able to understand various equipments like steam turbine, gas turbine, pumps, compressors, and power transmission system.					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	<b>Introduction- Concept of Stress:</b> Condition of Equilibrium for concurrent coplanar and non-concurrent coplanar forces. Deformation in solids- Hooke's law, stress and strain- tension, compression and shear stresses, Stress Strain Diagram, elastic constants and their relations volumetric, linear and shear strains.				6
2	<b>Introduction to Thermodynamics:</b> First Law of Thermodynamics, Steady-flow energy equation, Second Law of Thermodynamics				4
3	<b>Basics of Power Station</b> -Steam Generators Fire tube and Water tube boiler, Low pressure, and high-pressure boilers, Mountings and accessories, Boiler efficiency -Steam Turbines Working principle of steam, gas and water turbines, Concept of impulse and reaction steam turbines. -Compressors/Pumps Different Types of Compressors and their applications, Different Types of Pumps, and their applications				8
4	<b>Transmission of Power:</b> Introduction to various drives such as belt, rope, chain and gear drives, Introduction to mechanical elements such as keys, couplings, and bearings in power transmission (No numerical)				4
5	<b>Refrigeration and Air-conditioning</b> Vapour compression refrigeration cycle, Vapour absorption refrigeration systems, Properties of air such as DBT, WBT, DPT, relative humidity, Psychometric chart.				4
6	<b>Renewable Energy</b> Role and importance of non-conventional and alternate energy sources such as solar, wind, ocean, bio-mass and geothermal, hydrogen energy				4
<b>Total</b>					<b>30</b>
<b>List of Text Books/ Reference Books</b>					
1	Strength of Materials by S. Ramamrutham, Dhanpat Rai Pvt. Ltd				
2	Thermodynamics by P.K. Nag				
3	Power plant by Morse				
4	Heat Engines by P.L. Balani				
5	Hydraulic Machines by Jagdish Lal				
6	Renewable Energy resources by Tiwari and ghosal, Narosa publication.				
7	Non-conventional energy sources, Khanna publications				
8	Refrigeration and air conditioning by C.P. Arora				
9	Theory of Machines by Rattan. S.S				
10	Gas turbine theory by HiH Saravanamutoo				

Course Outcomes (Students will be able to.....)	
CO1	Understand different types of stresses and their effects on bodies. (K2)
CO2	Describe the working of steam boilers, mountings, and accessories. ( K2)
CO3	Explain the working principles of power developing systems such as steam turbines, gas turbines and internal combustion engines. (K2)
CO4	Describe the working principle of vapour compression and vapour absorption refrigeration systems. (K2)
CO5	Discuss different types of power transmission systems and their typical applications. (K2)
CO6	Explain the working principles of power absorbing devices such as pumps and compressors. (K2)

Mapping of Course Outcomes (Cos) with Programme Outcomes (Pos)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
<b>CO2</b>	K2	3	1	0	2	1	3	1	3	3	3	3	1	3	2
<b>CO3</b>	K3	3	3	2	2	2	3	3	3	3	2	3	2	2	3
<b>CO4</b>	K2	3	2	1	2	0	3	3	2	3	3	3	1	3	2
<b>CO5</b>	K2	3	2	1	2	0	3	3	2	3	3	3	1	3	2
<b>CO6</b>	K2	3	2	1	2	0	3	3	2	3	3	3	1	3	2
<b>Course</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain



ESC	Course Code: GET1125	Course Title: Electrical Engineering and Electronics	Credits = 2		
	Semester: II	Total Contact Hours: 30	L	T	P
<b>List of Prerequisite Courses</b>					
Standard XII Physics and Mathematics courses					
<b>List of Courses where this course will be prerequisite</b>					
Various Technology Courses and Professional Career					
<b>Description of relevance of this course in the B. Tech. Program</b>					
In this course, Students will get an insight to the importance of Electrical Energy in Chemical Plants. The students will understand the basics of electricity, selection of different types of drives for a given application process. They will get basic knowledge as regards to Power supplies, instrumentation amplifiers and thyristor application in industries.					
Sr. No.	Course Contents (Topics and Subtopics)				Required Hours
1	<b>Fundamentals of DC Circuits</b> Voltage and Current Sources, Basic Laws, Network Theorems, Superposition Theorem and Thevenin's Theorem,				4
2	<b>AC Fundamentals:</b> A.C. through resistance, inductance and capacitance, simple RL, RC and RLC circuits. Power, power factor				4
3	<b>Three Phase Systems:</b> Three phase system of emfs and currents, Star and Delta connections, three phase power				5
4	<b>Single phase transformers:</b> Principle of working, Efficiency, regulation.				5
5	<b>Electrical drives:</b> Basic concepts of different types of Electrical motors as drives, Their suitability for various applications.				5
6	<b>Regulated power supplies,</b> Diodes as rectifiers, Half wave and Full wave rectifier, Filters and Regulators				5
7	<b>Bipolar junction transistors:</b> Different configurations, Characteristics, Concept of basic amplifier circuits, Amplifier gain, Transistor as switch				3
8	<b>Introduction to Integrated circuits:</b> Basic concepts of ICs				2
9	<b>Introduction to data acquisition and signal conditioning,</b> Basic concept and Block diagram, Concept of conversion of physical quantity to electrical signal, signal conditioning, Introduction to A/D and D/A converters				3
10	<b>Introduction to instrumentation amplifiers and their applications</b> Operational Amplifier – Notation, Pin diagram, Differential and common mode gain, CMRR, Introduction to various applications such as Non-inverting, inverting amplifiers, adder, subtractor, integrator, differentiator,				3
<b>Total</b>					<b>45</b>
<b>List of Textbooks/Reference Books</b>					
1	Electrical Engineering Fundamentals by Vincent Deltoro				
2	Electronic devices and circuits by Boylestad, Nashelsky				
3	Electrical Machines by Nagrath, Kothari				
4	Electrical Technology by B.L. Theraja, A.K. Theraja vol I, II, IV				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	understand the basic concepts of D.C circuits. Solve basic electrical circuit problems.(K3)				
CO2	understand the basic concepts of single phase and three phase AC supply and circuits.(K2)				
CO3	understand the basic concepts of transformers and motors used as various industrial drives.(K2)				
CO4	understand the basic concepts of electronic devices and their applications.(K2)				

Mapping of Course Outcomes (Cos) with Programme Outcomes (Pos)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
<b>CO2</b>	K2	3	2	0	2	1	3	3	3	3	2	3	0	3	2
<b>CO3</b>	K2	3	2	1	2	0	3	3	2	3	3	3	1	3	2
<b>CO4</b>	K2	3	0	1	2	1	2	3	3	1	3	1	1	2	2
<b>Course</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

ESC	Course Code: CEP1720	Course Title: Process Calculations	Credits = 2		
	Semester:	Total contact hours: 60	L	T	P
			0	0	4
<b>List of Prerequisite Courses</b>					
	XII <sup>th</sup> Standard Mathematics, Chemistry, Physics				
<b>List of Courses where this course will be prerequisite</b>					
	This is a basic Course. This knowledge will be required in ALL subjects later.				
<b>Description of relevance of this course in the B. Tech. Program</b>					
This is a basic course. This knowledge will be required in almost all subjects later. This subject introduces the various concepts used in Chemical Engineering to the students. The knowledge of this subject is required for in All B. Tech. courses, etc. It can be applied in various situations such as process selection, economics, sustainability, environmental impacts					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Reqd. Hours</b>
1	Introduction to Chemical process calculations, overview of single stage and multistage operations, concept of process flow sheets				2
2	Revision of Units and Dimensions, Dimensional analysis of equations, Mathematical techniques				4
3	Mole concept, composition relationship, types of flow rates				2
4	Material balance in non-reacting systems: application to single and multistage processes				8
5	Stoichiometry				2
6	Material balance in reacting systems: application to single and multistage processes				6
7	Behavior of gases and vapors				4
8	Introduction to psychrometry, humidity and air-conditioning calculations.				6
9	Calculation of X-Y diagrams based on Raoult's law.				2
10	Applications of material balances to Multiphase systems				6
11	Basic concepts of types of Energy and calculations				2
12	Application of Energy balance to non-reacting systems				6
13	Application of Energy balance to reacting systems				6
14	Fuels and combustion.				4
	<b>Total</b>				<b>60</b>
<b>List of Text Books/ Reference Books</b>					
	Elementary Principles of Chemical Processes, Felder, R.M. and Rousseau,				
	Chemical Process Principles, Hougén O.A., Watson K. M.				
	Basic Principles and Calculations in Chemical Engineering, Himmelblau,				
	Stoichiometry, Bhatt B.I. and Vora S.M.				
<b>Course Outcomes (students will be able to.....)</b>					
1	Students will be able to convert units of simple quantities from one set of units to another set of units				
2	Students will be able to calculate quantities and /or compositions, energy usages, etc. in various processes and process equipment such as reactors, filters, dryers, etc.				

Mapping of Course Outcomes (Cos) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K2	3	2	0	2	1	3	3	3	3	3	3	1	3	2
<b>CO2</b>	K3	3	3	2	2	2	3	3	3	3	3	2	2	3	3
<b>Course</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

BSC	Course Code: CHP1343	Course Title: Physical and Analytical Chemistry Laboratory	Credits = 2		
	Semester: II	Total Contact Hours: 60	L	T	P
			0	0	4
<b>List of Prerequisite Courses</b>					
Standard XII <sup>th</sup> Chemistry Laboratory courses					
<b>List of Courses where this course will be prerequisite</b>					
This is a basic Course. This knowledge will be required in Applied Chemistry subjects later.					
<b>Description of relevance of this course in the B. Tech. Program</b>					
Students will become familiar with laboratory experimental skills, plan and interpretation of experimental tasks, understand the relevance of principles of physical chemistry in chemical processes					
Sr. No.	Course Contents (Topics and Subtopics)				Required Hours
1	(8 to 10 experiments will be conducted from following list) 1. To determine the total hardness of given water sample 2. To determine the dissociation constants of a polybasic acid using pH meter 3. To determine pK <sub>a</sub> of the given weak acid by potentiometric titration 4. To determine the critical micelle concentration (CMC) of the given surfactant by surface tension measurement using a stalagmometer 5. To determine the normality and volume of weak acid and strong acid in the given mixture using conductometric titration 6. To determine the rate constant of hydrolysis of an ester catalyzed by an acid 7. To study the kinetics of the reaction between K <sub>2</sub> S <sub>2</sub> O <sub>8</sub> and KI and hence, determine rate of the reaction 8. To verify Beer – Lambert's Law 9. To determine the equivalent conductance of strong electrolyte at infinite dilution and verify Ostwald's law of dilution, for dissociation of weak electrolyte 10. To determine the molecular weight of the given polymer by viscosity measurements 11. To determine the vitamin C concentration from the given tablet sample by titration 12. Demo of Gas chromatography and FT-IR.				4h per practical
<b>Total</b>					<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Practical physical Chemistry – B.Viswanthan and P.S. Raghavan				
2	Practical physical Chemistry- Alexander Findlay				
<b>Course Outcomes (students will be able to.....)</b>					
CO1	Identify reaction rate parameters				
CO2	List simple methods of chemical analysis				
CO3	Determination of physic chemical parameters using simple laboratory tools				

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	2	3	3	2	3	3
<b>CO2</b>	K4	3	3	1	3	1	2	3	1	3	3	0	2	3	3
<b>CO2</b>	K4	3	3	1	3	1	2	3	1	3	3	0	2	3	3
<b>Course</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
 K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

VSEC	Course Code: CHP1132	Course Title: Organic Chemistry Laboratory	Credits = 2		
	Semester: II		Total Contact Hours: 60	L	T
			0	0	4
<b>List of Prerequisite Courses</b>					
Standard XII <sup>th</sup> Organic Chemistry Laboratory					
<b>List of Courses where this course will be prerequisite</b>					
All the Applied Chemistry Practicals					
<b>Description of relevance of this course in the B. Tech. Program</b>					
The course is relevant for training the students for working with binary mixtures. The students are exposed to basics of organic separations and identification of organic compounds based on their physicochemical properties. The laboratory training is crucial for the students to carry out work-up of organic reactions leading to separation of crude products followed by purification using recrystallization and/or distillation or related methods.					
	<b>Course Contents (Topics and Subtopics)</b>				<b>Required Hours</b>
1	a) Principles of qualitative separation of organic mixtures using physical properties, chemical properties and their combination				4
	b) Principles of quantitative separation of organic mixtures using physical properties, chemical properties and their combination				4
2	a) Separation of solid-solid water insoluble binary organic mixtures				5X4
	b) Separation of solid-solid partly water soluble binary organic mixtures				2X4
	c) Separation of solid-solid mixtures by fractional crystallization				2X4
	d) Separation of liquid-liquid mixtures by distillation				2X4
	e) Separation of liquid-liquid mixtures by solvent extraction				2X4
	<b>Total</b>				<b>60</b>
<b>List of Textbooks/Reference Books</b>					
1	Arthur, Vogel. Textbook of Practical Organic Chemistry, 5 <sup>th</sup> edition, publishers Longman group Ltd, 1989				
2	F.G. Mann and B.C. Saunders, Practical Organic Chemistry, 4 <sup>th</sup> edition published by Orient Longman				
3	Keese, R, Martin P. B, and Trevor P. Toube. Practical Organic Synthesis: A Student's Guide. John Wiley & Sons, 2006.				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	work safely in the organic chemistry laboratory.(K3)				
CO2	separate binary organic mixtures by multiple techniques.(K4)				
CO3	understand basic principles for separation of binary organic mixtures qualitatively and quantitatively.(K3)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	3	3	3	0	3	3
<b>CO2</b>	K4	3	3	2	3	2	3	3	0	3	3	3	2	2	3
<b>CO3</b>	K3	3	1	2	1	2	2	3	3	3	3	1	2	3	1
<b>Course</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

# Semester-III

PCC	Course Code: PST1303	Course Title Spl 3- Polymer Chemistry & Technology	Credits = 4		
	Semester: III	Total Contact Hours: 60	L	T	P
<b>List of Prerequisite Courses</b>					
HSC (Science), polymer science and technology I, Introduction to coating technology					
<b>List of Courses where this course will be Prerequisite</b>					
High Polymer Chemistry , Structure Property Relationship, Compounding and Polymer Processing, Technology of Thermoplastics, Technology of Thermosets					
<b>Description of relevance of this course in the B. Tech. (Surface coating Tech.) Program</b>					
To teach students basic concepts of Polymer chemistry & Technology so that they can have good base to learn other subjects					
Sr. No.	Course Contents (Topics and subtopics)				Required Hours
1	Detailed classification of polymers Addition, condensation, commodity engineering and speciality copolymers, Monomer structure and Polymerizability. Crystalline/amorphous, step growth /chain growth, homochain / heterochain, crystalline / amorphous polymers, confirmation etc.				5
2	Homo& copolymers, graft, block alt, ladder etc. & nomenclature, configuration: cis/trans; tacticity, branched/ crosslinked, Addition and condensation polymerization mechanism				5
3	Techniques of polymerization: bulk, solution, suspension, emulsion, plasma etc.				5
4	Molecular weight and its distribution determination methods (Mn to Mz+1& MWD, Poly dispersity Index), calculations & problems based on it,				5
5	Carothers equation for condensation polymers & conditions to get high or desired molecular weight, calculations & problems based on it.				5
6	Transition temperatures such as Tg, Tc, Tm, their relevance to properties & processing and factors affecting them				5
7	Solubility parameter, solution properties, temperature, good/ bad solvent.				5
8	Different initiating systems such as free radical polymerization, redox with examples & their use choice of initiator half-life period. Measurement of polymer viscosity by different method				5
9	Copolymerization, reactivity ratios & kinetics of copolymerization (copolymer composition equation). Polymerization: Probability and statistics-statistics of polycondensation, chain polymerization, branching and gelation. Copolymer sequence distribution				5
10	Basic Rheological concepts of polymer solutions and melts , Newtonian / non Newtonian, time dependent/ independent				5
11	Mixing operations: Typical agitation system, dissolution, suspension, removal of water condensates high speed (low viscosity) stirring, low speed (high viscosity) stirring selection criterion, power consumption. Heat transfer characteristics, powder mixing times etc				5
12	Commercial applicability of Polymers as Plastics, paints, rubbers, fibers & adhesives				5
<b>Total</b>					<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Principles of Polymer Science, Bahadur and Sastry, Narosa Publishing House 2002				
2	Polymer Science, Gowarikar, Johan wiley and Sons 1986				
3	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc 1965				
4	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc 1988				
5	Polymer Chemistry, Malcolm P. Stevens, Oxford University Press, Inc, 1990.				
6	Text book of polymer Science, Billmeyer, John Wiley ans Sons 1984.				
7	Principles of Polymer Systems, Rodriguez, Hemisphere Publishing Corpn, 1982				
8	Introduction to Polymer Science and Technology, H. S. Kaufman and J. J. Falcetta, Wiley – Inter science Publication, 1977				
9	Principles of polymerization, G. Odian, Wiley – Inter science (1981)				
<b>Course Outcomes (Students will be able to.....)</b>					



CO1	Describe the basics of polymers, various terminologies and classifications of polymers. (K2)
CO2	Solve the problems regarding Calculation of MW – MWD & its relevance (K4)
CO3	Explain the basics of rheology & its effect on processing & application, mixing operations. (K2)
CO4	Compare various techniques of polymerization & initiating systems (K4)
CO5	Differentiate the various types of copolymerization & their commercial applications. (K4)
CO6	Discuss different mixing operations and operating parameters

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PS O1	PSO 2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+A	K3	K6+A+Psy	K3	K4
CO1	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
CO2	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO3	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO5	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
Course	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

