INSTITUTE OF CHEMICAL TECHNOLOGY Ordinances, Regulations and Syllabi relating to the Degree of Master of Technology in Bioprocess Technology (M. Tech. Bioprocess Technology) 2018-2019

1. Introduction

The Institute is revamping its academic structure especially for the master's courses by way of introducing the compulsory industrial training for a period of six months (to be taken in the third semester of the program). The number of credits in the first two semesters has also been increased and a research component has been included. The total credits in the first two semesters now stand at 27 each instead of earlier 21. All the courses will continue to be credit based and the evaluation will be grade based.

The Departmental administrative committee and academic program committee periodically proposed the program outcomes having consistency with the graduate attributes available with NBA. The committee critically analyzed information obtained from graduated students, employers and immediately passed out students. The program outcomes are as follows:

SR. NO.	PROGRAM OUTCOMES (POS)
1	The graduates will be able to apply knowledge of basic sciences (Mathematics, Physics, Chemistry, Biochemistry, Microbiology, Biology and Chemical Engineering Sciences) and applied engineering courses in getting solutions to issues pertaining to biotechnology, biochemical, biopharmaceutical and allied industries.
2	The graduates should be able to systematically break up complex processing problems in realizable steps and solve them.
3	The graduates will be able to design and develop a process, a product or a component of a biotech system or provide an engineering and technological solution for a specific task within realistic constraints
4	The graduates will be able to design and conduct experiments as well as analyze and interpret data.
5	The graduate will be able to use modern tools, software, equipment etc. to analyze and obtain solution to the problems.
6	The graduates will be able to study the impact of bioprocess industry in the global, economic, and societal context
7	The graduates should practice their profession considering environmental protection and sustainability
8	Graduates are expected to practice professional skills in an ethical manner
9	The graduates should have competence to undertake designated task on individual or team basis as per the requirement.
10	The graduates will be able to communicate effectively their points of view
11	The graduates will acquire attitude for life- long learning
12	The graduates should actively participate in project and financial management
SR.	PROGRAM SPECIFIC OUTCOMES (PSOs)
NO.	
13	Graduates will be acquainted with the latest development in different fields of bioprocessing so as to enable them to take up higher studies, research & developmental work
	Graduates will be introduced to industrial bioprocessing and technology managerial subjects

Credit system is a systematic way of describing an educational programme by attaching credits to its components. The definition of credits may be based on different parameters, such as student workload, learning outcomes and contact hours. It is a student-centric system based on the **student workload** required to achieve the objectives of a programme. It should facilitate academic recognition of the courses and mobility of the students. Credits assignment is based on the principle that Credits can only be obtained after successful completion of the work required and appropriate assessment of the learning outcomes achieved. As per the AICTE norms 2L/week of lectures are 2 credits, while 2h/week of practical//seminar/literature review/research work are 1 credit. This has been taken as the basis during the working of the proposed syllabus.

Student workload consists of the time required to complete all prescribed learning activities such as attendance at lectures/practical, seminars, projects, etc. Credits are allocated to all the educational components of a study programme and indicate the quantity of work each component requires to achieve its specific objectives.

Evaluation is an important component of any teaching-learning process. The Institute gives emphasis on continuous evaluation with considerable freedom to the teacher in deciding the mode of evaluation of the students. The performance of the student is documented by a **grade** at the end of the semester. The grading scale ranks the students on a statistical basis. Therefore, statistical data on student performance is a prerequisite for applying the grading system.

2. Course Credits

In general, a certain quantum of work measured in terms of **credits** is laid down as the requirement for a particular degree. The student acquires credits by passing courses every semester, the amount of credit associated with a course being dependent upon the number of hours of instruction per week in that course.

There are mainly two types of courses in the Institute - lecture courses and laboratory courses. Lecture courses consist of lecture (L) and tutorial (T) hours. Laboratory courses consist of practical (P) hours. The credit (C) for a course is dependent on the number of hours of instruction per week in that course, as given below:

- (1) 1h/week of lecture (L) or tutorial (T) = 1 credit
- (2) 2h/week of Practical (P) = 1 credit
- (3) Credit (C) for a theory course = No. of hours of lectures per week +
 - No. of hours of tutorials per week = L + T
- (4) Credits (C) for a Laboratory course/Seminar/research work =

 $\frac{1}{2}$ x No. of hours per week

Credits will be assigned to In-plant, Seminar, Projects and other mandatory course requirements also and these will be mentioned in the respective syllabi. There may be some non-credit requirements. A student is required to earn credits as mentioned in the syllabus.