PCC	<b>Course Code: PST1304</b>	<b>Course Title: Spl 4-Polymer Science &amp; Technology II</b>	<b>Credits = 2</b>		
	<b>Semester: III</b>	<b>Total Contact Hours: 30</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>1</b>	<b>1</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
HSC (Science), polymer science and technology I, Introduction to coating technology					
<b>List of Courses where this course will be Prerequisite</b>					
Raw materials Analysis & Characterization for Resin and Polymers, Analysis & Characterization of Resin and Polymers, Technology of Thermoset , Technology of Thermoset Polymers					
<b>Description of relevance of this course in the B. Tech. (Surface coating Tech.) Program</b>					
To train the students with respect to basics of polymers, Overview of Polymer and Coating Industry Manufacturing Chemistry, properties applications of monomers for synthetic and natural polymers and their handling hazards.					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	Natural Polymers: Chemical & Physical structure, properties, source, important chemical modifications, applications of polymers such as Lignin, starch, rosin, shellac, latexes etc.				5
2	Ethyl Cellulose Methyl Cellulose Nitro Cellulose, Cellulose acetates etc.				2
3	Vegetable oils and gums, proteins etc.				2
4	Polyols like ethylene glycol propylene ethylene glycol and their modification etc				3
5	Acrylic monomers like acrylic acid, acrylonitrile, methacrylic acid, methacrylate, acrylamide etc				3
6	Azelic acid sabacic acid aminododacnoic acid etc				2
7	Phenol modified phenols Formaldehyde Epiclorohydrine Bisphenol A melamanine isocynates etc				5
8	Storage Handling Hazards of monomers				3
9	Evaluation of raw materials and reactants for synthesis & manufacturing of resins and polymers.				5
	<b>Total</b>				<b>30</b>
<b>List of Text Books/ Reference Books</b>					
1	Raw Materials for Industrial Polymers by H Ulrich, Hanser Publication 1989.				
2	Principles of Polymer Science, by Bahadur and Sastry, Narosa Publishing House 2002.				
3	Polymer Science by Gowariker, Johan wiley and Sons 1986.				
4	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc 1965.				
5	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc 1988.				
6	Petrochemicals: The Rise of an Industry by Peter H. Spitz, Johan Wiley and sons 1988.				
7	Polymer Chemistry by Malcolm P. Stevens, Oxford University Press, Inc, 1990				
<b>Course Outcomes (Students will be able to.....)</b>					
CO2	Interpret the physical and chemical properties of raw materials (K3)				
CO3	Analyze the manufacturing routes and impurities in monomers and raw materials (K4)				
CO4	Discuss about the environmental concerns handling Safety and Hazards of Monomers (K2)				
CO5	Propose plan about evaluation of raw materials and reactants for synthesis & manufacturing of resins and polymers. (K5)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+ Psy	K3	K4
CO1	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
CO2	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
CO3	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO4	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
CO5	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

<b>EEM</b>	<b>Course Code:</b>	<b>Course Title:</b> <b>Basic Economics and Finance</b>	<b>Credits = 2</b>		
	<b>HUT1205</b>		<b>L</b>	<b>T</b>	<b>P</b>
	<b>Semester: III</b>		<b>Total Contact Hours: 30</b>	<b>2</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
<b>Course Outcomes (students will be able to.....)</b>					
1	Students will be able to know and apply accounting and finance theory.				
2	Students will be able to understand the mechanics of preparation of financial statements, their analysis and interpretation				
3	Students will be able to explain basic economic terms, concepts, and theories				
4	Students will be able to identify key macroeconomic indicators				
<b>List of Prerequisite Courses</b>					
	<b>MATHS-1 AND MATHS -2 OF FIRST YEAR COURSEWORK</b>				
<b>List of Courses where this course will be prerequisite</b>					
	<b>PROJECT ECONOMICS</b>				
	<b>FUNDAMENTALS OF MARKETING MANAGEMENT AND MARKET RESEARCH</b>				
<b>Description of relevance of this course in the BACHELOR'S Program</b>					
	<b>Course Contents (Topics and subtopics)</b>				<b>Reqd. hours</b>
1	INTRODUCTION Explaining the Economy The Supply and Demand Model Using the Supply and Demand Model				3
2	THE COMPETITIVE EQUILIBRIUM MODEL Deriving Demand Deriving Supply Market Equilibrium and Efficiency				5
3	DEVIATIONS FROM COMPETITION Monopoly and Market Power Between Monopoly and Competition Antitrust Policy and Regulation				5
4	MACRO FACTS AND MEASURES				5

	Getting Started with Macroeconomic Ideas Measuring Production, Income and Spending of Nations	
5	ACCOUNTING TRANSACTIONS  Journal entries Debit credit rules Compound journal entry Journal and ledger Rules of posting entries Trial balance	5
6	CAPITAL AND REVENUE  Income and expenditure Expired costs and income Final accounts Manufacturing accounts Trading accounts Profit and Loss account Suspense account Balance sheet	5
7	CONCEPT OF DEPRECIATION	2
List of Textbooks		
	Finance and Accounting for Nonfinancial Managers: All the Basics You Need to Know -William G. Droms and Jay O. Wright  Microeconomics: Basic Principles and Applications- A A Temu, D W Ndyetabula, et al <b>PRINCIPLES OF ECONOMICS(12e)- E. Case Karl, C. Fair Ray, et al</b>	
List of Additional Reading Material / Reference Books		
	Basic Finance for Nonfinancial Managers: A Guide to Finance and Accounting Principles for Nonfinancial Managers- Kendrick Fernandez  Microeconomic Theory: Basic Principles and Extensions- Walter Nicholson and Christopher Snyder  Macroeconomics(10e) Part of: Pearson Series in Economics (23 books) - by Froyen	

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)														
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
	K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4

<b>CO1</b>	<b>K3</b>	3	3	2	2	2	3	3	3	3	3	3	2	3	3
<b>CO2</b>	<b>K3</b>	3	2	2	2	2	3	3	3	1	2	3	2	3	1
<b>CO3</b>	<b>K3</b>	3	3	2	2	1	1	3	3	3	3	3	2	2	3
<b>CO4</b>	<b>K3</b>	3	3	2	2	2	3	0	2	3	3	3	2	3	1
<b>CO5</b>	<b>K3</b>	3	2	2	0	2	3	3	3	1	3	0	2	3	3
<b>Course</b>	<b>K3</b>	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

PCC	<b>Course Code:</b> PSP1301	<b>Course Title:</b> Pr 1- Raw materials Analysis for Resins and Polymers	<b>Credits = 2</b>		
	<b>Semester: III</b>	<b>Total contact hours: 60 hrs</b>	<b>L</b>	<b>T</b>	<b>P</b>
			-	-	4
<b>List of Prerequisite Courses</b>					
Physical Chemistry I, Physical Chemistry II, Analytical Chemistry, Applied Mathematics- I					
<b>List of Courses where this course will be prerequisite</b>					
Technology of Thermoplastic Polymers (PST1504), Technology of Thermoset Polymers (PST1506), Synthesis & Characterization of Resins & Polymers Lab (PSP1503), Analysis and characterization of Resins and polymers Lab (PSP1504)					
<b>Description of relevance of this course in the B. Tech (Coatings)</b>					
To train the students with respect to various raw materials used in resin synthesis and characteristics of the same, various test methods for determining the purity of the RMs for application in polymer & resin synthesis					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
	1) To Check the colour of oil & resins. 2) To Check the colour of oils & resins on heating. 3) To check the viscosity of oils & resins solution using Ford Cup or Brookfield viscometer. 4) To check the melting range of given resin by capillary tube method. 5) To find the acid value of given sample. 6) To find Aniline point of given solvent. 7) To find the distillation large of given solvent. 8) To find the evaporation rate of given solvent. 9) To find flash point of given solvent. 10) To find moisture content of solvent (qualitative analysis) 11) To find specific gravity of solvent by pycnometer. 12) To find the moisture content of pigment. 13) To find the water soluble matter of pigment. 14) To check the Acidly & Alkalinity of pigment. 15) To check bleeding of pigment. 16) To find oil absorption value of pigment. 17) To find minimum surfactant demand by Daniel flow-point method 18) Analysis and Determination of purity of Phenols and substituted phenols by Bromination Formaldehyde Phthalic Anhydride Hexamine Epichlorohydrine Melamine etc. 19) Analysis of Water Glycerine Calcium Chloride Sodium / Potassium dichromate Hydrogen peroxide etc.				<b>1x4hr/week</b>
<b>List of Text Books/ Reference Books</b>					
1	Testing of Paints by S.Patil, Current Awareness Service Publisher, 1993				
2	Vogel's Qualitative Inorganic Analysis (7th Edition) By Svehla Prentice Hall; 7 edition (March 7, 1996)				
3	Quantitative organic analysis via functional groups. Second Edition. SIDNEY SIGGIA. Wiley, New York,				
4	Quantitative organic analysis via functional groups. Second Edition. SIDNEY SIGGIA. Wiley, New York, 1954 publication Code No. PCN, Philadelphia, Thirteenth edition, 1972				
5	Qualitative Organic Analysis-Author: Arthur I. Vogel Publisher:Longman Group Ltd. London Sixth Edition, 1970				
<b>Course Outcomes (students will be able to.....)</b>					
CO1	Examine raw material purity and its significance in polymer synthesis (K4)				
CO2	Calculate the physical parameters of raw materials including viscosity, specific gravity, melting point etc. (K3)				
CO3	Analysis of functional group and to determine purity of functional raw materials (K3)				
CO4	Manage to separate various solvents from their mixture (K5)				

CO5	Design experiment to determine purity of pigments with respect to their physical parameters (K5)
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Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+ Psy	K3	K4
CO1	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO2	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
CO3	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
CO4	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO5	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

PCC	Course Code: PSP1302	Course Title: Pr 2- Synthesis and Characterization of Resins and Polymers I Common	Credits = 2		
	Semester: III	Total contact hours: 60 hrs	L	T	P
			0	0	4
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset, Technology of Thermoplastics, Raw material Analysis of resins and polymers, Analysis and characterization of resins and polymers lab.					
<b>List of Courses where this course will be prerequisite</b>					
Compounding and Polymer Processing, Project I, Environment Health and Safety of Polymers and Coating , Evaluation and testing of Polymers and Coatings, Structure Property relationship. Paint Processing II, Project I, Project II					
<b>Description of relevance of this course in the B. Tech. Program</b>					
To give understanding of laboratory scale synthesis processes, properties and applications of various types of thermoplastic and thermoset polymers. Knowledge of subject will help student to carry out Production, Research and development in the areas of polymer Synthesis, Polymer nanocomposites ,coating formulation development, Fiber reinforced composites, Polymer processing etc.To make them aware of Environmental concerns of Polymer Synthesis. Handling Hazards of raw materials monomers, Work ethics in group, Ability design and conduct experiments, Ability to analyze and interpret data, process parameters . To understand and do calculations observations formulations involved team work and understanding practical problems related to the experiment					
Sr. No	Course Contents				Reqd. hours
1	Bulk, Solution and Suspension polymerization of monomers like styrene, MMA etc. and to analyses % solids, %yield, melting range etc				<b>1x4hr/Week</b>
2	Emulsion polymerization of monomers like vinyl acetate, styrene etc and to analyse polymer content, %solids etc.				
3	Aqueous polymerization of monomers like AA, Acrylamide etc. and analyse %solids, %yield, melting range etc.				
4	Synthesis of phenolic resin such as novalac, resol and to analyse free formaline, free phenol content, %solids, curing charecterestics etc.				
5	Synthesis of epoxy resin and to find epoxy value, epoxy equivalent yield etc.				
6	Synthesis of Unsaturated polyesters and to analyse Acid value, yield etc.				
7	Synthesis of copolymer of styrene and acrylate and to analyse yield melting range				
8	Polymer nanocomposites via insitu polymerization				
9	To study kinetics of free radical polymerization				
10	To synthesis superabsorbant, hydrogels and its analysis				
11	Plastisol core and shell polymers and its analysis				
12	Synthesis of amino resins like Melamine formaldehyde and urea formaldehyde resin And its analysis and application.				
<b>List of Text Books/ Reference Books</b>					
1	Polymer Chemistry: A Practical Approach (The Practical Approach in Chemistry Series)1st Edition <u>Fred J. Davis</u> Oxford University Press 2004				
2	A Practical Course in Polymer ChemistryS. H. Pinner, Borough Polytechnic,London, Pergamon Press,he., New York, 1961				
3	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994				
4	Polymer Science by Gowarikar,John Wiley and Sons 1986				
5	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc 1965.				
6	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc 1988				
7	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994				
8	Principles of polymerization, G.Odian, Wiley – Interscience (1981)				
9	PVC Technology 4th edition by W.V.Titow Elsevier Applied Science Publishers, London, 1984				
10	Phenolic Resins chemistry, Applications, Standardization, Safety and Ecology by L.Knop,Springer-Verlag Berlin Heidelberg 2000				



11	Chemistry and Technology of Epoxy Resins by Eliss Brayn ,Springer Nethelands,1993
12	Plastics Materials, 7th Edition by John Brydson, Elsevier 1999
13	Experimental Plastics A practical course for students by C.A.Redfran, Interscience Publisher Inc.NY 1971
14	Testing of Paints by S.Patil, Current Awareness Service Publisher, 1993
<b>Course Outcomes (students will be able to.....)</b>	
CO1	Perform laboratory scale experiment for synthesis of polymers like PS PMMA polyacrylamide Epoxy Polyesters nanocomposites .etc (K5)
CO2	Design and conduct experiments for synthesis of Resins and polymers and understand the practical problems related to the experiment (K5)
CO3	Analyze and characterize polymers by finding yield melting point epoxy value acid value % solid etc within realistic constraints of the experiment (K4)
CO4	Interpret and compare data, process parameters within realistic constraints of the experiment (K4)
CO5	Collect various experimental results, manage to work effectively in team work and understanding of professional and ethical responsibility (K5)

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+ Psy	K3	K4
CO1	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO2	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO3	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO5	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

# Semester-IV

PCC	<b>Course Code:</b> CET1105	<b>Course Title:</b> Transport Phenomena	<b>Credits = 4</b>		
	<b>Semester: IV</b>	<b>Total Contact Hours: 60</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>3</b>	<b>1</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
XII <sup>th</sup> Standard Physics and Mathematics					
<b>List of Courses where this course will be prerequisite</b>					
This is a basic course required in special subjects that deal with flow offluids, heat and mass transfer, etc.					
<b>Description of relevance of this course in the B. Tech. Program</b>					
This basic course introduces concepts of momentum, heat and mass transfer to students. Various other concepts such as pressure, momentum, energy are introduced as well. Laws related to conservation of momentum, energy, mass are taught. Applications of these laws to various engineering and technological situations and process equipments are explained with the help of several problems.					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	Fluid Statics and applications to engineering importance.				4
2	Applications of Bernoulli's Equation, Pressure drop in pipes and Fittings, meters, and fluid moving machinery such as pumps.				10
3	Particle Dynamics, Flow through Fixed and Fluidised Beds				4
4	Equations of Continuity and Motion in laminar flows and its applications for simple Couette flow and Poiseuille flow applications				6
5	Heat conduction. Convective heat transfer and concept of heat transfer coefficient.				4
6	Design and constructional aspects of exchangers: Types of flows: Concurrent, counter-current and cross flows, log mean temperature difference, double pipe and Shell and tube heat exchangers. Introduction to other heat exchangers like, PHE, finned tube heat exchangers, graphite block, etc.				10
7	Heat transfer aspects in agitated tanks, condensers, reboilers and evaporators.				6
8	Fundamentals of mass transfer: Molecular diffusion in fluids, concept of mass transfer coefficients, and interface mass transfer.				4
9	Theories of Mass transfer, Analogies for heat and mass transfer, Empirical correlations				4
10	Mass transfer applications in simple 1-D situations.				8
			<b>Total</b>		<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Transport Phenomena, Bird R.B., Stewart W.E., Lightfoot E.N.				
2	Fluid Mechanics, Kundu Pijush K.				
3	Fluid Mechanics, F. W. White				
4	Unit Operations of Chemical Engineering, McCabe, Smith				
<b>Course Outcomes (students will be able to.....)</b>					
CO1	Students should be able to calculate friction factor, pressure drop, power requirements of single phase flow in a circular pipe				
CO2	Students will be able to calculate flow and power required for pumps				
CO3	Students should be able to calculate heat transfer coefficients and do basic sizing of double pipe and shell and tube heat exchangers				
CO4	Students should be able to calculate mass transfer coefficients and estimate mass transfer rates in simple situations				

<b>Mapping of Course Outcomes (Cos) with Programme Outcomes (Pos)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
<b>CO2</b>	K3	3	3	1	2	1	3	1	3	3	3	1	2	3	3
<b>CO3</b>	K3	3	1	2	2	2	2	3	2	3	3	3	2	2	3
<b>CO4</b>	K3	3	3	2	0	2	3	3	3	3	2	3	0	3	3

<b>Course</b>	<b>K3</b>	3	3	2	2	2	3	3	3	3	3	3	2	3	3
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3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	Course Code: PST1401	Course Title: Spl 5 -Technology of Thermoplastic Polymers	Credits = 3		
	Semester: IV	Total Contact Hours: 45	L	T	P
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Raw material Analysis of resins and polymers, High Polymer Chemistry					
<b>List of Courses where this course will be Prerequisite</b>					
Compounding and Polymer Processing, Environment Health and Safety of Polymers and Coating, Evolution and testing of Polymers and Coatings, Technology of Plastic Packaging .					
<b>Description of relevance of this course in the B. Tech. (Surface coating Tech.) Program</b>					
To give understanding of industrial manufacturing processes, properties and applications, processing of various types of thermoplastic polymers. Knowledge of subject will help student to carry out research and development in the areas of polymer blends polymer nanocomposites, coating formulation development, Fiber reinforces composites, Polymer processing, Rheology of polymers etc. To make aware of Environmental concerns of Polymer products, Recycling of Polymers, industrially produced different grades trade names of polymers.					
Sr. No.	Course Contents (Topics and subtopics)				Required Hours
1	Industrial Manufacturing processes, properties and applications, processing environmental concerns of various types of polymers polyolefins like LDPE HDPE etc.				5
2	Polypropylene and copolymer of PP Plastomers				3
3	Copolymer of polyolefines like EVA LLDPE EAA etc.				2
4	Polystyrene, HIPS, SAN				2
5	ABS, important copolymers of styrene maleic anhydride and styrene acrylics copolymers, toughening mechanism of impact modified plastics.				5
6	Saturated Polyesters such as PET, PBT, PTT				3
7	Polycarbonates, Polyacetals				2
8	Polymamides- Nylon 6, Nylon 6,6, Nylon 11 etc., aromatic polyamide such as Kevlar				5
9	Acrylic polymers & copolymers, Polyacrylamide, PMMA, Polyacrylonitrile etc.				5
10	Polyvinyl chloride & its copolymers Compounding of PVC				3
11	Cellulose esters and ethers such as Ethyl cellulose, CMC, CN, cellulose acetates etc.				5
12	Thermoplastic PU, Poly vinyl acetate, Polyvinyl alcohol etc.				5
<b>Total</b>					<b>45</b>
<b>List of Text Books/ Reference Books</b>					
1	Plastics Materials, 7th Edition by John Brydson, Elsevier 1999.				
2	Text book of polymer Science by Bill Meyer, John Wiley and Sons 1984				
3	Principles of Polymer Science, by Bahadur and Sastry, Narosa Publishing House 2002.				
4	Polymer Science by Gowarikar, John Wiley and Sons 1986.				
5	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc.1965.				
6	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc1988.				
7	Handbook of Thermoplastics, Second Edition Olagoke Olabisiby CRC Press2015				
8	Thermoplastic Materials by Ibeh, Christopher C, Taylor Francis Inc 2013				
9	Introduction to Polymer Science and Technology by H. S. Kaufman and J. J. Falcetta, Wiley Inter science Publication, 1977				
10	Handbook of Polyethylene, A. J. Peacock, Marcel Dakker Inc,2000				
11	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994.				
12	Engineering Thermoplastics Polycarbonates Polyacetals Cellulose Esters, L. Bottenbruch, Hanser Publishers, 1996.				
13	Polymer and Resins; Their Chemistry and Chemical Engg, Brage Golding, D. Van Nostrand Company Inc, 1959.				

14	Structures of Cellulose, Atlla, American Chemical society, 2003.
<b>Course Outcomes (Students will be able to.....)</b>	
CO1	Inspect the industrial manufacturing process, compare the advantages disadvantages of such processes, define the process parameters of the thermoplastics polymers and discuss the environmental concerns of their products (K4)
CO2	Analyze properties like physical mechanical thermal rheological etc (K4)
CO3	Describe the basic processing methods related to of the thermoplastics polymers. (K2)

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+ Psy	K3	K4
CO1	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO2	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO3	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
Course	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	Course Code: PST1505	Course Title: Spl 6- Technology of Thermoset Polymers	Credits = 3		
	Semester: IV	Total Contact Hours: 45	L	T	P
<b>List of Prerequisite Courses</b>					
Polymer science and Technology (PST1301), Polymer chemistry and Technology (PST1303), Raw material Analysis of resins and polymers (PSP1301), High Polymer Chemistry (PST 1404)					
<b>List of Courses where this course will be Prerequisite</b>					
Processing of Paint lab -I (SCP 1606), Processing of Paint lab- II (SCP 1609) , Project I (PSP1713), Project II (PSP 1811) Environment Health and Safety of Polymers and Coating(PST1712), Evolution and testing of Polymers and Coatings(PST1711), Technology of Plastic Packaging(PET1712).					
<b>Description of relevance of this course in the B. Tech. (Surface coating Tech.) Program</b>					
To give understanding of alkyd resins, types, synthesis, properties and modification of alkyd resins. Understanding of polyester resins, raw materials used and various curing systems. Basics of Phenolics, polyurethane, silicone and acrylics resins. Their synthesis, modification, processing, chemistry and applications.					
Sr. No.	Course Contents (Topics and subtopics)				Required Hours
1	Alkyd resins Basic components like polyfunctional alcohols, poly- basic acids, vegetable oils/fatty acids. Different types of drying oils: drying, semi-drying and non-drying with examples. Influence of all these components in the synthesis and properties of the final alkyds obtained. Modification of alkyds: modifications with rosin, maleic anhydride, acrylics, vinyls, imides, etc.				5
2	Polyesters Resins – unsaturated polyesters resins: Raw material: poly-basic acids, polyfunctional glycols. Curing of resins through unsaturation of the resin/polymer backbone. Curing systems, catalysts and accelerators. Molding compositions, fibre and film forming compositions				5
3	Phenolics. Basic Components of the polymer. Different kinds of phenols to aldehyde on the nature and the property of the polymer. Theory of resinification and effect of pH on the reaction mechanism and the reaction product. Curing of Phenolics.				5
4	Modification of Phenolics such as oil soluble and oil reactive. Phenolic moulding compounds ingredients, compounding and applications				3
5	Polyurethanes – Thermoplastic and Thermoset: Basic components diisocyanates and diols, different diisocyanates and diols used Reactions of isocyanates with various other functional groups synthesis of polymers polyurethane foams, polyester and polyether foams.				5
6	Processes like one-shot process, Polyether pre-polymers, Quasi- pre-polymer polyether foams, etc. Flexible foams Polyurethanesin Coatings Polyisocyanates IPN using polyurethanes-acrylicblends.				5
7	Silicones Thermoplastic and Thermoset; Preparation of intermediates, Grignard's method, directs method, olefin addition method, sodium condensation method, rearrangement of organochlorosilanes.				2
8	Nature and effect of Si-H, Si-O, Si-Si, and Si-C bond. Silicone fluids, resins, elastomers.				3
9	Compounding, Processing and applications of Silicone resins. Modified silicone resins.				5
10	Thermosetting acrylics: Synthesis of acrylic polymers and co- polymers, different techniques. Structure property relationship application of thermosetting acrylics, like anaerobicadhesives, laminating resins, etc				5
11	Miscellaneous thermosetting polymers.				2
<b>Total</b>					<b>45</b>
<b>List of Text Books/ Reference Books</b>					
1	Text book of Polymer Science by Bill Meyer, John Wiley Ans Sons 1984.				

2	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc 1965.
3	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc 1988.
4	Polymer Chemistry by Malcolm P. Stevens, Oxford University Press, Inc, 1990.
5	Introduction to Polymer Science and Technology by H. S. Kaufman and J. J. Falcetta, Wiley – Interscience Publication, 1977
6	Handbook of Thermoplastics, O. Olabisi, Marcel Dekker, 1997.
7	Resins for Surface Coatings, Polyurethanes Polyamides Phenolplasts Aminoplasts Maleic Resins (Waterborne & Solvent Based Surface Coatings Resins & Applications) (Volume III) Volume III Edition
8	Resins for Surface Coatings, Volume 1 2nd Edition, Resins for Surface Coatings: Acrylics and Epoxies 2nd Edition by H. Coyard (Author), P. Deligny (Author), N. Tuck (Author), P. K. T. Oldring (Editor)
9	Resins for surface coating- Oldring series
10	Basics of Paint Technology Part I, V. C. Malshe.
11	Organic coatings science and technology, third edition, Zeno Wicks, 2007
12	Plastics Materials J. A. Brydson, Butterworth Scientific, 1990.
13	Polymer chemistry, Seymour and Carraher, Marcel Dekker, 2003.
14	Polymer and Resins; Their Chemistry and Chemical Engg, Brage Golding, D. Van Nostrand Company Inc, 1959.
15	Structures of Cellulose, Atlla, American Chemical society, 2003.
16	Polymer Technology by Miles and Briston Falcetta, Wiley – Interscience Publication, 1977
17	Polymer Technology by Miles and Briston
<b>Course Outcomes (Students will be able to.....)</b>	
CO1	To study the basics of alkyd resins and differentiate between the various types of alkyds. To understand the chemistry of alkyd resins and provide inputs for modification of alkyds. (K4)
CO2	To study the chemistry of polyurethanes. Compare the various raw materials and their reactivity for polyurethanes and provide inputs for modification (K4)
CO3	Interpret the importance of silicones resins. (K3)
CO4	Identify the role of various types of phenolic resin in polymer and paint industry (K2)
CO5	Distinguish between various chemistries of acrylic and polyester (K4)

<b>Mapping of Course Outcomes (COs) with Program Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+ Psy	K3	K4
CO1	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO2	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO3	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
CO4	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
CO5	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
Course	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain



	<b>Course Code:</b> HUT1206	<b>Course Title: Environmental Sciences</b>	<b>Credits = 2</b>		
	<b>Semester: III</b>		<b>Total contact hours: 30</b>	<b>L</b>	<b>T</b>
			<b>2</b>	<b>0</b>	<b>0</b>
<b>Course Outcomes (students will be able to.....)</b>					
1	Describe the methods of industrial effluent treatment				
2	apply the learning for selection and implementation of appropriate waste management technique for sustainable development				
<b>List of Prerequisite Courses</b>					
	<b>Course Contents (Topics and subtopics)</b>				
	<b>Reqd. hours</b>				
1	(a) Concept of circular economy, EHS management (b) Environment management systems in the chemical industry (c) Legal provisions for environmental management: EP Act 1986; Air Act, 1981; Water Act, 1974; Hazardous waste management Rules, 2019				6
2	Importance of ecology, effluent treatment and discharging norms for treated water				6
3	SPCB consent parameters, monitoring and analysis				4
4	External monitoring of ambient air, noise, stacks, etc				4
5	Air pollutants, sources and effects on human health and environment, monitoring and analysis				6
6	Life cycle analysis, environmental impact assessment				4
<b>List of Text Books</b>					
1	Introduction to Environmental Engineering and Science by Gilbert M Masters and Wendell P Ela				
2	Environmental Pollution Control Engineering, C. S. Rao				
3	Principles of Instrumental Analysis by D. A. Skoog, F. James Holler and S. R. Crouch, Cengage Learning, 2007				
<b>List of Additional Reading Material / Reference Books</b>					

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+S	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	2	3	3	3	2	3	3
<b>CO2</b>	K3	3	3	2	2	0	3	3	3	3	3	3	1	3	3
<b>CO3</b>	K3	3	3	0	2	2	3	1	3	3	1	3	2	2	3
<b>CO4</b>	K3	3	1	2	2	2	3	3	3	3	3	0	2	3	3
<b>CO5</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
<b>Course</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

	<b>Course Code: CET1805</b>	<b>Course Title: Chemical Process Economics</b>	<b>Credits=2</b>		
			<b>L</b>	<b>T</b>	<b>P</b>
	<b>Semester: IV</b>	<b>Total contact hours: 30</b>	<b>2</b>	<b>0</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
Material and Energy Balance Calculations, Equip Design and Drawing I, Energy Engineering, Ind Eng Chem.					
<b>List of Courses where this course will be prerequisite</b>					
Home Paper I and II					
<b>Description of relevance of this course in the B Tech. Program</b>					
This course is required for the future professional career					
	<b>Course Contents (Topics and subtopics)</b>				<b>Reqd.</b>
1	Estimation of Plant and Machinery cost, Capacity Index, Cost Indices				8
2	Relationship between price of a product and project cost and cost of production, EV Analysis. Elements of cost of production, monitoring of the same in a plant, Meaning of Administrative expenses, sales expenses etc. Introduction to various components of project cost and their estimation. Project financing, debt: equity ratio, promoters, contributors, shareholders				8
4	Project financing, debt: equity ratio, promoters, contributors, shareholders contribution, source of finance, time value of money. Concept of interest, time value of money, selection of various alternative equipment or system based on this concept. Indian norms, EMI calculations. Depreciation concept, Indian norms and their utility in estimate of working results of project. Working capital concept and its relevance to project.				8
5	Estimate of working results of proposed project. Capacity utilization, Gross profit, operating profit, profit before tax, Corporate tax, dividend, Net cash accruals. Project evaluation: Cumulative cash flow analysis Break-Even analysis, incremental analysis, various ratios analysis, Discounted cash flow analysis				6
<b>List of Text Books/ Reference Books</b>					
1	Chemical Project Economics, Mahajani V.V. and Mokashi SM.				
2	Plant Design and Economics for Chemical Engineers, Peters M.S., Timmerhaus K.D.				
3	Process Plant and Equipment Cost Estimation, Kharbanda O.P.				
<b>Course Outcomes (students will be able to.....)</b>					
1	Calculate working capital requirement for a given project				
2	Calculate cost of equipment used in a plant total project cost				
3	Calculate cashflow from a given project				
4	Select a site for the project from given alternatives				
5	List out various mile stones related to project concept to commissioning				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
<b>CO2</b>	K3	3	3	2	2	2	3	3	3	3	3	2	2	2	3
<b>CO3</b>	K3	3	3	1	0	2	3	1	3	3	3	3	2	3	2
<b>CO4</b>	K4	3	3	2	3	2	2	3	3	3	3	3	2	3	3
<b>CO5</b>	K2	3	2	1	2	1	3	3	3	3	0	3	1	3	2
<b>Course</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain



VSEC	<b>Course Code:</b> PSP1401	<b>Course Title:</b> <b>Pr 3- Synthesis and Characterization of Resins and Polymers Lab-II</b> <b>Common</b>	<b>Credits = 2</b>		
	<b>Semester: IV</b>	<b>Total contact hours: 60 hrs</b>	<b>L</b>	<b>T</b>	<b>P</b>
<b>0</b>					
<b>0</b>					
<b>4</b>					
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset, Technology of Thermoplastic, Raw material Analysis of resins and polymers, Analysis and characterization of resins and polymers lab					
<b>List of Courses where this course will be prerequisite</b>					
Compounding and Polymer Processing ,Project I, Environment Health and Safety of Polymers and Coating, Evaluation and testing of Polymers and Coatings, Structure Property relationship(PST1609). Paint Processing II, Project , Project II					
<b>Description of relevance of this course in the B. Tech. Program</b>					
To give understanding of laboratory scale synthesis processes, properties and applications of various types of thermoplastic and thermoset polymers. Knowledge of subject will help student to carry out Production, Research and development in the areas of polymer Synthesis, Polymer nanocomposites ,coating formulation development, Fiber reinforced composites, Polymer processing etc.To make them aware of Environmental concerns of Polymer Synthesis. Handling Hazards of raw materials monomers, Work ethics in group, Ability design and conduct experiments, Ability to analyze and interpret data, process parameters . To understand and do calculations observations formulations involved team work and understanding practical problems related to the experiment					
	<b>Course Contents</b>				<b>Reqd. hours</b>
1	Bulk, Solution and Suspension polymerization of monomers like styrene, MMA etc. and to analyses % solids, % yield, melting range etc				<b>1x4hr/Week</b>
2	Emulsion polymerization of monomers like vinyl acetate, styrene etc and to analyse polymer content, %solids etc.				
3	Aqueous polymerization of monomers like AA, Acrylamide etc. and analyse %solids, % yield, melting range etc.				
4	Synthesis of phenolic resin such as novalac, resol and to analyse free formaline, free phenol content, %solids, curing charecterestics etc.				
5	Synthesis of epoxy resin and to find epoxy value, epoxy equivalent yield etc.				
6	Synthesis of Unsaturated polyesters and to analyse Acid value, yield etc.				
7	Synthesis of copolymer of styrene and acrylate and to analyse yield melting range				
8	Polymer nanocomposites via insitu polymerization				
9	To study kinetics of free radical polymerization				
10	To synthesis superabsorbant, hydrogels and its analysis				
11	Plastisol core and shell polymers and its analysis				
12	Synthesis of amino resins like Melamine formaldehyde and urea formaldehyde resin And its analysis and application.				
<b>List of Text Books/ Reference Books</b>					
1	Polymer Chemistry: A Practical Approach (The Practical Approach in Chemistry Series)1st Edition <u>Fred J. Davis</u> Oxford University Press 2004				
2	A Practical Course in Polymer ChemistryS. H. Pinner, Borough Polytechnic,London, Pergamon Press,he., New York, 1961				
3	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994				
4	Polymer Science by Gowarikar,John Wiley and Sons 1986				
5	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc 1965				
6	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc 1988.				
7	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994				
8	Principles of polymerization, G.Odian, Wiley – Interscience (1981)				
9	PVC Technology 4th edition by W.V.Titow Elsevier Applied Science Publishers, London, 1984				

10	Phenolic Resins chemistry, Applications, Standardization, Safety and Ecology by L.Knop, Springer-Verlag Berlin Heidelberg 2000
11	Chemistry and Technology of Epoxy Resins by Eliss Brayn ,Springer Netherlands,1993
12	Plastics Materials, 7th Edition by John Brydson, Elsevier 1999
13	Experimental Plastics A practical course for students by C.A.Redfran, Interscience Publisher Inc.NY 1971
14	Testing of Paints by S.Patil, Current Awareness Service Publisher, 1993
<b>Course Outcomes (students will be able to....)</b>	
CO1	Perform laboratory scale experiment for synthesis of polymers like PS, PMMA, polyacrylamide, Epoxy, Polyesters, nanocomposites, etc (K5)
CO2	Design and conduct experiments for synthesis of Resins and polymers and understand the practical problems related to the experiment (K5)
CO3	Analyze and characterize polymers by finding yield, melting point, epoxy value, acid value, % solid, etc. within realistic constraints of the experiment (K4)
CO4	Interpret and compare data, process parameters within realistic constraints of the experiment (K4)
CO5	Collect various experimental results, manage to work effectively in team work and understanding of professional and ethical responsibility (K5)

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+ Psy	K3	K4
CO1	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO2	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO3	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO5	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

# Semester-V

PCC	Course Code: CET1806	Course Title: Chemical Reaction Engineering	Credits = 2		
	Semester: V	Total contact hours: 30	L	T	P
			1	1	0

#### List of Prerequisite Courses

Physical Chemistry I and II, Transport Phenomena

#### List of Courses where this course will be prerequisite

Environmental Engineering and Process Safety, Chemical Project Economics

#### Description of relevance of this course in the B.Tech. Program

Chemical Reaction Engineering is concerned with the utilization of chemical reactions on a commercial scale. This course is very relevant but not limited to the following industries: Inorganic chemicals, organic chemicals, petroleum & petrochemicals, Pulp & paper, Pigments & paints, rubber, plastics, synthetic fibres, Foods, Dyes and intermediates, Oils, oleo chemicals, and surfactants, Minerals, clean sing agents, Polymers and textiles, Biochemicals and biotechnology, pharmaceuticals and drugs, Microelectronics, energy from conventional and non-conventional resources, Metals

	Course Contents (Topics and subtopics)	Reqd. hours
1	Kinetics of homogeneous reactions, Interpretation of batch reactor data, Single ideal reactors including design aspects	8
2	Multiple reactions, Temperature, and pressure effects	3
3	Introduction to Non ideal flow, RTD measurements, Models to predict conversions	2
4	Homogeneous and Heterogeneous Catalysis, Kinetics of Solid Catalyzed Reactions. Design of gas – solid catalytic reactors	8
5	Introduction to Multiphase reactors	4
6	Mass transfer with chemical Reactions: Regimes of operation and Model contactors	5
	<b>Total</b>	<b>30</b>

#### List of Textbooks

1 Elements of Chemical Reaction Engineering – H.Scott Fogler

#### List of Additional Reading Material / Reference Books

1 Heterogeneous Reactions, Vol.I and II –L.K. Doraiswamy, M.M.Sharma

#### Course Outcomes (students will be able to.....)

1 Describe and discuss principles of various types of reactors  
 2 Calculate rates of reactions based on given reaction scheme  
 3 Design various components of reactors used in industrial practice  
 4 Compare various reactors and select an appropriate reactor for a given situation  
 5 Describe and discuss principles of various types of reactors

#### Mapping of Course Outcomes (COs) with Programme Outcomes (POs)

		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K3	3	3	2	2	1	3	3	3	3	3	3	2	3	3
CO2	K3	3	2	2	2	2	1	3	0	3	3	2	0	3	3
CO3	K3	3	3	2	1	2	3	3	3	3	3	3	2	3	3
CO4	K4	3	3	2	3	0	2	3	3	1	3	3	1	3	3
Course	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