3. Evaluation

3.1 The weightage of different modes of assessments shall be as under.

	In-Semester evaluation		E d	
	Continuous mode	Mid Semester- Exam	End- Semester- Exam	Components of continuous mode
Theory	20%	30%	50%	Quizzes, class tests (open or closed book), home assignments, group assignments, <i>vivavoce</i> assignments, discussions
Practical	50%	-	50%	Attendance, <i>viva -voce</i> , journal, assignments, project, experiments, tests
Seminar/ critical review/ Research work	-	-	100%	Continuous evaluation not applicable, End semester evaluation will be based on written report evaluation and presentation in front of the external examiner within the Department

3.2. In-Semester Evaluation:

- (a) It is expected that the teacher would conduct at least two assessments (in any form as quizzes, tests, homework, group work etc.) under the continuous mode in a Semester.
- (b) The teacher will announce at the beginning of the respective course the method of conducting the tests under the continuous mode and the assignment of marks
- (c) In-semester performance of all students should be displayed and sent to the academic office by the teacher at least 15 days before the end-semester examination.
- (d) For the theory courses, there will be one mid-semester test for each course to be held as per the schedule fixed in the Academic Calendar.
- (e) For mid –semester examinations in theory papers, duration of examination will be 1 hour for 3 credit courses and 2 hours for 4 credit courses

3.3. End-Semester examination:

- a) The semester end examination will cover the full syllabus of the course and will be conducted as per the Institutional timetable at the end of each semester.
- b) For end –semester examinations in theory papers, duration of examination will be 1 hour for 3 credit courses and 2 hours for 4 credit courses
- c) For the end semester evaluation of seminar/research work, student will be expected to submit a written report and make a presentation. The evaluation will be based on the quality of the written report and presentation.

3.4 Passes and Fail

- (a) The candidates who obtain 40% and more marks of the total marks of a course head shall be deemed to have **passed** the respective course head.
- (b) The candidates who obtain marks less than 40% of the total marks of a course head shall be deemed to have **failed** in the respective course head (**Grade FF**).

3.5 Grades:

- (a) The performance of a student shall be documented by a **Letter grade**. Each letter grade has a **Grade point** associated with it. The Grades and Grade points shall be assigned to each head of passing and both will be indicated in the mark-list of the semester examination.
- (c) The total marks (in-semester + end-semester) of a candidate in a subject head are converted into a letter grade, based on the relative (and sometimes the absolute) performance of the student.

Letter Grade	Grade Point
AA	10
AB	9
BB	8
BC	7
CC	6.5
CD	6
DD	5.5
EE	5

- (d) For granting class, a grade point of 6.0 and above will be considered equivalent to First class.
- (c) The grades to be allotted in the case of students who fail or do not appear at the end-semester examination shall be as under.

Letter Grade	Grade Point	Explanation
FF	0	The candidate fails in course head. The candidate will be allowed to take end semester repeat or subsequent examinations as per rule.

XX		The candidate has not kept term for the course head due to attendance less than requisite. Further see 3.5(g) below. In the above cases, the candidate has to repeat the respective course by paying the fees.
I	0	The candidate has kept term for the course head, has taken all the internal examinations with satisfactory performance, but has failed to take the end- semester examination or repeat examination due to genuine reasons. The candidate will be allowed to take end-semester repeat or subsequent examinations as per rule.
FR	0	The candidate has exhausted all the permissible chances to clear the end- semester examinations. The candidate has to register for the respective semester again for all the subject heads or will be out of the respective degree course as per the rules.
DR	0	 (i) The candidate hasn't participated in academic program. (ii) The candidate has taken a drop for the subject head; - provided he/she intimates the same (i or ii) at least 7 days in advance of the commencement of the end-semester examination for the respective year.

- (d) Grades **FF** and **I** are place-holders only and do not enter into CPI/SPI calculations directly. These grades get converted to one of the regular grades after the end-semester examination.
- (e) A candidate with an **FR** grade is not eligible for any repeat examination in that course and has to re-register for that semester by paying the appropriate fees.
- (f) I grade will not be continued beyond the permissible number of end-semester/repeat examinations.
- (g) 'XX' Grade: The grade XX in a course is awarded if (i) candidate does not maintain the minimum 75% attendance in the Lecture/Tutorial/Practical classes, (ii) candidate receives less than 20% of the combined marks assigned for continuous assessment and mid-semester examination, and (iii) candidate indulges in a misconduct/uses unfair means in the examination, assignments, etc., of a nature serious enough to invite disciplinary action in the opinion of the teacher.