PCC	Course Code: CET1807	Course Title: Chemical Engineering Operations	Credits = 2		
	Semester: V	Total contact hours:30	L	T	P
			1	1	0
<b>List of Prerequisite Courses</b>					
	Process Calculations, Transport Phenomena				
<b>List of Courses where this course will be prerequisite</b>					
	This is a basic course. It is required in many other courses that involve physical processes				
<b>Description of relevance of this course in the B. Tech. Program</b>					
This is a basic Chem Engg. course. The principles learnt in this course are required in almost all the courses and throughout the professional career of student					
	<b>Course Contents (Topics and subtopics)</b>				<b>Reqd. hours</b>
1	Distillation: Fundamentals of flash, batch and continuous distillation, distillation columns internals, steam and azeotropic distillation				10
2	Liquid-Liquid Extraction: Solvent selection, construction of ternary diagrams, staged calculations, types of extraction equipment.				5
3	Crystallization: Phase diagram (temp/solubility relationship), evaporative and cooling crystallization, introduction to different types of crystallizers				5
4	Filtration: Mechanism of filtration, basic equation, constant volume, constant pressure filtration, rate expressions with cake and filter cloth resistances, compressible and incompressible cakes, introduction to various types of filters				5
5	Drying: Drying mechanism, drying rate curves, estimation of drying time and types of dryers				5
	<b>Total</b>				<b>30</b>
<b>List of Text Books/ Reference Books</b>					
1	Richardson, J.F., Coulson, J.M., Harker, J.H., Backhurst, J.R., 2002. Chemical engineering: Particle technology and separation processes. Butterworth-Heinemann, Woburn, MA.				
2	Seader, J.D., Henley, E.J., 2005. Separation Process Principles, 2 ed. Wiley, Hoboken, N.J.				
3	Svarovsky, L., 2000. Solid-Liquid Separation. Butterworth-Heinemann, Woburn, MA.				
4	McCabe, W., Smith, J., Harriott, P., 2004. Unit Operations of Chemical Engineering, 7 ed. McGraw-Hill Science/Engineering/Math, Boston.				
5	Green, D., Perry, R., 2007. Perry's Chemical Engineers' Handbook, Eighth Edition, 8 ed. McGraw-Hill Professional, Edinburgh.				
6	Dutta, B.K., 2007. Principles of Mass Transfer and Separation Process. Prentice-Hall of India Pvt. Ltd, New Delhi.				
<b>Course Outcomes (students will be able to.....)</b>					
1	Do basic sizing of continuous and batch distillation columns				
2	Analyze filtration data and select systems based on requirements, estimate filtration area for given requirements, understand filter aids and their usage				
3	Describe few industrial crystallization, filtration and drying equipment				
4	Describe the need and importance of other separation processes like adsorption, ion exchange and membrane				
5	Gain a practical perspective of unit operation in chemical industries				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	1	3	3	3	3	3	0	2	3	3
<b>CO2</b>	K4	3	3	2	3	2	3	2	3	3	2	3	2	3	3
<b>CO3</b>	K2	3	2	0	2	1	3	3	2	3	3	3	1	3	2
<b>CO4</b>	K2	3	2	1	2	0	3	3	3	3	1	3	1	2	2
<b>CO5</b>	K3	3	3	2	2	2	1	3	3	1	3	3	2	3	3
<b>Course</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain



PCC	Course Code: SCT1501	Course Title: Spl - 7 Paint Technology I	Credits = 3		
	Semester: V	Total Contact Hours: 45	L	T	P
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymer					
<b>List of Courses where this course will be Prerequisite</b>					
Paint Technology II, Environment Health and Safety of Polymers and Coating, Evaluation and testing of Polymers and Coatin					
<b>Description of relevance of this course in the B. Tech. (Surface coating Tech.) Programme</b>					
To give understanding of industrial manufacturing processes, properties and applications, processing of various types of paints. Knowledge of subject will help student to carry out research and development in the areas of paints and coatings, coating formulation development, setting up a paint industry and plant, basics of research and development, etc. To make aware of Environmental concerns of paints and coatings e.g., release of VOCs and the effect of VOCs on the environment.					
Sr. No.	Course Contents (Topics and subtopics)				Required Hours
1	Colloidal chemistry of coatings, surface chemistry of pigment				4
2	Pigment dispersion and wetting, flushing of pigments, effect of pigment volume concentration on paint properties				4
3	Paint additives (wetting and dispersing agents, rheology modifiers, etc.) and solvents				10
4	Basics of Paint formulations				5
5	Machinery for grinding of pigments and extender				2
6	Paint manufacturing machinery for pigment dispersion (Ball mill, Sand mill, Attritor mills, basket mill, kaddy mills, twin shaft dispenser, alpine mills, horizontal vs. vertical mills, etc.)				8
7	Manufacture of Powder Coatings, dry distempers, cement paints, oil-based distempers and paints, other stiff paints, putties, etc.				4
8	Manufacturing of alkyds, emulsions and hard resins, filtration of resins, paints; forming of hard resins, marking and labeling of packaged products				6
9	Utilities in paint plant (steam, hot oil, cooling water, chilled water, compressed air, etc.)				5
10	Plant layout, Inventory control, use of computers in paint industry, interphasing with R&D.				6
11	Solvent emission, recovery and disposal, environmental, health and safety issues.				6
<b>Total</b>				<b>60</b>	
<b>List of Text Books/ Reference Books</b>					
1	Polymer Chemistry: A Practical Approach (The Practical Approach in Chemistry Series)1st Edition Fred J. Davis Oxford University Press 2004.				
2	Basics of Paint Technology Part I, V. C. Malshe.				
3	Polymer Science by Gowarikar, John Wiley and Sons 1986				
4	Resins for Surface Coatings, Polyurethanes Polyamides PhenolplastsAminoplasts Maleic Resins (Waterborne & Solvent Based Surface Coatings Resins & Applications) (Volume III) Volume III Edition				
5	Resins for Surface Coatings, Volume 1 2nd Edition, Resins for Surface Coatings: Acrylics and Epoxies 2nd Edition by H. Coyard (Author), P. Deligny (Author), N. Tuck (Author), P. K. T. Oldring (Editor)				
6	Basics of Paint Technology Part II, Part 2, V. C. Malshe, Prakash C. Malshe, 2008 - Coatings - 624 pages				
7	Principles of polymerization, G. Odian, Wiley – Interscience (1981)				
8	Outlines of Paint Technology Hardcover – December 1, 2000 by Morgan (Author)				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Analyze various factors affecting the stability of paint (K4)				
CO2	Interpret the importance of additive and their dosage in paints coating formulation. (K3)				
CO3	Design basic criteria for paint recipe (K5)				

CO4	Formulate paint formulation considering various ingredients (K5)
CO5	Prepare and Perform paint processing by handling various machineries and equipment used in laboratory commercial scale. (K5)
CO6	Discuss manufacturing processes for different resins and coating types (K6)
CO7	Discover industrial operations of paint industries including layout, utilities, and environment, health and safety considerations (K3)

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K4	3	3	2	3	2	3	3	1	3	3	3	2	3	3
CO2	K3	3	3	2	2	2	3	3	2	3	3	3	2	3	3
CO3	K5	3	3	3	3	2	3	3	2	3	3	3	3	3	3
CO4	K5	3	3	3	3	2	3	3	2	3	3	3	3	3	3
CO5	K5	3	3	3	3	3	3	3	1	3	3	3	3	3	3
CO6	K6	2	1	1	1	2	2	3	2	1	2	1	3	3	3
CO7	K3	3	2	2	2	1	3	3	2	1	2	2	2	2	3
Course	K6	3	3	3	3	2	3	3	2	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PEC	<b>Course Code:</b> <b>PST1609</b>	<b>Course Title:</b> <b>Spl 8 - Structure property Relationship</b>	<b>Credits = 3</b>		
			<b>L</b>	<b>T</b>	<b>P</b>
	<b>Semester: V</b>	<b>Total contact hours: 45</b>	<b>2</b>	<b>1</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
Polymer Science & Technology (PST1301), Polymer Chemistry & Technology (PST1303), Technology of Thermoplastics (PST1504), Technology of Thermosets (PST1506)					
<b>List of Courses where this course will be prerequisite</b>					
Project I (PSP1714), Project II (PSP1811) Seminar (PSP1712), Speciality Polymers (PET1816)					
<b>Description of relevance of this course in the B. Tech. Program</b>					
To study the General structural features of polymers: Effects of atoms types of bonds, bond dissociation energy and functional groups on properties of polymers. To study the Configuration and conformation and structure properties of polymers and Molecular mass heterogeneity and structure properties. To study the Polymers solutions: thermodynamics of dissolution, factors effecting dissolution and swelling of polymers, phase equilibrium of polymer-solvent systems, polymer solution, Florry-Huggins theory					
<b>Sr. No</b>	<b>Course Contents</b>				<b>Reqd. hours</b>
1	General structural features of polymers: Effect of types of bonds, bond dissociation energy and functional groups on properties of polymers				10
2	Configuration and conformation and structure properties of polymers				5
3	Molecular mass heterogeneity and structure properties				5
4	Polymers solutions: thermodynamics of dissolution, factors effecting dissolution and swelling of polymers, phase equilibrium of polymer-solvent systems, polymer solution, Florry-Huggins theory				5
5	Polymer Chain flexibility: concept of flexibility, various factors deciding flexibility of polymers with case studies. properties of polymers affected by flexibility				5
6	Intermolecular orders: Amorphous, crystalline and oriented forms of polymers, crystallinity in polymers, factors affecting crystallinity, properties affected by crystallinity of polymers				5
7	Thermal properties of polymers: fire retardant polymers, factors affecting glass transition temperature, heat stability etc. with case studies				5
8	Degradation and stabilization: Various stresses acting on polymers and their influence, method of improving the stability of polymers with case study				5
<b>List of Text Books/ Reference Books</b>					
1	Polymer Structure, Properties and application, R.D. Deanin, American Chemical Society, 1974.				
2	Relating Materials, Properties to Structure; Handbook and Software for Polymer calculations and Materials Properties, D. J. david and Ashok Mishra, Technical Publishing Componey, Inc, 1999.				
3	Properties of Polymer; Correlations with Chemical Structures and their numerical Estimation and Predication from Additive Group Contribution van Krevelen, Elsevier Publication Company, 1990.				
4	Relating Materials Properties to structure, D. J. David, Technical Publishing Company Inc, 1999.				
5	Polymer Chemistry, C. E. Carrshar, Marcel Dakker Inc, 2003.				
6	Physical chemistry of Polymers, A. Tager, Mir Publishers, 1978.				
7	Polymer Association Structures M. A. EL-Nokally, American Chemical Society, 1989.				
8	Polymer Solutions; Introduction to Physical Properties, Teraoka, Iwao, John Wiley and Sons. Inc, 2002.				
9	Polymer Chemistry; An Introduction, M. P. Stevens, Oxford University Press, 1990.				
<b>Course Outcomes (students will be able to.....)</b>					
CO1	Explain the general structural features of polymers (K2)				
CO2	Describe the concept of Configuration and conformation and structure properties of polymers and Molecular mass heterogeneity and structure properties (K2)				
CO3	Discuss the thermodynamics characteristics and identify factors affecting dissolution, polymer chain flexibility and thermal properties of polymers (K2)				
CO4	Interpret about the intermolecular orders and the crystallinity properties. (K3)				
CO5	Apply knowledge to understand the degradation/stabilization of polymers and to analyses the respective case studies (K4)				
CO6	Describe the various thermal properties and factors affecting these properties				

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)														
		PO1	PO2	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+Psy	K3	K4
CO 1	K2	3	2	2	1	3	3	3	3	3	3	1	3	2
CO 2	K2	3	2	2	1	3	3	3	3	3	3	1	3	2
CO 3	K2	3	2	2	1	3	3	3	3	3	3	1	3	2
CO 4	K3	3	3	2	2	3	3	3	3	3	3	2	3	3
CO 5	K4	3	3	3	2	3	3	3	3	3	3	2	3	3
CO 6	K2	3	3	3	2	3	3	3	3	3	2	2	3	3
Course	K4	3	3	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
 K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	<b>Course Code:</b> PST1501	<b>Course Title:</b> <b>Honour Course I - High Polymer Chemistry</b>	<b>Credits = 4</b>		
	<b>Semester: V</b>	<b>Total contact hours: 60</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>3</b>	<b>1</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
Polymer chemistry and Technology, Raw material Analysis of resins and polymer					
<b>List of Courses where this course will be prerequisite</b>					
Compounding and Polymer Processing , Project I and Project II, Environment Health and Safety of Polymers and Coating , Evaluation and testing of Polymers and Coatings,Technology of Plastic Packaging					
<b>Description of relevance of this course in the B. Tech. Program</b>					
To give understanding of mechanisms of free radical and ionic polymerization. To make aware of polymemer synthesis via CRP,ROP GTP etc, They will learn about catalyst used in polymers synthesis like ziegglar-natta, metallocene etc.					
<b>Sr. No.</b>	<b>Course Contents</b>				<b>Reqd. hours</b>
1	Kinetics of free radical polymerization along with different examples & its efficiency, effect on molecular weight/ MWD & effect on tacticity Thermodynamics of free radical polymerization, effect of temp and pressure, enthalpies, entropies, free energies, activation energies of polymerization				5
2	Introduction to anionic polymerization with examples of different systems, Kinetics of anionic polymerization along with different examples & its efficiency, effect on molecular weight/ MWD & effect on tacticity				5
3	Introduction to cationic polymerization with examples of different systems, Kinetics of cationic polymerization along with different examples & its efficiency, effect of counter ion , effect on molecular weight/ MWD & effect on tacticity				5
4	Interfacial polymerization, Melt polycondensation, Solution polycondensation.				5
5	Advanced polymer synthesis and mechanisms , Ring opening metathesis polymerization (ROMP), ring forming polymers,				5
6	Group transfer Polymerization ,Photopolymerization ,Mini-dispersion polymerization,				5
7	Cyclopolymerisation, Oxidative polymerization, Dispersion polymerization ,Metal catalyzed olefin polymerization				5
8	Introduction to Ziegglar natta catalyst its Mechanism with examples of different systems,Effect of catalyst, co- catalyst their ratio, types of metals used their form & pendent groups				5
9	Supported unsupported catalysts, soluble insoluble system, efficiency& rate affecting factors like catalyst/ co catalyst, effect on molecular weight/ MWD & effect on tacticity				5
10	Introduction to Metallocene catalysts with examples of different systems				5
11	Hyperbranched polymers, Dendrimers, Interpenetrating Networks				5
12	Microbial synthesis of polymers, Template polymerization				5
<b>List of Text Books/ Reference Books</b>					
1.	Principles of Polymer Science, Bahadur and Sastry, Narosa Publishing House 2002.				
2.	Polymer Science, Gowarikar, Johan wiley and Sons 1986.				
3.	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc 1965.				
4.	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc 1988.				
5.	Polymer Chemistry, Malcolm P. Stevens, Oxford University Press, Inc, 1990.				
6.	Text book of polymer Science, Bill Meyer, John Wiley ans Sons 1984.				
7.	Principles of Polymer Systems, Rodriguez, Hemisphere Publishing Corpn, 1982.				
8.	Introduction to Polymer Science and Technology, H. S. Kaufman and J. J. Falcetta, Wiley – Interscience Publication, 1977				
9.	Principles of polymerization, G.Odian, Wiley – Interscience (1981)				
<b>Course Outcomes (students will be able to.....)</b>					
CO1	Explain about Kinetics of polymerization & how to control it (K2)				
CO2	Comparison of various monomers and their selection based on achieving required properties (K4)				
CO3	Describe and Design advanced techniques of polymerization (K5)				
CO4	Distinguish about various catalyst used in polymers synthesis like ziegglar-natta, Metallocene etc. (K4)				
CO5	Interpret the importance of advanced polymer synthesis and its commercial implications. (K3)				

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4
CO1	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
CO2	K2	2	3	2	2	2	3	1	3	3	2	3	2	3	3
CO3	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO5	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
 K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	<b>Course Code:</b> SCP1606	<b>Course Title:</b> Pr 4- Processing of Paints Lab-I	<b>Credits = 2</b>		
	<b>Semester: V</b>	<b>Total Contact Hours: 60 hrs</b>	<b>L</b>	<b>T</b>	<b>P</b>
			0	0	4
<b>List of Prerequisite Courses</b>					
Technology of Thermoset Polymers(PST1506), Synthesis & Characterization of Resins & Polymers Lab (PSP1503), Analysis and characterization of Resins and polymers Lab (PSP1504)					
<b>List of Courses where this course will be Prerequisite</b>					
Advanced paint Technology (SCT1815), Analysis and testing of Paints (SCP1808), Corrosion Science and Corrosion Prevention (SCT 1816)					
<b>Description of relevance of this course in the B. Tech. (Surface Coating Tech.) Programme</b>					
Study of synthesis of various resin required as binder for processing of paints. To study the formulation, synthesis and processing of various types of paints.					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	Evaluation of paints as per IS 1012				1x 4hr/week
2	<b>Preparation of</b> <ol style="list-style-type: none"> <li>a. Alkyd resin and its evaluation (Long, Medium and short by different groups)</li> <li>b. Acrylic/vinyl acetate emulsion</li> <li>c. Plastic emulsion paint and evaluation (To include determination of surfactant demand by Daniel flow point method and evaluation of final properties of the prepared paint. Scrub resistance, stain resistance, detergent and soap resistance to be evaluated)</li> <li>d. Polyester polyol from Aliphatic and aromatic dibasic acids, aliphatic diol, triols and its characterizations (A.V. and Hydroxyl value)</li> <li>e. Suspension polymer from MMA and Butyl methacrylate</li> <li>f. Cement paint and application on exterior surface</li> <li>g. Alkyd paint for base coat and top coat at different PVC</li> <li>h. High gloss coating from the polyol and evaluation of the coating properties</li> <li>i. Varnishes for wood finishing</li> </ol>				
3	Flushing of a pigment cake and comparison of the colour properties of the flush with the dry pigment.				
<b>List of Text Books/ Reference Books</b>					
1	Text book of Polymer Science by Bill Meyer, John Wiley Ans Sons 1984.				
2	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc 1988.				
3	Polymer Chemistry by Malcolm P. Stevens, Oxford University Press, Inc, 1990.				
4	Introduction to Polymer Science and Technology by H. S. Kaufman and J. J. Falcetta, Wiley – Interscience Publication, 1977				
5	Handbook of Polyethylene, A. J. Peacock, Marcel Dakker Inc, 2000				
6	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994.				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Perform and analyze various testing of paints (K4)				
CO2	Formulate and Synthesize alkyd, polyester polyol resin. Synthesis of polymers and copolymers by emulsion polymerization, suspension polymerization (K5)				
CO3	Formulate and Synthesize cement paint, alkyd paint, varnishes etc (K5)				
CO4	Test and analyze the synthesise resin and paint to ensure the resin/paint has been successfully formed (K4)				
CO5	Use equipment like flusher and able to compared properties of synthesise pigment with standard pigment (K3)				

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+ Psy	K3	K4
CO1	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO2	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO3	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO5	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
 K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain



PCC	Course Code: PSP1504	Course Title Pr5- Analysis and characterization of Resins and Polymers Lab	Credits = 2		
			L	T	P
	Semester: V	Total Contact Hours: 60 hrs	0	0	4
<b>List of Prerequisite Courses</b>					
Analytical Chemistry Lab, Polymer science and Technology, Polymer chemistry and Technology , Technology of Thermoset, Technology of Thermoplastics, Raw material Analysis of resins and polymers, Analysis and characterization of resins and polymers lab					
<b>List of Courses where this course will be Prerequisite</b>					
Project I, Project II , Research and Development in the area of Polymer Synthesis, analysis and characterization.					
<b>Description of relevance of this course in the B. Tech. (Surface Coating Tech.) Program</b>					
To understand the laboratory scale quality control analysis. Research and Development of Polymer Synthesis. Ability to analyze and interpret data, process parameters. It helps to improve the ability to identify an unknown resin.					
Sr. No.	Course Contents (Topics and subtopics)				Required Hours
1	To determine Acid value, amine value, iodine value, hydroxyl, epoxy, SAP value, ester value of polymers.				<b>1x4hr/Week</b>
2	Refractive Index of resins				
3	Viscosity of resins by various analysis.				
4	K- Value of PVC				
5	Analysis of emulsion polymer				
6	End group analysis of polymers				
7	To determine the melting range and softening range of polymers like Polyolefines, styrenics, engineering polymers.				
8	Determine the chlorine content of the chlorinated polymers				
<b>Total</b>					<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Basics of paint technology I- V.C.Malshe				
2	Testing of paints- Shreekant patil				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	To characterize various resins and polymers (K4)				
CO2	Calculate Acid value, amine value, iodine value, hydroxyl, epoxy, SAP value, ester value of polymers				
CO3	Analyze and characterize polymers and resin for viscosity, refractive index, melting point etc. (K4)				
CO4	Analyze various emulsions and resin (K4)				
CO5	Collect various experimental results, manage to work effectively in team work and understanding of professional and ethical responsibility (K5)				
CO6	To analyze end groups of different resins and polymers				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+ Psy	K3	K4
CO1	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO2	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO3	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO5	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO6	K4	3	3	3	3	2	3	3	2	3	3	3	3	2	3

Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3
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3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
 K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

# Semester-VI

PCC	<b>Course Code:</b> SCT1502	<b>Course Title:</b> SPL-8 Additives and processing of paint	<b>Credits = 4</b>		
	<b>Semester: VI</b>	<b>Total Contact Hours: 45</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>3</b>	<b>1</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
Polymer science and Technology , Polymer chemistry and Technology, Raw material Analysis of resins and polymers					
<b>List of Courses where this course will be Prerequisite</b>					
Compounding and Polymer Processing, Project I , Project II , Environment Health and Safety of Polymers and Coating, Evolution and testing of Polymers and Coatings , Technology of Plastic Packaging . Structure Property relationship, Paint Processing, Paint Technology.					
<b>Description of relevance of this course in the B. Tech. (Surface coating Tech.) Program</b>					
To study various properties of pigments and extenders					
To understand the basics of pigment dispersion.					
To study different inorganic and organic pigments and their different properties.					
To study theory of color formation and effect of auxiliary groups on the shade and hue of the pigment					
To study properties and application of various additives.					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	An overview of paint additives, types of Coating Additive and the Main Technical Trends, need and importance additives.				6
2	Pigment wetting and dispersing additives, Rheological additives, Substrate wetting additives.				6
3	Defoamers and de-reactors, Antioxidants and formulation stabilizers, Surface control additives: flow, leveling, matting agents, Additives to improve adhesion, slip.				10
6	Colorants, Fillers, Thickeners, Surface Active agents, Additives for surface modification.				6
6	Coalescing Agent, Catalytically Active additive.				6
7	Machinery for grinding of pigments and extender, Paint manufacturing machinery for pigment dispersion (Ball mill, Sand mill, Attritor mills, basket mill, kaddy mills, twin shaft dispenser, alpine mills, horizontal vs. vertical mills, etc.)				10
8	Manufacture of Powder Coatings, dry distempers, cement paints, oil-based distempers and paints, other stiff paints, putties, etc.				6
9	Manufacturing of alkyds, emulsions and hard resins, filtration of resins, paints; forming of hard resins, marking and labeling of packaged products, Plant layout, Inventory control, interphasing with R&D, Solvent emission, recovery and disposal, environmental, health and safety issues				10
<b>Total</b>					<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Additives for coating, Johan Bieleman , 2008				
2	Handbook Of Coating Additives, John J. Florio, Daniel J. Miller · 2004				
3	Basics of Paint Technology Part I, V. C. Malshe.				
4	Organic coatings science and technology, third edition, Zeno Wicks, 2007				
5	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc1965				
6	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc1988				
<b>Course Outcomes (students will be able to.....)</b>					
CO1	Identify and discuss about various pigments and additives for a particular application (K2)				
CO2	Explain synthesis techniques for alkyds and different commonly used paints (K2)				
CO3	Plan activities related to the grinding and dispersion methods of pigments and extenders in paint formulations (K5)				
CO4	Classify the various pigments, the dosage and choose various types of additives based on formulation (K2)				

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K1	2	2	3	2	1	2	2	2	3	2	2	3	3	3
CO2	K2	1	1	2	1	2	3	2	1	2	1	2	2	2	3
CO3	K5	2	2	1	1	2	2	1	1	3	2	1	2	3	3
CO4	K2	2	2	2	2	1	2	2	2	2	2	1	3	3	2
Course	K5	2	2	3	2	2	3	2	2	3	2	2	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
 K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	Course Code: SCT1601	Course Title: Spl 10- Paint Technology II	Credits = 3		
	Semester: VI	Total Contact Hours: 45	L	T	P
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymer , Analysis and characterization of resins and polymers lab, Paint Technology I					
<b>List of Courses where this course will be Prerequisite</b>					
Environment Health and Safety of Polymers and Coating, Evaluation and testing of Polymers and Coatings					
<b>Description of relevance of this course in the B. Tech. (Surface coating Tech.) Programme</b>					
To give understanding of industrial manufacturing processes, properties and applications, processing of various types of high-performance paints and coatings. Knowledge of subject will help student to carry out research and development in the areas of high-performance paints and coatings, their formulation development, etc. To make aware of Environmental concerns of high-performance paints and coatings e.g., release of VOCs and the effect of VOCs on the environment.					
Sr. No.	Course Contents (Topics and subtopics)				Required Hours
1	Paints industry overview, Problems and prospects				2
2	Formulation of Primers, zinc rich epoxy, Micaceous iron oxide, zinc chromate and tetraoxy and terraoxy chromate zinc phosphate- based primers, wash primers				4
3	Anti-fouling coatings, Paints for marine environments, vinyl paints				4
4	Road marking paints, Cement paints				2
5	Automotive protection products, paints, finishing and refinishing, Electrodeposition coatings, UV curable coatings				4
6	Coatings for high temperature, Coatings for aerospace and aircrafts				4
7	Electrical insulation coatings, Electrical conducting coatings				4
8	Thermal sensitive paints, Thermal Insulating paints				4
9	Metallic paints, Powder coatings, Coil coatings, Wood finishing, Strippable coatings, lacquers				6
10	Treatment of air for paint application, Surface treatment and paint application methods, Treatment of over sprays				4
11	Reworking of painted products				2
12	Paint application and curing machinery				2
13	Formulation and application of sealants and adhesives				3
<b>Total</b>					<b>45</b>
<b>List of Text Books/ Reference Books</b>					
1	Polymer Chemistry: A Practical Approach (The Practical Approach in Chemistry Series)1st Edition Fred J. Davis Oxford University Press 2004.				
2	Basics of Paint Technology Part I, V. C. Malshe.				
3	Polymer Science by Gowarikar, John Wiley and Sons 1986.				
4	Resins for Surface Coatings, Polyurethanes Polyamides Phenoplasts Aminoplasts Maleic Resins (Waterborne & Solvent Based Surface Coatings Resins & Applications) (Volume III) Volume III Edition				
5	Resins for Surface Coatings, Volume 1 2nd Edition, Resins for Surface Coatings: Acrylics and Epoxies 2nd Edition by H. Coyard (Author), P. Deligny (Author), N. Tuck (Author), P. K. T. Oldring (Editor)				
6	Basics of Paint Technology Part II, Part 2, V. C. Malshe, Prakash C. Malshe, 2008 - Coatings - 624 pages				
7	Principles of polymerization, G. Odian, Wiley – Interscience (1981)				
8	Outlines of Paint Technology Hardcover – December 1, 2000 by Morgan (Author)				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Differentiate various types of paint based on their formulation and application (K4)				
CO2	Analyze various factor affecting synthesize, application of paint and ability to solve the problems observed during either manufacturing or during application of paint. (K4)				
CO3	Formulate the paint recipe based on its final application. (K5)				
CO4	Discuss methods of substrate surface treatment, paint application and curing mechanisms (K2)				
CO5	Design paint formulation considering various ingredients (K5)				

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K4	3	3	2	3	2	3	3	3	3	2	2	2	3	3
CO2	K4	3	3	2	3	2	3	3	3	3	1	2	2	3	3
CO3	K5	3	3	3	3	3	3	3	3	3	2	2	3	3	3
CO4	K2	3	2	2	1	2	3	2	1	1	2	1	3	1	3
CO5	K5	2	1	1	1	1	1	2	1	2	2	1	2	3	1
Course	K5	3	3	3	3	3	3	3	3	3	2	2	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
 K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PEC	Course Code: PST1712	Course Title: Spl 11- Environment Health and Safety of Polymers and Coating	Credits = 4		
	Semester: VI	Total Contact Hours: 60	L	T	P
			2	1	0
<b>List of Prerequisite Courses</b>					
Polymer chemistry and Technology , High Polymer Chemistry, Paint Technology II					
<b>List of Courses where this course will be Prerequisite</b>					
Synthesis of Polymer and resins at laboratory scale and at industrial level. For recycling industry, plastic waste management					
<b>Description of relevance of this course in the B. Tech. (Surface Coating Tech.) Programme</b>					
To give understanding of basics of care to be taken while handling polymer and resin. Safety and hazardous of their manufacturing processes. Knowledge of subject will help student to see the environmental impact by plastic and resin. Current understanding of the benefits and concerns surrounding the use of plastics and look to future priorities, challenges and opportunities. It is evident that plastics bring many societal benefits and offer future technological and medical advances. However, concerns about usage and disposal are diverse and include accumulation of waste in landfills and in natural habitats, physical problems for wildlife resulting from ingestion or entanglement in plastic, the leaching of chemicals from plastic products and the potential for plastics to transfer chemicals to wildlife and humans.					
Sr. No.	Course Contents (Topics and subtopics)				Required Hours
1	Introduction to Health and safety				1
2	Plastics and coatings in the society				1
3	Plastics and coating in the environment				2
4	Plastic waste and coating waste management				2
5	Plastic waste in the marine and terrestrial environment				3
6	Plastic and coating material degradation Regulations for hazardous chemicals in articles/plastic products, coated article.				4
7	Plastic and coating composition and hazardous chemicals like phthalate base plasticizers and Release potential Degradation products Exposure				5
8	Effects Hazard and risk assessment.				4
9	Toxicity Product leaching tests				2
10	Toxicity Identification Evaluations (TIEs)				2
11	Hazard ranking and assessment of plastic and coating Chemicals in plastic and coating formulations				4
12	Polymer Production, Paint production and hazard classifications				4
13	Toxicity of discarded electronic products				3
14	Recycling methods of plastic waste and coating waste and their environmental impact				5
15	Health safety and environment related to Solvent based coating UV coatings				5
16	Hygiene coatings Industrial coatings wood coatings, marine coatings etc.				5
17	Cytotoxicity of nano particles				2
18	Environment Health and Safety Indian and world Policy of Polymers and Coating				3
19	A more sustainable use of plastics and coatings.				3
<b>Total</b>					<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Plastics Materials by J.A. Brydson, Butterworth-Heinemann, 1999 - <u>Technology &amp; Engineering</u> - 920 pages				
2	Handbook of Industrial Chemistry: Organic Chemicals by Mohammad Farhat Ali, Ph.D., Bassam M. El Ali, Ph.D., James G. Speight, Ph.D. McGraw-Hill Education: New York, Chicago, San Francisco, Athens, London, Madrid, Mexico City, Milan, New Delhi, Singapore, Sydney, Toronto, 2005.				
3	SPI Plastics Engineering Handbook of the Society of the Plastics Industry, Inc. by Berins, Michael L., 1991.				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Apply knowledge to understand the environmental and safety issues in chemical industry. (K3)				
CO2	Examine various handling precautions for safely handling monomer and resins (K4)				
CO3	Plan activities to reduce the impact of final product of polymer and coating on environment after use and its waste management. (K5)				
CO4	Identify, formulate and know Polymer & Resins (K5)				
CO5	Practice safety rule and regulation for polymer and resins. Manufacturing process and application				



	Impact and health hazards study of polymer and resins. (K3)
CO6	Discuss various hazard, risk and toxicity evaluation and assessment techniques

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	Course Code: SCT1815	Course Title: Spl 12 - Advanced Paint Technology	Credits = 4		
	Semester: VI	Total Contact Hours: 60	L	T	P
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymers, Analysis and characterization of resins and polymers lab					
<b>List of Courses where this course will be Prerequisite</b>					
Project II					
<b>Description of relevance of this course in the B. Tech. Programme</b>					
To understand in detail the paint rheology and the different additives, called rheology modifiers, used for adjustment of viscosity as per the need. To study in detail surface pretreatment methods and application methods used along with their working principles, advantages and limitations.					
Sr. No.	Course Contents (Topics and subtopics)				Required Hours
1	Industry overview, problems and prospects, Surface pretreatments for metallic substrates like zinc chromate and tetraoxy chromate, zinc phosphate.				5
2	Primers for Metallic substrates like shop primers and wash primers consisting of zinc rich epoxy, Micaceous iron oxide, Electrodeposition primer.				5
3	Primer surface and sealer coat for metallic substrates. Metallic and solid colour top coat and clear coat. Refinishing of automotive paints. Coatings for aerospace and aircrafts.				5
4	Coil coatings, Anti-fouling coatings Electrical conducting coatings Thermal sensitive paints Insulating paints				5
5	Coatings for high temperature Road marking paints				5
6	Paint film defects causes and remedies, Architectural coatings				5
7	Anti-carbonation coating Heat reflective coatings Wood Finishing				5
8	Strippable coatings, lacquers Treatment of air for paint application				5
9	Paint application methods Treatment of over sprays				5
10	Paint application and curing machinery Formulation and application of sealants and adhesives				5
11	Radiation Curing coatings Metallic Coatings				5
12	Paint rheology and different rheology modifiers, Analysis & testing of paints & Paint film				5
<b>Total</b>					<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Polymer Chemistry: A Practical Approach (The Practical Approach in Chemistry Series) 1st Edition Fred J. Davis Oxford University Press 2004				
2	A Practical Course in Polymer Chemistry S. H. Pinner, Borough Polytechnic, London, Pergamon Press, he., New York, 1961				
3	Polymer Science by Gowariker, John Wiley and Sons 1986.				
4	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc 1965.				
5	Principles of polymerization, G. Odian, Wiley – Interscience (1981)				
6	PVC Technology 4th edition by W.V. Titow Elsevier Applied Science				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Develop the concept of paint rheology (K3)				

CO2	Analyze and compare the various Paint properties and solve their defects (K4)
CO3	Prepare primers and analyse compositions for metallic substrates
CO4	Prepare and make the surface ready for further coating application (K5)
CO5	Identify paint film defects and suggest remedies for the same (K2)

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K3	3	3	2	2	2	3	3	3	3	3	2	2	3	3
CO2	K4	3	2	2	2	2	2	2	1	2	3	2	3	2	2
CO3	K5	2	1	1	1	2	1	2	2	2	2	2	3	2	3
CO4	K4	3	3	2	3	2	3	3	3	3	2	2	2	3	3
CO5	K4	3	3	2	3	2	3	3	3	3	2	2	2	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	2	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	<b>Course Code:</b> PST1610	<b>Course Title:</b> Honors Course-II Biopolymers	<b>Credits = 3</b>		
	<b>Semester: VI</b>	<b>Total contact hours: 45</b>	<b>L</b>	<b>T</b>	<b>P</b>

**List of Prerequisite Courses**

Polymer chemistry and Technology, High Polymer Chemistry, Paint Technology II

**List of Courses where this course will be prerequisite**

Synthesis of Polymer and resins at laboratory scale and at industrial level. For recycling industry, plastic waste management

**Description of relevance of this course in the B. Tech. Program**

The course on Biopolymers is highly relevant as it equips students with a deep understanding of the environmental impact of polymer industries and the significance of sustainable materials in various engineering applications. With the knowledge gained, B. Tech. students can contribute to designing eco-friendly products, developing efficient waste management strategies, and implementing green technologies, fostering a sustainable approach in the field of engineering and technology.