(h) The names/roll numbers of students to be awarded the **XX** grade should be communicated by the teacher to the Academic office as per academic calendar before the last date of submission of the application for end semester examination.

3.6. Awarding the grades

The grading scale ranks the students on a statistical basis on the basis of the overall performance of the students of a given class in the given course head. Therefore, statistical data on students' performance is a prerequisite for applying the grading system. While assigning grades in a given course head, it is essential to know the **average marks** (**AM**) obtained by the students *who have passed the subject head* and the **highest marks** (**HM**) obtained in the *same subject head*.

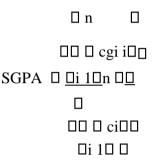
- **3.6.1.** If the **average marks** (AM) obtained by the students *who have passed the subject head* is <60%, the interval AM shall be awarded grade CC and the other grades shall be decided as follows:
- (i) AA, AB, BB, and BC grades shall be decided between the AM and HM by dividing the range in equal intervals.
- (ii) CD, DD and EE grades shall be decided between the AM and minimum marks required for passing the head (i.e. 40%) by dividing the range in equal intervals.
- **3.6.2.** If the **average marks** (**AM**) obtained by the students *who have passed the subject head* is such that $60\% \le AM < 70\%$, the interval AM shall be awarded grade BC and the other grades shall be decided as follows:
- (i) AA, AB, BB grades shall be decided between the AM and HM by dividing the range in equal intervals.

⁽Note: Award of the XX grade in the case of g(iii) above shall be done by Disciplinary Action Committee (DAC)).

- (ii) CC, CD, DD and EE grades shall be decided between the AM and minimum marks required for passing the head (i.e. 40%) by dividing the range in equal intervals.
- **3.6.3.** If the **average marks** (AM) obtained by the students *who have passed the subject head* is \geq 70%, the interval AM shall be awarded grade BB and the other grades shall be decided as follows:
 - (i) AA and AB grades shall be decided between the AM and HM by dividing the range in equal intervals.
 - (ii) BC CC, CD, DD and EE grades shall be decided between the AM and minimum marks required for passing the head (i.e. 40%) by dividing the range in equal intervals.

4. SPI and CPI

(a) Semester Performance Index (SPI): The performance of a student in a semester is indicated by Semester Performance Index (SPI), which is a weighted average of the grade points obtained in all the courses taken by the student in the semester and scaled to a maximum of 10. (SPI is to be calculated upto two decimal places.) A Semester Grade Point Average (SGPA) will be computed for each semester as follows:



Where

'n' is the number of courses for the semester,

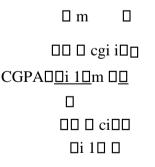
'ci' is the number of credits allotted to a particular course, and

 ${}^{\circ}g_{i}{}^{\circ}$ is the grade-points awarded to the student for the course based on his performance as per the above table.

SGPA will be rounded off to the second place of decimal and recorded as such.

(b) Cumulative Performance Index (CPI): An up to date assessment of the overall performance of a student from the time he entered the Institute is obtained by calculating Cumulative Performance Index (CPI) of a student. The CPI is weighted average of the grade points obtained in all the courses registered by the student since he entered the Institute. CPI is also calculated at the end of every semester (up to two decimal places).

Starting from the first semester at the end of each semester (S), a Cumulative Grade Point Average (CGPA) will be computed as follows:



Where

'm' is the total number of courses from the first semester onwards up to and including the semester S,

'c_i' is the number of credits allotted to a particular course, and

 g_i is the grade-points awarded to the student for the course based on his performance as per the above table. CGPA will be rounded off to the second place of decimal and recorded as such.

(c) The CGPA, SGPA and the grades obtained in all the subjects in a semester will be communicated to every student at the end of every semester / beginning of the next semester.

(d) **When** a student gets the grade 'FF', or I' in any subject head during a semester, the SGPA and CGPA from that semester onwards will be tentatively calculated, taking only 'zero' grade point for each such 'FF' or 'I' grade. When the 'FF' grade(s) has / have been substituted by better grades after the repeat examination or subsequent semester examination, the SGPA and CGPA will be recomputed and recorded.