Sr. No.	Course Contents	Req. hours
1	Environmental issues related to polymer industry, Design for environment, Life cycle approach, Contribution to energy, feedstock, transport, Gross and net calorific values.	10
2	Polymers in packaging, Common packaging plastics, Waste Stream Categories, Source reduction, Reuse and recycling. Separation and Identification of Plastics Process Technologies for Plastics Recycling	10
3	Polymers in agriculture, Greenhouse films, Plastics in Mulch films, Plastics in silage, Disposal of waste plastic films, Drip irrigation system.	5
4	Flammability of polymers, Release of polymer vapours, Ignition, Combustion of polymer vapours, Fire propagation, Thermal destruction of waste plastics.	10
5	Biopolymers, biobased, bio sourced, compostable, Carbohydrates, polysaccharides, lactides, hydroxyalkanoates, bio isoprene	5
6	Bio additives, starch, cellulose, chitosan, vegetable oils	5
<b>Total</b>		<b>45</b>

**List of Text Books/ Reference Books**

1	"The Environment and Sustainable Development" - Adisa Azapagic, Alan Emsley, Ian Hamerton, University of Surrey, Guildford, UK, Edited by Ian Hamerton
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**Course Outcomes (students will be able to....)**

CO1	Explain the environmental impact of polymer industries and apply the principles of design for the environment and life cycle approach. (K2)
CO2	Illustrate sustainable packaging strategies, including source reduction, reuse, and recycling of common packaging plastics. (K3)
CO3	Analyze the application of polymers in agriculture, such as greenhouse films, mulch films, silage protection, drip irrigation systems, and discuss waste plastic disposal methods. (K4)
CO4	Compare and evaluate the flammability and thermal properties of polymers, along with measures to mitigate flammability risks and thermal destruction of waste plastics. (K3)

**Mapping of Course Outcomes (COs) with Programme Outcomes (POs)**

		PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO9	PO1 0	PO1 1	PO1 2	PSO 1	PSO 2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4
CO1	K2	3	3	2	3	3	2	2	2	3	3	3	3	3	3
CO2	K3	3	3	3	3	3	3	3	3	3	3	3	1	3	3
CO3	K4	3	3	3	2	3	3	3	3	3	3	3	3	3	3
CO4	K3	3	3	3	3	3	3	3	1	3	2	3	3	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

VSEC	Course Code: CEP1714	Course Title: Chemical Engineering Laboratory	Credits = 2		
	Semester: VI		Total contact hours: 60	L	T
			0	0	4
<b>List of Prerequisite Courses</b>					
	Process Calculations, Transport Phenomena, Chemical Engineering Operations, Chemical Reaction Engineering				
<b>List of Courses where this course will be prerequisite</b>					
	Other B. Tech. courses				
<b>Description of relevance of this course in the B. Tech. Program</b>					
Chemical Engineering lab provides students the firsthand experience of verifying various theoretical concepts learnt in theory courses. It also exposes them to practical versions of typical chemical engineering equipment's and servers as a bridge between theory and practice. This particular lab focuses on fluid dynamics, distillation, filtration, drying and sedimentation.					
	<b>Course Contents (Topics and subtopics)</b>				<b>Reqd. hours</b>
1	4 - 6 Experiments on fluid dynamics and heat transfer				24
2	3 - 5 Experiments on Chemical Engineering Operations				16
3	2 – 4 Experiments on Reaction Engineering				12
4	1 – 3 Experiments on process dynamics and control				8
	<b>Total</b>				<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	McCabe W.L., Smith J.C., and Harriott P. Unit Operations in Chemical Engineering, 2014				
2	Bird R.B., Stewart W.E., and Lightfoot, E.N. Transport Phenomena, 2007				
3	Coulson J.M., Richardson J.F., and Sinnott, R.K. Coulson & Richardson's Chemical Engineering: Chemical engineering design, 1996.				
4	Green D. and Perry R. Perry's Chemical Engineers' Handbook, Eighth Edition, 2007.				
<b>Course Outcomes (students will be able to.....)</b>					
1	Learn how to experimentally verify various theoretical principles				
2	Visualize practical implementation of chemical engineering equipment's				
3	Develop experimental skills				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
<b>CO2</b>	K4	3	3	2	1	2	3	3	0	3	3	3	2	2	3
<b>CO3</b>	K4	3	3	2	3	2	2	3	3	3	3	2	2	3	2
<b>Course</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

PCC	<b>Course Code:</b> SCP1608	<b>Course Title:</b> Pr 6- Synthesis, processing and characterization of colorants	<b>Credits = 2</b>		
	<b>Semester: VI</b>	<b>Total Contact Hours: 60 hrs</b>	L	T	P
			0	0	4
<b>List of Prerequisite Courses</b>					
Organic Chemistry, Color Physics					
<b>List of Courses where this course will be Prerequisite</b>					
Advanced paint Technology (SCT1815), Analysis and testing of Paints (SCP1808), Project I (PSP1714), Project II (PSP1811)					
<b>Description of relevance of this course in the B. Tech. (Surface Coating Tech.) Programme</b>					
Study about the types of pigment, their method of synthesis, differentiation between various pigments, characterization of synthesized pigments with various methods					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	Synthesis of pigments like 1. Iron oxide, Iron blue etc 2. Lemon chrome 3. Middle chrome 4. Zinc phosphate and Zinc Chromate 5. Para red 6. Toluidine red 7. Hansa Yellow 8. Lithol red 9. Pthalocyanine blue				1x 4hr/week
2	Characterization and testing of pigments like moisture content, hiding power, yield, bulk density etc.				
3	Use of Muller and Pigment Flusher for dispersion				
4	Qualitative analysis of Pigments & Pigment mixtures.				
<b>List of Text Books/ Reference Books</b>					
1	Encyclopedia of Color Science and Technology, Editors: Luo, Ronnier (Ed.)				
2	Modern colorants: synthesis and structure by A T Peters; H S Freeman				
3	SYNTHESIS OF CHROMOTROPIC COLORANTS. By Ralph A Coleman; John Kazan; Mary Louise Vega; americancyanamid co bound brook nj.				
4	Food Colorants: Chemical and Functional Properties by Carmen Socaciu				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Prepare the various organic and inorganic pigments (K5)				
CO2	Analyze the synthesized pigments qua quantitatively and qualitatively (K4)				
CO3	Plan experiments to separate pigments from the mixture and the analysis (K5)				
CO4	Estimate the process of dispersion, factors affecting on it and use the machineries to perform the same (K5)				
CO5	Use the equipment such as flusher, muller etc. used for processing in paint industry. (K3)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO9	PO1 0	PO1 1	PO1 2	PSO 1	PSO 2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4
CO1	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO2	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO3	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO4	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO5	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
 K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PEC	<b>Course Code:</b> SCP1609	<b>Course Title:</b> Pr 7- Processing of Paints Lab-II	<b>Credits = 2</b>		
	<b>Semester:</b> VI	<b>Total Contact Hours:</b> 60 hrs	<b>L</b>	<b>T</b>	<b>P</b>
			0	0	4
<b>List of Prerequisite Courses</b>					
Organic Chemistry, Color Physics					
<b>List of Courses where this course will be Prerequisite</b>					
Advanced paint Technology , Analysis and testing of Paints, Corrosion Science and Corrosion Prevention					
<b>Description of relevance of this course in the B. Tech. (Surface Coating Tech.) Programme</b>					
Students should be learned about synthesis of some organic pigments and their characterization					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	Identification of pigment and determine Acidity and Alkalinity				1x 4hr/week
2	To Determine Oil absorption value, bulk density, Bleeding tendency and Moisture Content of various Pigments.				
3	Preparation of an Azo pigment.				
4	Synthesis of whiting (CaCO <sub>3</sub> ) and Iron Oxide Pigment				
5	To synthesize various grades of lead chrome pigment.				
6	Preparation of phthalocyanine pigments.				
<b>List of Text Books/ Reference Books</b>					
1	Encyclopedia of Color Science and Technology, Editors: Luo, Ronnier (Ed.)				
2	Modern colorants: synthesis and structure by A T Peters; H S Freeman				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Perform and analyze various testing of pigments (K4)				
CO2	Prepare few extender and pigments e.g. Calcium carbonate and iron oxide, phthalocyanine etc. (K5)				
CO3	Preparation of lead based and phthalocyanine pigments (K5)				
CO4	Test various pigments for their identification (K4)				
CO5	Use the acidity and alkalinity test for the pigment to determine its acidic/ basic behaviour (K3)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO 1	PO 2	PO 3	PO4	PO5	PO 6	PO7	PO 8	PO9	PO1 0	PO1 1	PO1 2	PSO 1	PSO 2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4
CO1	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO2	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO3	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO5	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

# Semester-VII



PCC	Course Code: PST1711	Course Title: Spl 13- Evaluation and testing of polymer and coatings	Credits = 3		
	Semester: VII	Total Contact Hours: 45	L	T	P
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymers , Analysis and characterization of resins and polymers lab					
<b>List of Courses where this course will be Prerequisite</b>					
Project I, Project II, Analysis and Testing of Paints					
<b>Description of relevance of this course in the B. Tech. (Surface Coating Tech.) Program</b>					
Student will able to design the product. Suggest the product for suitable applications. Subject will help student to carry out work in the area of material sciences					
Sr. No.	Course Contents (Topics and subtopics)				Required Hours
1	Glass transition temperature, melting temperature, heat distortion temperature, etc. Sample preparation, standardization, conditioning of sample, processability test, dynamic mechanical analysis, melt flow rate, Vicat softening temperature. Study of a dilatometer. Study of thermo-chemical analysis and differential scanning calorimeter, GPC.				5
2	Fourier transform infrared spectrometry, Ultraviolet - visible spectrometry, Nuclear magnetic resonance spectrometry, Mass spectrometry, X-ray diffraction spectrometry, Gas chromatography. Scanning electron microscopy, travelling electron microscope Molecular weight determination Viscosity of polymer solutions and polymers: Their significance, application to polymers using different viscometers.				5
3	Surface volume resistivity, Breakdown voltage, Arc resistance, Tan Delta, Tensile strength, flexural strength, impact resistance, percentage elongation, tear test, fatigue and wear, hardness, compressive strength time dependant properties like creep, stress, relaxation, etc. Refractive index, gloss, color matching, haze, limiting oxygen index, smoke density, Tests for adhesives Identification of polymers using chemical methods ESCR.				5
4	Analysis of Paints, Theory and practice in testing of paints, Paint film defects and their remedies. Analytical instruments in paints technology, UV, IR, GCMS, X-Ray Diffraction, LCMS MS, Microscopy				5
5	Particle size analysis of pigments, Accelerated weathering of paints Evaluation and Testing of Synthetic Enamel, Primer, Emulsion paint, Intermediate Coat				5
6	NVM, Viscosity, WPL, Grind, Hiding, Drying Time, Scratch Hardness, Impact Test, Flexibility, Gloss Dry Film Thickness.				5
7	Acid Alkali, and Water Resistance, Adhesion As per IS101, Corrosion Resistance by Salt Spray and Humidity Cabinet				5
8	Accelerated Exposure of Paints in QUV and Atlas Apparatus, % Solids, Scrub Resistance, Stain Resistance				5
9	Rheology of Paint system, Colour Matching of Synthetic Enamel, Plastic Emulsion Paint and Distemper.				5
<b>Total</b>					<b>45</b>
<b>List of Text Books/ Reference Books</b>					
1	Polymer Chemistry: A Practical Approach (The Practical Approach in Chemistry Series) 1st Edition <u>Fred J. Davis</u> Oxford University Press 2004				
2	A Practical Course in Polymer Chemistry S. H. Pinner, Borough Polytechnic, London, Pergamon Press, he., New York, 1961				
3	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co,1994				
4	Polymer Science by Gowarikar, John Wiley and Sons 1986.				
5	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc1965				
6	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc1988				
7	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co,1994				

8	Principles of polymerization, G.Odian, Wiley – Interscience (1981)
9	PVC Technology 4th edition by W. V. TitowElsevier Applied Science
<b>Course Outcomes (Students will be able to.....)</b>	
CO1	Interpret the significance for polymer characterization technique such as NMR (K3)
CO2	Analyse and understand the properties of polymers such as mechanical, electrical etc. hence they can suggest the various polymer depending upon specific application (K4)
CO3	Illustrate the significance of rheology is well understood by student and correlation of rheology and temperature is understood hence student can apply this knowledge while processing of polymer (K3)
CO4	Interpret theoretically importance of FTIR, NMR etc. hence in case of any hand on experiment with such equipment they can relate this knowledge to practice. (K4)
CO5	Relate theoretical knowledge to identify any unknown sample. (K4)
CO6	Analyze and evaluate variety of wet paint and film properties including mechanical, chemical, corrosion, adhesion and rheology

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO9	PO1 0	PO1 1	PO1 2	PSO 1	PSO 2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4
CO1	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
CO2	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO3	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO5	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO6	K4	3	3	3	2	3	3	3	3	3	3	3	2	3	3
Course	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	Course Code: <b>SCT1701</b>	Course Title: <b>Spl 14- Curing Mechanism of Coating</b>	Credits = 2		
	Semester: VII	Total contact hours: 30	L	T	P
			1	1	0
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymers, Analysis and characterization of resins and polymers lab.					
<b>List of Courses where this course will be prerequisite</b>					
Adhesion and adhesives					
<b>Description of relevance of this course in the B. Tech. Program</b>					
The course on "Curing Mechanism" holds significant relevance in the realm of polymer science and engineering. Understanding the diverse curing processes and their mechanisms is vital for tailoring the properties of cured materials to specific applications. Graduates of this course will be equipped with the knowledge to optimize curing methods, ensuring the production of high-quality materials with enhanced mechanical, chemical, and physical attributes, leading to improved performance and longevity in various industrial sectors, such as manufacturing, electronics, coatings, adhesives, and biomedical applications.					
<b>Sr. No.</b>	<b>Course Contents</b>				<b>Reqd. hours</b>
1	Need for curing mechanism				6
2	Resin combination for curing				8
3	Types of curing process: room temperature, thermal, air drying, ultraviolet (UV), electron beam (EB)				8
4	Various mechanisms of curing				8
<b>List of Text Books/ Reference Books</b>					
1	Radiation Curing by S. Peter Pappas, 2013				
2	UV and EB Curing Formulation for Printing Inks Coatings & Paints by R. Holman ,				
3	Organic Coatings by Zeno W. Wicks, Jr, S. Peter Pappas, Douglas A. Wicks, 2005				
<b>Course Outcomes (students will be able to.....)</b>					
CO1	Explain the importance and necessity of curing mechanisms in the context of polymer-based materials and their applications. (K2)				
CO2	Illustrate the process of resin combination for curing and demonstrate how different combinations affect the properties of the cured materials. (K3)				
CO3	Develop a comprehensive understanding of the various curing processes, including room temperature, thermal, air drying, ultraviolet (UV), and electron beam (EB) curing methods. (K6)				
CO4	Analyze and compare the different mechanisms of curing, evaluating their advantages, limitations, and suitability for specific material types and applications. (K2)				
CO5	Classify and determine the optimal curing processes for specific materials and manufacturing scenarios, considering factors such as material composition, curing time, and environmental conditions. (K4)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K2	2	2	1	2	3	3	2	1	1	2	2	2	3	2
CO2	K3	3	2	2	3	1	1	1	1	1	2	1	2	2	2
CO3	K6	3	1	1	2	3	2	1	1	2	3	2	3	3	2
CO4	K2	3	2	2	3	2	2	2	2	2	3	2	2	2	1
CO5	K4	2	2	2	3	2	3	2	2	2	2	2	3	2	2
Course	K6	3	2	2	3	3	3	2	2	2	3	2	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PEC	<b>Course Code:</b> SCT1813	<b>Course Title:</b> PEC - Printing Inks	<b>Credits = 3</b>		
	<b>Semester: VII</b>	<b>Total Contact Hours: 45</b>	<b>L</b>	<b>T</b>	<b>P</b>
			2	1	0

**List of Prerequisite Courses**

Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymers, Analysis and characterization of resins and polymers lab

**List of Courses where this course will be Prerequisite**

Adhesion and adhesives, Analysis and testing of paints, Sustainability of polymers

**Description of relevance of this course in the B. Tech. Programme**

To understand the basic printing inks and its various formulations.  
 To study about various testing and analysis methods for printing inks.  
 To understand the basic concept behind the ink-substrate interactions like adhesion, smudging, water resistance, etc.  
 To study about various printing inks application methods like flexographic printing, lithographic printing, screen printing, ink-jet printing, UV curable printing, etc.

Sr. No.	Course Contents (Topics and subtopics)	Required Hours
1	Manufacture of paper qualities and properties of paper	5
2	Letterpress printing: Process characteristics raw materials formulations for different substrates- ink related problem and their solution, latest developments	6
3	Screen printing: Process characteristics raw materials formulations for different substrates- ink related problem and their solution, latest developments	6
4	Flexography: Process- characteristics raw materials formulations for different substrates ink related problem and their solution, latest developments	6
5	Gravure: Process characteristics raw materials formulations for different substrates ink related problem and their solution, latest developments.	6
6	Lithography: Process characteristics raw materials formulations for different substrates ink related problem and their solution, latest developments.	6
7	Non impact printing	5
8	Other than above printing method: pad printing, transfer printing and latest development	5
	<b>Total</b>	<b>45</b>

**List of Text Books/ Reference Books**

1	Modern Technology Of Printing Inks
2	The Printing Ink Manual, R. H. Leach, Springer Science & Business Media, 30-Sep-1993 - Art - 993 pages
3	Printing Ink Technology Books Industrial Technologies, India Nai Sarak, New Delhi, Delhi
4	Gravure: Process and Technology Hardcover – Import, Dec 1997by Gravure Association of America (Author)
5	Gravure Process and Technology Hardcover – 2003by Gravure Education Foundation (Author)

**Course Outcomes (Students will be able to.....)**

CO1	Explain the importance of printing ink in various industries (K2)
CO2	Describe about manufacturing of paper and properties of the same (K2)
CO3	Analyse and differentiate between various types of printing inks (K4)
CO4	Apply the knowledge to understand printing ink properties. (K3)
CO5	Illustrate and Analyse the surface preparation methods for printings (K4)

**Mapping of Course Outcomes (COs) with Programme Outcomes (POs)**

	PO 1	PO 2	PO 3	PO4	PO5	PO 6	PO7	PO 8	PO9	PO1 0	PO1 1	PO1 2	PSO 1	PSO 2
	K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4

CO1	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
CO2	K2	3	2	1	2	1	3	3	3	3	3	3	1	3	2
CO3	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO4	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
CO5	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO6	K2	2	2	2	3	3	1	2	2	2	3	2	3	2	3
Course	K4	3	3	2	3	3	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PEC	Course Code: SCT1703	Course Title: High Performance Coatings	Credits = 2		
	Semester: VII	Total Contact Hours: 30	L	T	P
			1	1	0
<b>List of Prerequisite Courses</b>					
Technology of Thermoset Polymers, Structure Property Relationship, Paint Technology I, Advanced Paint Technology, Synthesis and Characterization of Resins and Polymers Lab					
<b>List of Courses where this course will be prerequisite</b>					
Paint Technology II, Corrosion Science and Corrosion Prevention, Analysis and Testing of Paints					
<b>Description of relevance of this course in the B. Tech. Program</b>					
<b>Course Contents (Topics and Subtopics)</b>					<b>Required Hours</b>
1	Coatings for off shore structure, Coatings for chemical plant				5
2	Anti carbonation coating, pipe line coatings antireflective coating wind turbine coating				10
3	Powder coatings and other high performance coatings & their importance, polymers used in powder coatings				10
4	Coatings for electronics				5
	<b>Total</b>				<b>30</b>
<b>List of Textbooks/Reference Books</b>					
1	A Guide to High-performance Powder Coating by Bob Utech				
2	User's Guide to Powder Coating, Fourth Edition by Nicholas Liberto				
3	Beginning Powder Coater's Handbook: An Introduction to Powder Coating by Tracy Norris				
4	High Performance organic coating by A.S. Khanna, Woodhead publishing				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Design, develop and formulate chemical resistant coatings (K5)				
CO2	Design, develop and formulate anti carbonation and wind turbine coatings (K5)				
CO3	Design, develop and formulate environmental aspects of powder coatings (K5)				
CO4	Design, develop and formulate environmental and economic aspects of coatings in electronics (K5)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K5	3	2	2	2	3	3	2	1	1	1	1	2	2	2
CO2	K5	3	2	1	1	2	3	2	1	1	2	2	3	3	3
CO3	K5	2	1	3	3	2	3	3	2	2	1	1	3	3	3
CO4	K5	1	2	3	2	3	1	3	2	2	2	3	2	2	1
Course	K5	3	2	3	3	3	3	3	2	2	2	3	3	2	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

PEC	Course Code: PHT1440	Course Title: Intellectual Property Rights	Credits = 2		
	Semester: VII		L	T	P
		Total Contact Hours: 30	1	1	0
<b>List of Prerequisite Courses</b>					
Nil					
<b>List of Courses where this course will be Prerequisite</b>					
Nil					
<b>Description of relevance of this course in the B. Tech. (Pharm. Chem. Tech.) Programme</b>					
To train the students with respect to basics of Intellectual Property Rights (IPR)					
Sr. No.	Course Contents (Topics and Subtopics)				Required Hours
1	Introduction to Intellectual Property: overview describing definition, need and evolution				2
2	IPR related laws: Biodiversity				2
3	Introduction to WIPO and Treaties under WIPO				2
4	<b>Type of Intellectual Property: Copyright</b> Introduction, Process of filing, rights achieved				2
5	<b>Type of Intellectual Property: Trademarks</b> Introduction, Process of filing, rights achieved				2
6	<b>Type of Intellectual Property: Geographical Indications</b> Introduction, Process of filing, rights achieved				2
7	<b>Type of Intellectual Property: Industrial Design</b> Introduction, Process of filing, rights achieved				2
8	<b>Type of Intellectual Property: Trade Secret</b> Introduction, Process of filing, rights achieved				3
9	<b>Type of Intellectual Property: patent</b> Introduction Patent and traditional knowledge Indian patent Act Process of filing Rights achieved				4
10	Patentability w.r.t. regional requirements				2
11	Patent filing under Paris Convention Treaty (PCT)				3
12	Role of IPR in Pharmaceuticals				4
<b>Total</b>					<b>30</b>
<b>List of Text Books/Reference Books</b>					
1	All documentation from World Intellectual Property Organization (www.wipo.int )				
2	Indian Patent Act (www. ipindia.nic.in)				
3	Pharmaceutical Product Development: Insights into Pharmaceutical Processes, Management and Regulatory Affairs, Patravale V, Rustomjee M, Dsouza J. 2016, CRC press				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	explain various types of Intellectual Property Rights.(K2)				
CO2	explain the importance of Intellectual Property Rights in relevance to pharmaceutical inventions.(K2)				
CO3	implement the desired practises during professional activities for preserving IPRs.(K4)				
CO4	interpret and analyze reactions having different functionalities, deduce and solve problems related to the reactions as well as apply them, if need be.(K4)				

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+S	K3	K4
<b>CO1</b>	K2	3	3	2	3	2	0	3	3	3	3	3	2	3	3
<b>CO2</b>	K2	3	3	3	1	3	3	3	2	3	3	0	3	3	3
<b>CO3</b>	K4	3	2	2	3	3	3	2	3	2	3	2	2	1	3
<b>CO4</b>	K4	3	3	3	3	2	3	3	3	3	3	3	3	3	3
<b>Course</b>	K4	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain



PCC	<b>Course Code:</b> PST1714	<b>Course Title:</b> Honors Course III- Nanomaterials and their Applications	<b>Credits = 4</b>		
	<b>Semester: VII</b>	<b>Total Contact Hours: 60</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>3</b>	<b>1</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymer analysis and characterization of resins and polymers lab, Environment Health and Safety of Polymers and Coating, Evaluation and testing of Polymers and Coatings .					
<b>List of Courses where this course will be Prerequisite</b>					
Adhesion and adhesives					
<b>Description of relevance of this course in the B. Tech. Programme</b>					
Able to understand the significance of nanosize. Able to synthesized various nanomaterials and nanocomposites Gets aware about new and emerging technology in Polymer and Coating industry such as carbon nanotubes and anticorrosive coating with the use of same.					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	Definition, Classification of nanomaterial and its unique properties.				7
2	Synthesis, properties and applications of Carbon nanotubes.				7
3	Synthesis, properties and applications fullerenes.				7
4	Synthesis, properties and applications in organic nanomaterials like titanium dioxide, zinc oxide etc.				7
5	Synthesis, properties and applications of nanoparticles of gold, silver celluloses etc.				10
6	Dendrimers, Nanoclay sand its differnt treatment.				7
7	Polymer nanocomposites and its processing properties, application sand charecterization.				8
8	Nanocoatings, safety regulations of nanomaterials.				7
				<b>Total</b>	<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Structural Nanocomposites: Perspectives for Future Applications (Engineering Materials) Hardcover – Import, 16 Dec 2013 by James Njuguna.				
2	Multifunctional Polymer Nanocomposites, ISBN13 : 9781439816820 ISBN10 : 1439816824 Publisher : Taylor & Francis Inc Pages : 466..				
3	Nanocomposites Organiques a Matrice de Silicium Poreux (French, Paperback, Diyana Badeva)				
4	Thermoset Nanocomposites for Engineering Applications, Author : Kotsilkova, R.				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Identify the significance of nanosize. (K3)				
CO2	Design various nanomaterials and nanocomposites (K5)				
CO3	Discover safety measurements and to deal with any emergency when working with nanoparticles (K4)				
CO4	Examine property variation with differentiation of particle size of any filler, pigment etc. in polymer composite, coating etc. (K4)				
CO5	Inspect about new and emerging technology in Polymer and Coating industry such as carbon nanotubes and anticorrosive coating with the use of same.(K4)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO9	PO1 0	PO1 1	PO1 2	PSO 1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4
CO1	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
CO2	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO3	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

CO5	K4	3		3	3	3	2	33	3	3	3	3	3	3	3
CO6	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

RM	<b>Course Code:</b> <b>PHT1442</b>	<b>Course Title: Literature Review (Research Methodology – I)</b>	<b>Credits = 2</b>		
	<b>Semester: VII</b>	<b>Total contact hours: 45</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>1</b>	<b>0</b>	<b>2</b>
<b>Course Outcomes (students will be able to.....)</b>					
<b>List of Prerequisite Courses</b>					
1	NA				
<b>List of Courses where this course will be prerequisite</b>					
1	NA				
<b>Description of relevance of this course in the B. Chem. Engg. Program</b>					
The formal exposure to various elements of research methods such as problem formulation, literature search, planning of various activities, documentation, budgeting, purchase, report/thesis compilation, manuscript writing, patent drafting, is critical for polishing the naïve research attitude and aptitude in the PG students of the programme. The course is designed to formally introduce various concepts of research methodology in stepwise manner to the students					
	<b>Course Contents (Topics and subtopics)</b>				<b>Reqd. hours</b>
1	Introduction of Course Academic Honesty Practices General philosophy of science & Arguing About Knowledge Case studies in science history				3
2	Motivation and Background Motivation/Demotivation for Research, Building Background for Research and How to read research papers				3
3	Time Management (Academic and Non-academic time), Effort Management, Plan execution, Energy Management Issue, Role and expectation of research supervisor and student				4
4	Finding and Solving Research Problems What is Research, How to start?, Approaches to find research problems and psychological experiments Literature survey, Textbooks, Review and research papers How to ask Questions What is worthwhile research problem, Analytical and synthetic research approach				4
5	Finding and Solving Research Problems What is Research, How to start?, Approaches to find research problems and psychological experiments Literature survey, Textbooks, Review and research papers, critical review of research papers, how to write literature survey report, How to ask Questions, formulating research questions,				4
6	What is worthwhile research problem, Analytical and synthetic research approaches How to solve research problems, designing work plan, importance of objectives, activity and strategizing research work. Design of timeline for work plan (Gantt Chart etc), Grant Writing Guidelines				4
7	Experimental Research Inventory Management, Material Management Learning required skills for research, Documentation and lab notebook guidelines, Safety aspects in chemical/biological research				4
8	Methods and Tools used in Research: Qualitative studies; Quantitative studies; Simple data organization; Descriptive data analysis; Limitations and sources of error; Inquiries in form of Questionnaire, Opinionnaire or by interview; Statistical analysis of data including Variance, Standard deviation, Students 't' test and Analysis of variance (ANOVA), Correlation data and its interpretation, Computer data analysis				6
9	Scientific Writing Skeleton of research paper, author guidelines, good writing skills, importance of discussion, Macro-level discussion. Structure of the documents. General issues of presentability. Micro-level discussion. Stylistic issues. Examples of bad and good writings.				6
10	Publishing and Reviewing Publication process, How to publish papers, where to submit, Review process and reacting to a review report Reviewing scientific papers				4

11	Scientific Norms and Conventions Authorship. Plagiarism. Simultaneous submissions. Reviewing norms. Referring to other papers. Use of data. Collaborative Research Work	3
<b>List of Textbooks</b>		
	Menzel, D.; Writing a Technical Paper; McGraw-Hill, United States (1961).	
	Best, J. W., Kahn, J. V., Jha, A. K.; Research in Education; 10th ed.; Pearson, New Delhi, India (2005)	
<b>List of Additional Reading Material / Reference Books</b>		

<b>Course Outcomes (Students will be able to.....)</b>	
CO1	Understand the basic concepts of research and the components therein, formally (K2)
CO2	Understand and appreciate the significance of statistics in Chemical Technology, Pharmacy and Chemical Engineering (K2)
CO3	Understand and apply importance of literature survey in research design (K3)
CO4	Understand an in-depth knowledge on the documentation in research(K2)
CO5	Evaluate importance of various parts of a research report/paper/thesis in presentation of research results(K4)
CO6	<b>Prepare and Deliver a model research presentation (K5)</b>
CO7	Understand the significance of various types of IPRs in research(K1)
CO8	Create a model research project(K6)

	<b>Course Code: PHP1443</b>	<b>Course Title: Design and Analysis of Experiments (Research Methodology – II)</b>	<b>Credits =2</b>		
	<b>Semester: VII</b>	<b>Total contact hours: 45</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>1</b>	<b>-</b>	<b>2</b>
<b>List of Prerequisite Courses</b>					
	Applied Mathematics I				
<b>List of Courses where this course will be prerequisite</b>					
	This course is required for graduating engineers to function effectively in Industry, Academia and other professional spheres. This course is in Semester VIII				
<b>Description of relevance of this course in the B.Tech. Program</b>					
Modern day manufacturing activities and R&D activities need decisions taken with a scientific rigour and should be well-supported by 'statistics'. Chemical Technologist graduates who will serve industry as well as postgraduate research students who will serve industry, R&D organisations, or academic research should have a reasonably good background of statistical decision making. This also involves extraction of meaningful data from well-designed minimal number of experiments at the lowest possible material costs. This course will also help the students in all domains of their life by imparting them a vision for critical appraisal and analysis of data.					
	<b>Course Contents (Topics and subtopics)</b>				<b>Reqd. hours</b>
1	Fundamental principles of classical design of experiments Strategy of Experimentation, Typical applications of Experimental design, Basic Principles, Guidelines for Designing Experiments.				4
2	Review of Probability and basic statistical inference: Concepts of random variable, probability, density function cumulative distribution function. Sample and population, Measure of Central tendency; Mean median and mode, Measures of Variability, Concept of confidence level. Statistical Distributions: Normal, Log Normal & Weibull distributions, Hypothesis testing.				3
3	Experiments with a Single Factor: The Analysis of Variance Fixed effect model and Random effect model, Model adequacy checking, Contrasts, Orthogonal contrasts, Regression Models and ANOVA, Violation of Normality Assumption: Kruskal-Wallis test. Randomized block designs, Latin square designs, Balanced Incomplete Block Designs				6
4	Factorial designs: Definition, Estimating model parameters, Fitting response curves and surfaces.				3
5	The $2^k$ Factorial Design, Blocking and Confounding in the $2^k$ Factorial Design; Focus of $2^2$ and $2^3$ designs, Blocking and Confounding in the $2^k$ Factorial Design.				6
6	Plackett Burman methods, Central Composite Design (CCD)				3
7	Descriptive Statistics, Probability Distribution and testing of Hypothesis using R				4
8	Regression techniques, diagnostic checks, ANOVA using R and implementation of contrasts.				4
9	Construction of Balanced Incomplete Block Designs and data analysis using R				4
10	Analysis of factorial designs using R, understanding output and interpretation.				4
11	Factorial designs, Data analysis and interpretation.				4
<b>List of Text Books / Reference Books</b>					
1	Douglas C. Montgomery, Design and Analysis of Experiments, 8 <sup>th</sup> Edition, John Wiley & Sons, Inc. 2013				
2	Box, G. E., Hunter, W.G., Hunter, J.S., Hunter, W.G., Statistics for Experimenters: Design, Innovation, and Discovery, 2nd Edition, Wiley, 2005.				
3	John Lawson, Design and Analysis of Experiments with R, CRC Press, 2015				
4	Dieter Rasch, Jürgen Pilz, Rob Verdooren, Albrecht Gebhardt Optimal Experimental Designs with R. CRC Press, 2011.				
5	José Unpingco, Python for Probability, Statistics, and Machine Learning, Springer, 2019				
6	Response Surface Methodology: Process and Product Optimization using Designed Experiments: R. H. Myers, D. C. Montgomery.				
7	Introduction to Statistical Quality Control: D. C. Montgomery.				
8	Design of Experiments in Chemical Engineering: Živorad R. Lazić.				
<b>Course Outcomes (students will be able to.....)</b>					
1	Students should be able to understand basic principles of design of experiments.				
2	Students should be able to perform statistical analysis of single experiments and do post hoc analysis.				

3	Students should be able to conduct experiment and analyse the data using statistical methods.	
4	Students should be able to choose an appropriate design given the research problem.	
5	Students should be able to perform statistical analysis of different designs using R and interpret the results.	

Mapping of Course Outcomes (COs) with Programme Outcomes (POs)															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
<b>CO2</b>	K3	3	2	2	2	2	3	3	3	1	2	3	2	3	1
<b>CO3</b>	K3	3	3	2	2	1	1	3	3	3	3	3	2	2	3
<b>CO4</b>	K3	3	3	2	2	2	3	0	2	3	3	3	2	3	1
<b>CO5</b>	K3	3	2	2	0	2	3	3	3	1	3	0	2	3	3
<b>Course</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

<b>Project</b>	<b>Course Code:</b>	<b>Course Title:</b>	<b>Credits = 4</b>		
		<b>Project – I</b>	<b>L</b>	<b>T</b>	<b>P</b>
	<b>Semester: VII</b>	<b>Total Contact Hours: 120</b>	<b>0</b>	<b>0</b>	<b>8</b>
<b>List of Prerequisite Courses</b>					
Research Methodology					
<b>List of Courses where this course will be prerequisite</b>					
Project – II					
<b>Description of relevance of this course in the B. Tech. Program</b>					
The course is designed to help students develop a skill-set for solving a research problem related to Pharmaceutical Sciences and Technology. The course presents an opportunity to the students for fine-tuning their scientific communication skills, oral as well as written.					
	<b>Course Contents (Topics and Subtopics)</b>				<b>Required Hours</b>
1	The Teachers will communicate various research topics of potential interest to the Pharmaceutical Sciences and Technology field to all the students based on the interest and facilities available. Each student, based on his/her interest and merit, selects the research topic and is allotted a supervisor. The work involves detailed review of the literature, formulation of research project, hypothesis, objectives, methodology, possible expected outcomes, planning for experimentation, experimental trials, data generation and analysis. Finally, the student will compile the report as per the communicated format and then present in front of the Evaluators.				120
	<b>Total</b>				<b>120</b>
<b>List of Textbooks/Reference Books</b>					
1	Relevant research articles, patents, review articles, conference proceeding, book chapters and books				
<b>Course Outcomes (Students will be able to.....)</b>					
<b>CO1</b>	Develop critical thinking to identify the research gap for the project (K5)				
<b>CO2</b>	Formulate a scientific question and approach to solve it (K6)				
<b>CO3</b>	Plan the experimental methodology for the project (K5)				
<b>CO4</b>	Develop skills to communicate the research plan effectively (K6)				
<b>CO5</b>	Develop skills for writing a scientific document on the research work (K6)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		<b>PO1</b>	<b>PO2</b>	<b>PO3</b>	<b>PO4</b>	<b>PO5</b>	<b>PO6</b>	<b>PO7</b>	<b>PO8</b>	<b>PO9</b>	<b>PO10</b>	<b>PO11</b>	<b>PO12</b>	<b>PSO1</b>	<b>PSO2</b>
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+S	K3	K4
<b>CO1</b>	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
<b>CO2</b>	K6	3	3	3	3	3	3	3	3	3	3	2	3	3	1
<b>CO3</b>	K5	3	2	3	3	3	3	3	1	3	3	3	3	3	3
<b>CO4</b>	K6	3	3	3	3	3	2	3	3	3	0	3	3	2	3
<b>CO5</b>	K6	3	3	3	3	3	3	3	3	3	3	3	3	3	3
<b>Course</b>	K6	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain

PCC	<b>Course Code:</b> SCP1701	<b>Course Title:</b> Pr 8 - Processing and characterization of paints	<b>Credits = 2</b>		
	<b>Semester: VII</b>	<b>Total Contact Hours: 60</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>0</b>	<b>0</b>	<b>2</b>
<b>List of Prerequisite Courses</b>					
Polymer science and Technology (PST1301), Polymer chemistry and Technology (PST1303) Technology of Thermoset polymers (PST1506), Analysis and characterization of resins and polymers lab (PSP1504)					
<b>List of Courses where this course will be Prerequisite</b>					
Adhesion and adhesives, Analysis and testing of paints, Sustainability of polymers					
<b>Description of relevance of this course in the B. Tech. Programme</b>					
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>				<b>Required Hours</b>
1	High-speed stirring				<b>1x4hr/week</b>
2	Ball milling				
3	Sand milling				
4	NVM, Viscosity, WPL, Grind, Hiding, Drying Time,				
5	Scratch Hardness, Impact Test, Flexibility				
6	Sap Value, colour, Refractive Index				
	<b>Total</b>				<b>60</b>
<b>List of Text Books/ Reference Books</b>					
1	Basics of paint technology I- V.C.Malshe				
2	Testing of paints- Shreekant patil				
<b>Course Outcomes (Students will be able to.....)</b>					
CO1	Discuss wet paint testing methods (K2)				
CO2	Discover pigment grinding and dispersing methods with working, advantages and disadvantages of each (K3)				
CO3	Explain evaluation of different mechanical and optical properties of paints (K2)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
CO1	K2	3	2	2	1	2	2	2	1	2	2	1	2	2	2
CO2	K3	3	3	3	3	3	2	1	2	2	2	1	3	3	2
CO3	K2	3	2	2	3	3	1	1	1	2	2	1	3	2	2
Course	K3	3	3	3	3	3	2	2	2	2	2	1	3	3	2