5. Repeat End-Semester Examination

- **5.1.** For those candidates who fail in a subject head or are eligible for appearing at the repeat examination, **Repeat End-Semester Examination** will be conducted within one month from the declaration of the results of regular end-semester examination, as per **Regulation R.14**.
- **5.2.** The marks obtained by candidates in the in-semester examinations (continuous assessment and Mid Semester Examination) will be carried forward in such cases.
- **5.3. Grading the performance in the Repeat Examination:** The grades will be assigned as per 3.5 and 3.6 above. However, for a candidate taking any repeat examination or subsequent regular semester examination or performance improvement examination shall be awarded **one grade lower** than that decided on the basis of the actual marks obtained; provided 'EE' grade obtained in such an examination shall remain 'EE'. For reference see the table below.

Grade obtained in repeat or subsequent end-semester examination	Grade to be assigned	Grade point
AA	AB	9.0
AB	BB	8.0
BB	BC	7.0
BC	CC	6.5
CC	CD	6.0
CD	DD	5.5
DD	EE	5.0
EE	EE	5.0

5.4. Revaluation of end-semester and repeat examination: Candidate's performance in these examinations will be displayed on proper notice board and after 3 days of such display the marks will be sent to the Academic Office. No revaluation of these examinations will be allowed.

6. Passing of a Semester examination

A candidate shall be declared as 'PASSED' any semester examination if he/she has

- (a) Cleared all heads of passing by securing grades EE or higher in all the heads;
- (b) Passed all the heads of passing such as project, seminar, training, etc as per the rules;
- (c) Satisfactorily completed all the mandatory requirements of the course;
- (d) paid all the Institute dues;
- (e) No case of indiscipline pending against him/her.

7. Eligibility for the Award of a Degree

A candidate shall be declared eligible for the award of a degree, if he/she has cleared all the semester examinations as given in (6) above.

8. Allowed to keep terms (ATKT)

8.1 A candidate who has I grade in one or more heads of passing of an odd semester of an academic year shall be allowed to keep terms for the respective even semester.

8.2. A candidate shall be allowed to keep terms for the subsequent academic year if he/she has FF or I grade in not more than two heads of passing from all the heads of passing of the two terms of the previous academic year taken together. Such a candidate shall be declared as **FAILED**, **ATKT**.

9. Repeating a course

9.1 A student is required to repeat the course under the following situations:

- (a) A student who gets an XX, FR, or DR grade in a course; or
- (b) A student has exhausted all permissible chances to clear the course.
 - **9.2** A candidate from first year who remains absent for the regular end-semester examination of a semester and the corresponding repeat examination for **ALL SUBJECTS** shall have to take fresh admission for the corresponding year; unless the candidate has dropped out / terminated from the course.
 - **9.3** If a candidate at the Second, fails to pass any semester examination in not more than 4 consecutive examinations, including the repeat examinations, from the date of registering for the respective year, the candidate shall have to take readmission for the corresponding year again in which the failure has occurred, provided the course is not changed.

10. Improvement of performance

A candidate will be allowed to appear at the **entire examination** after the regular end-semester examination as per the respective rules to improve the performance. In such a case if the result of the examination repeated -

1. Is better than the previous one, the previous result shall be declared null and void; and 2. Is worse than the previous one, the result of the subsequent examination shall not be declared.

3. However, awarding of final grade will be made under the provision of sub clause 5.3 above.

11. Exit rules for poorly performing students

A candidate shall be excluded from a course under the following conditions:

- (a) If he/she fails to pass any semester examination of the any year of the course in not more than four consecutive attempts (Examination conducted by Institute) from the date of joining the course.
- (b) If he/she does not keep two consecutive terms without giving any reasonable justification (as prescribed by the institute) for doing so.
- (c) If a candidate fails to fulfill all the requirements of his/her respective degree within the prescribed period from the date of taking admission to the course, the candidate shall be excluded from the course.

12. Miscellaneous

- (a) Although CPI will be given in the Semester grade report, the final degree certificate will not mention any **Class** whatsoever.
- (b) Not with standing anything said above if a course is revised /restructured then transient provisions applicable at the time of revision /restructuring shall be applicable.