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain



# Semester-VIII

PCC	Course Code: PST1801	Course Title: SPL-15: Adhesion and adhesives	Credits = 3		
	Semester:	Total Contact Hours: 45	L	T	P
			3	0	0
<b>List of Prerequisite Courses</b>					
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymer analysis and characterization of resins and polymers lab, Environment Health and Safety of Polymers and Coating, Evaluation and testing of Polymers and Coatings.					
<b>List of Courses where this course will be Prerequisite</b>					
Adhesion and adhesives					
<b>Description of relevance of this course in the B. Tech. Programme</b>					
The course on "Adhesion and Adhesives" holds significant relevance in various engineering and industrial sectors. It equips students with essential knowledge about adhesive bonding, joint design, and surface preparations, which are essential in industries like aerospace, automobile, construction, and electronics. Understanding different types of adhesives and coatings enables students to select appropriate materials for specific applications, contributing to efficient and cost-effective manufacturing processes. Additionally, knowledge of surface coatings and their evaluation is crucial for professionals in the paint and coating industry, ensuring the development of high-quality and durable surface finishes in a wide range of applications.					
Sr. No.	Course Contents (Topics and Subtopics)				Required Hours
1	Concepts and terminology, functions of adhesives, Theories of adhesion, advantages and disadvantages of adhesive bonding, criteria for selection of adhesives, applications, advantages and limitations, troubleshooting, various polymers used in adhesive applications, Types of substrates.				8
2	Types of adhesives, structural adhesives, Urethane structured adhesives, Modified acrylic structural adhesives, phenolic adhesives and modifiers, anaerobic adhesives, cyanoacrylate adhesives, methods of adhesive, Hot melt adhesives, pressure sensitive adhesives, RTV Silicone adhesives, sealants, water based adhesives. Specialty adhesives, adhesives in aerospace, adhesive in automobile industry, conductive adhesives,				8
3	Joint design, stress, types of joints, selection of joint detail, joint criteria, surface preparation of adherend -metals, plastics and rubbers. Adhesive bonding process- methods for adhesives application and bonding equipment-, testing and quality control. Testing of adhesives Industrial adhesives				8
4	Fundamentals of surface phenomenon, surface energy and surface tension. Basics of adhesion. Surface preparations, Introduction to surface coatings –Components of paints. Pigments, pigment properties, different types, extenders, solvents, oils, driers, diluents, lacquers, varnishes, paint preparation, formulation, factors affecting pigment dispersion, and preparation of pigment dispersion.				8
5	Different types of paints- classification based on polymeric resin, emulsion, oil and alkyd paints, acrylic paints, epoxy coatings, polyurethane, silicones, chlorinated rubbers. Classification based on application, fluoro polymers, vinyl resins, appliance furnishes, automotive finishes, coil coatings, can coatings, marine coatings, and aircraft coatings. Surface preparation and paint application				8
6	Paint properties and their evaluation – mechanism of film formation, factors affecting coating properties, methods used for film preparation – barrier properties, optical properties, ageing properties, rheological properties and adhesion properties of coatings				5
<b>Total</b>					<b>45</b>
<b>List of Text Books/ Reference Books</b>					
1	Handbook of Adhesives – Skeist, Irvind, Van Nistrand, New York, 1990, 3rd Edition Gerald L. Schreberger, Adhesive in manufacturing, Marcel Dekker Inc., New York, 1983				
2	W.C. Wake, Adhesion and the formulation of adhesives. Applied Science Publishers, London, 1976				
3	Swaraj Paul, Surface Coatings, John Wiley & Sons, NY, 1985				
4	George Mathews, Polymer Mixing Technology, Applied Science Publishers. Sheilds, Hand book of				

adhesives, Butterworths, 1984	
<b>Course Outcomes (Students will be able to.....)</b>	
CO1	Explain the fundamental concepts and theories of adhesion and adhesives, including the functions, advantages, and limitations of adhesive bonding in various applications. (K2)
CO2	Illustrate the different types of adhesives and their classifications, such as structural adhesives, specialty adhesives, and water-based adhesives, while analyzing their specific properties and suitability for diverse substrates. (K3)
CO3	Develop an understanding of joint design and stress distribution, and determine appropriate joint details and surface preparation techniques for adherends made of metals, plastics, and rubbers in adhesive bonding processes. (K4)
CO4	Analyze the role of industrial adhesives in specific industries, including aerospace, automotive, electrical, and construction, evaluating the significance and challenges faced in their applications. (K4)
CO5	Compare various surface coatings, paints, and their formulations, examining factors affecting pigment dispersion, paint properties, and film formation, while discussing the evaluation of coating properties like adhesion, barrier properties, and optical properties.(K5)
CO6	Discuss the concepts of surface phenomena, surface properties and importance of surface preparation in coatings

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A+Psy	K3	K4
CO1	K2	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO2	K3	3	2	2	3	3	3	3	3	3	3	3	3	3	3
CO3	K4	3	3	3	2	3	3	3	3	3	3	3	3	2	3
CO4	K4	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO5	K5	3	3	2	3	2	3	3	3	3	3	3	2	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	Course Code: SCT1816	Course Title: Honors Course-IV - Corrosion Science and Corrosion Prevention	Credits = 4		
	Semester: VIII		Total Contact Hours: 45	L	T
			3	1	0

**List of Prerequisite Courses**

Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymers, Analysis and characterization of resins and polymers lab

**List of Courses where this course will be Prerequisite**

None

**Description of relevance of this course in the B. Tech. Program**

To understand the basics of corrosion- theory, causes, mechanism of corrosion. To study how corrosion can be detected and prevented

Sr. No.	Course Contents (Topics and subtopics)	Required Hours
1	Introduction to corrosion	5
2	Mechanism of corrosion Types of corrosion	5
3	Detection of corrosion	5
4	Methods of preventing corrosions	5
5	Pigments used in corrosion prevention.	5
6	Binders used in corrosion prevention.	5
7	Formulations of primers for Industrial and non-industrial environment.	5
8	Best methods and practices followed before and during application of paints.	5
9	Different characterization and test methods for prevention of corrosion of metallic substrates.	5
<b>Total</b>		<b>45</b>

**List of Text Books/ Reference Books**

1	Polymer Chemistry: A Practical Approach (The Practical Approach in Chemistry Series) 1st Edition Fred J. Davis Oxford University Press 2004
2	A Practical Course in Polymer Chemistry S. H. Pinner, Borough Polytechnic, London, Pergamon Press, he., New York, 1961
3	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994
4	Polymer Science by Gowariker, John Wiley and Sons 1986
5	Encyclopedia of Polymer Science and Technology, John Wiley and Sons, Inc 1965.
6	Encyclopedia of Polymer Science and Engineering, John Wiley and Sons, Inc 1988.
7	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994

**Course Outcomes (Students will be able to....)**

CO1	Distinguish various types of corrosion- theory, causes, mechanism of corrosion. (K4)
CO2	Analyse various factors/environments that facilitate corrosion. (K4)
CO3	Plan and propose various technique for detection and prevention of corrosion (K5)
CO4	Design and formulate the anticorrosive paint by choosing pigments, binders and additives for corrosion prevention. (K5)
CO5	Analyze the recent developments in corrosion protection materials etc. (K4)
CO6	Discover surface preparation and paint application methods (K4)

**Mapping of Course Outcomes (COs) with Programme Outcomes (POs)**

		PO 1	PO 2	PO 3	PO4	PO5	PO 6	PO7	PO 8	PO9	PO1 0	PO1 1	PO1 2	PSO 1	PSO 2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4
CO1	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

CO2	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO3	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO4	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
CO5	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO6	K4	3	2	2	2	2	1	1	2	2	3	2	3	3	2
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	<b>Course Code:</b> PST1713	<b>Course Title:</b> Honors Course V- Sustainability of polymers			<b>Credits = 3</b>		
	<b>Semester: VIII</b>	<b>Total Contact Hours: 45</b>			<b>L</b>	<b>T</b>	<b>P</b>
<b>List of Prerequisite Courses</b>							
Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymer analysis and characterization of resins and polymers lab, Environment Health and Safety of Polymers and Coating, Evaluation and testing of Polymers and Coatings .							
<b>List of Courses where this course will be Prerequisite</b>							
Project II							
<b>Description of relevance of this course in the B. Tech. Programme</b>							
Able to understand the sustainability approach in polymer and coating industry							
<b>Sr. No.</b>	<b>Course Contents (Topics and subtopics)</b>						<b>Required Hours</b>
1	Polymers and Environment – Environmental issues related to polymer industries, design for environment life cycle approach, contribution to energy, feedstock, transport, gross and net calorific value. Effect of plastic waste on wildlife, aquatic life and water pollution, Positive impact of plastic on the environment. Effluent treatment at latex and rubber industries.						10
2	Polymers in agriculture – Greenhouse films, Plastics in mulch films, plastics in silage, drip irrigation systems. Polymers in packaging – Common packaging						7
3	Sustainability approach in paint – decorative paint, automotive paint, industrial coatings						10
4	Recycling – Polyethylene terephthalate and styrene based polymers, disposal of waste plastics films. Energy recovery from waste polymer products. Disposal of plastic goods, Reuse and recycling of household plastic, recycling of e-waste, disposal and recycling of biodegradable plastics and food waste, biogas production, and production of cooking gas from waste plastics. Tyre recycling, recycling of dipped goods and non-tyre products.						6
5	Flammability of polymers – Release of polymer vapours, ignition, combustion of polymer vapours. Fire propagation, fire-resistant polymers. Methods to improve the fire resistance of polymers. Carcinogenic polymers and rubber chemicals.						10
<b>Total</b>						<b>45</b>	
<b>List of Text Books/ Reference Books</b>							
1	Handbook of Sustainable Polymers Structure and Chemistry by Edited By Vijay Kumar Thakur, Manju Kumari Thakur ,2016						
2	Advances in Sustainable Polymers Processing and Applications by Vimal Katiyar, Amit Kumar, and Neha Mulchandani, Nov 14, 2019\						
3	Recycling of Polymers: Methods, Characterization and Applications By Raju Francis, 7 October 2016						
<b>Course Outcomes (Students will be able to.....)</b>							
CO1	Identify and explain the effect of plastic waste (K3)						
CO2	Design the sustainable approach for polymers and coatings (K6)						
CO3	Compare various approaches of recycling of polymers (K5)						
CO4	Select the polymer chemistry for agriculture application (K4)						
CO5	Fire resistance and flammability of polymers						

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4
CO1	K3	3	3	3	3	2	2	3	3	3	3	3	2	3	3
CO2	K6	3	2	3	3	3	3	3	3	3	3	3	2	3	3
CO3	K5	3	3	3	3	2	2	3	3	2	3	3	2	3	3
CO4	K4	3	3	2	3	2	3	3	2	3	3	3	3	3	3

CO5	K3	3	3	3	2	3	3	3	3	3	3	3	3	3	3
Course	K6	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; – No Contribution  
K, knowledge level from cognitive domain; A, Affective domain; S, Psychomotor domain

PCC	Course Code:	Course Title: Project – II (Experiments)	Credits = 3		
	Semester: VIII		Total Contact Hours: 90	L	T
			0	0	12
<b>List of Prerequisite Courses</b>					
Project – I					
<b>List of Courses where this course will be prerequisite</b>					
Relevant courses in previous courses (Sem. I to Sem. VII)					
<b>Description of relevance of this course in the B. Tech. Program</b>					
The course is designed to develop skills necessary for executing and solving a unique research problem in Pharmaceutical Sciences and Technology field. After the laboratory work, the findings of the research are presented in a coherent manner, which may result in a patent, publication and/or presentation.					
	<b>Course Contents (Topics and Subtopics)</b>				<b>Required Hours</b>
1	The topic of the research with clearly defined Objectives and Hypotheses should be explored systematically, in a scientifically planned rational set of experiments. Students should have actual experimental data collected on the chosen research topic.				60
2	Oral presentation of the proposed research work with data generated during actual laboratory work along with computational studies, if any, targeted towards fulfilling the objectives. The outcome is submitted in the form of a report.				30
	<b>Total</b>				<b>90</b>
<b>List of Textbooks/Reference Books</b>					
1	Relevant review articles, research papers, patents, book chapter, books, etc.				
<b>Course Outcomes (Students will be able to.....)</b>					
<b>CO1</b>	Perform experiments & troubleshoot to generate reliable data (K5)				
<b>CO2</b>	Apply different statistical tools for scientific data analysis (K4)				
<b>CO3</b>	Evaluate critically the experimental data and draw meaningful inferences (K5)				
<b>CO4</b>	Develop skills to communicate the research outcome effectively (K6)				
<b>CO5</b>	Develop skills for writing a complete document on the project work (K6)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+P	K3	K4
<b>CO1</b>	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
<b>CO2</b>	K4	3	3	2	3	2	3	3	3	2	3	3	2	3	3
<b>CO3</b>	K5	3	3	3	3	3	0	3	3	3	3	3	3	3	3
<b>CO4</b>	K6	3	3	3	3	3	3	1	3	3	3	3	2	3	3
<b>CO5</b>	K6	3	3	3	3	3	3	3	3	3	3	3	3	3	3
<b>Course</b>	K6	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution  
K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain



PEC	<b>Course Code:</b> SCP1801	<b>Course Title:</b> Pr 9 – Analysis and Testing of Paints	<b>Credits = 2</b>		
			<b>L</b>	<b>T</b>	<b>P</b>
	<b>Semester: VIII</b>	<b>Total Contact Hours: 60</b>	<b>0</b>	<b>0</b>	<b>4</b>

**List of Prerequisite Courses**

Polymer science and Technology, Polymer chemistry and Technology, Technology of Thermoset polymers, Analysis and characterization of resins and polymers lab

**List of Courses where this course will be Prerequisite**

Adhesion and adhesives

**Description of relevance of this course in the B. Tech. Program**

To give understanding of industrial manufacturing processes, properties and applications, processing of various types of paints. Knowledge of subject will help student to carry out research and development in the areas of paints and coatings, coating formulation development, setting up a paint industry and plant, basics of research and development, etc. To make aware of Environmental concerns of paints and coatings eg. release of VOCs and the effect of VOCs on the environment.

Sr. No.	Course Contents (Topics and subtopics)	Required Hours
1	Analysis of Linseed Oil (IV, Sap Value, color, Refractive Index, Viscosity)	1x4h/week
2	Analysis of A Synthetic Enamel (Black, Red, White)	
3	Zinc Chrome Primer, Red Oxide , Primer, Intermediate Coat, (NVM, Viscosity, WPL, Grind, Hiding, Drying Time, Scratch Hardness, Impact Test, Flexibility, Gloss, Dry Film Thickness, Acid, Alkali, and Water Resistance, Adhesion, Corrosion Resistance By Salt Spray Humidity Cabinet, Accelerated Exposure Of Paints In QUV And Atlas Apparatus	
4	Analysis of Emulsion Paint (NVM, % Solids, Scrub Resistance, Stain Resistance) Analysis of Architectural Paints, Plastic Emulsion Paint and Distemper	
5	Color Matching Of Synthetic Enamel.	
6	Analysis of Pigments (Solvent Bleed in about 10 Different Solvents, Resistance to acids, alkalis, light)	
<b>Total</b>		<b>60</b>

**List of Text Books/ Reference Books**

1	Polymer Chemistry: A Practical Approach (The Practical Approach in Chemistry Series) 1st Edition Fred J. Davis Oxford University Press 2004
2	A Practical Course in Polymer Chemistry S. H. Pinner, Borough Polytechnic, London, Pergamon Press, he., New York, 1961
3	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994
4	Polymer Science by Gowariker, John Wiley and Sons 1986.
5	Encyclopedia of Polymer Science and Technology, Johan Wiley and Sons, Inc 1965.
6	Encyclopedia of Polymer Science and Engineering, Johan Wiley and Sons, Inc 1988.
7	PVC Technology, A. S. Athalye and Prakash Trivedi, Multi-Tech Publishing Co, 1994

**Course Outcomes (Students will be able to.....)**

CO1	Analyze the linseed oil and some oil samples to determine acid value, iodine value etc. (K4)
CO2	Characterize the given paint for its properties such as Mechanical, Liquid Properties etc. (K4)
CO3	Characterize given emulsion paint. (K4)
CO4	Analyze different Pigments' Properties. (K4)
CO5	Perform color matching (K5)

**Mapping of Course Outcomes (COs) with Programme Outcomes (POs)**

		PO 1	PO 2	PO 3	PO4	PO5	PO 6	PO7	PO 8	PO9	PO1 0	PO1 1	PO1 2	PSO 1	PSO 2
		K3	K4	K6	K5	K6	K3	K3+P	K3	K3+A	K2+ A	K3	K6+A +Psy	K3	K4
CO1	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3

CO2	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO3	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO4	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	3
CO5	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3
Course	K5	3	3	3	3	3	3	3	3	3	3	3	3	3	3

<b>OJT</b>	<b>Course Code:</b> <b>PHP1451</b>	<b>Course Title:</b> <b>Internship with Industry</b>	<b>Credits = 12</b>		
	<b>Semester: VIII</b>	<b>Total Contact Weeks: 12-16</b>	<b>L</b>	<b>T</b>	<b>P</b>
			<b>0</b>	<b>0</b>	<b>0</b>
<b>List of Prerequisite Courses</b>					
None					
<b>List of Courses where this course will be prerequisite</b>					
Project – I (PHP1074), Project – II (PHP1075)					
<b>Description of relevance of this course in the B. Tech. Program</b>					
The course is designed to –					
1. develop a systematic thinking about an industrial problem;					
2. develop skills for communication, networking, personal grooming & professional conduct within an industrial environment, and					
3. develop the attitude for individual and teamwork.					
	<b>Course Contents (Topics and Subtopics)</b>				<b>Required Weeks</b>
1	Each Student will be involved in R & D/manufacturing (QA/QC/Plant Engineering /Stores and Purchase)/marketing/finance/consultancy/Technical services/ Engineering/Projects, etc., as deemed necessary by the assigned/chosen industry.  Oral presentation & written report of the in-plant training will be evaluated along with industry feedback.				12
	<b>Total</b>				<b>12</b>
<b>Course Outcomes (Students will be able to.....)</b>					
<b>CO1</b>	Apply the concept of project & production management in further planning (K3)				
<b>CO2</b>	Develop critical thinking regarding the various operations involved in dyestuff technology and allied industry (K4)				
<b>CO3</b>	Solve certain industrial challenges in dyestuff technology and allied field (K6)				
<b>CO4</b>	Present and communicate an industrial problem effectively (K6)				
<b>CO5</b>	Write a scientific report on the training (K6)				

<b>Mapping of Course Outcomes (COs) with Programme Outcomes (POs)</b>															
		<b>PO1</b>	<b>PO2</b>	<b>PO3</b>	<b>PO4</b>	<b>PO5</b>	<b>PO6</b>	<b>PO7</b>	<b>PO8</b>	<b>PO9</b>	<b>PO10</b>	<b>PO11</b>	<b>PO12</b>	<b>PSO1</b>	<b>PSO2</b>
		K3	K4	K6	K5	K6	K3	K3+S	K3	K3+A	K2+A	K3	K6+A+S	K3	K4
<b>CO1</b>	K3	3	3	2	2	2	3	3	3	3	3	3	2	3	3
<b>CO2</b>	K4	3	3	2	3	2	3	3	3	3	3	3	2	3	2
<b>CO3</b>	K6	3	3	3	3	3	3	2	3	1	3	2	3	3	3
<b>CO4</b>	K6	3	3	2	3	3	3	3	0	3	3	3	3	2	3
<b>CO5</b>	K6	3	3	3	3	1	3	3	3	3	2	3	3	3	3
<b>Course</b>	K6	3	3	3	3	3	3	3	3	3	3	3	3	3	3

3, Strong Contribution; 2, Moderate Contribution; 1, Low Contribution; 0, No Contribution

K, Knowledge level from cognitive domain; A, Affective domain; P, Psychomotor domain