Syllabus Details for the degree of Master of Technology (Bioprocess Technology) Program

Subject	Subject			Ir/Week		Marks			
code		Credit	L	Т	Р	Continuous Assessment	Mid-semester Examination	Final Examination	Tota
				SI	EME	STER I			
BST 2101	Core I: Bioreaction Engineering	3	2	1	0	10	15	25	50
BST 2102	Core II: Unit Operations in Bioprocessing	3	2	1	0	10	15	25	50
CET 2002	Core III: Transport phenomenon	3	2	1	0	10	15	25	50
BST 2103	Core IV: Biocatalyst Engineering	3	2	1	0	10	15	25	50
	Elective I	3	2	1	0	10	15	25	50
	Elective II	3	2	1	0	10	15	25	50
BSP 2101	Bioprocess Engineering Laboratory	3			6	25	-	25	50
BSP 2102	Seminar and Critical Review	3			6	-	-	30 (Report) 20 (Presentation)	50
BSP 2103	Research Project-I	6			12	-	-	60 (Report) 40 (Presentation)	100
	TOTAL:	27	10	5	24				450
				SE	CMES	STER II			•
BST 2105	Core V: Bioprocess Equipment Design and Industrial process Automation	3	2	1	0	10	15	25	50
BST 2112	Core VI: Adsorptive, Chromatographic and Membrane separations	3	2	1	0	10	15	25	50
BST 2104	Core VII: Bioprocess and Biosystem Engineering	3	2	1	0	10	15	25	50
BSE 2111	Core VIII: Industrial biocatalysis	3	2	1	0	10	15	25	50
	Elective III	3	2	1	0	10	15	25	50
	Elective IV	3	2	1	0	10	15	25	50

BSP 2104	Biosciences and Bioprocess Technology Laboratory	3			6	25		25	50
BSP 2105	Research Project-II	9			18	-	-	90 (Report) 60 (Presentation)	150
	TOTAL:	27	10	5	24	-	-	-	450
				S	EMI	ESTERS III			
BSP 2106	Industrial Training (15 weeks to maximum of 6 months)	30	-	-	40			270 (Report) 180 (Presentation)	450
	SEMESTER IV								
BSP 2107	Research Project-III	30	-	-	40	-	-	270 (Report) 180 (Presentation)	450

Note: Semester III and Semester IV Evaluation will be conducted be at end of IV semester.

Brief Overview of Syllabus SEMESTER - I

	Course Code: BST 2101 (Core subject)	Course Title: BST 2101 Bioreaction Engineering (Marks 50)				
	Semester: I	Total contact hours: 30				
Module	Course Contents (Topics and subtopics)					
1	Material and Energy Balance Computations					
2	Basic Biochemistry, Basic Microbiology and Basic Molecular Biology, Principles of biochemical reactions and kinetics					
3	Thermodynamics of bioreactions and biotransformation's					
4	Unstructured and simpl morphologically structured	e structured models, Mechanistic models and l models				

	Course Code: BST 2102 (Core subject) Semester: I	Course Title: BST 2102Unit operations in Bioprocessing (Marks 50)Total contact hours: 30
Module Course Contents (Topics and su		Course Contents (Topics and subtopics)

1	Downstream Processing in Biotechnology, Selection of unit operation with due consideration of physical, chemical and biochemical aspect of biomolecules, basic review of bioprocess designing.
2	Primary separation and recovery processes: Cell disruption methods for intracellular products, removal of insolubles, biomass (and particulate debris) separation techniques, flocculation and sedimentation, centrifugation and filtration methods.
3	Enrichment operations: Membrane – based separations (micro and ultrafiltration, precipitation methods, extractive separation, aqueous twophase extraction, supercritical extraction, insitu product removal, integrated bioprocessing.

	Course Code: CET 2002 (Core Subject)	Course Title: CET 2002 Transport Phenomenon (Marks 50)			
	Semester: I	Total contact hours: 30			
Module	Course Contents (Topics and subtopics)				
1	Basic laws of one-dimensional diffusive transport: momentum, heat and mass transfer and their analogies; characteristics of transport processes;				
2	Flow equation, simple shear flow and developing flows.				
3	Convective transport; Transport in turbulent conditions.				
4	Non-steady state transport; Transport phenomena in bioprocesses and biosystem: interphase, diffusion in biofilm-floc, determination of transport coefficients, agitation power, and evaluation of oxygen transport rate as a function of operating variables.				
5	Introduction to microfluidics in bioprocessing unit operations				

	Course Code: BST 2103 (Core subject)	Course Title: BST 2103 Biocatalyst Engineering (Marks 50)
	Semester: I	Total contact hours: 30
Module	Course Contents (Topics and subtopics)	
1	Biocatalysts, aspects of biocatalytic process design and development, steps from laboratory to industrial scale, applications, global market and societal challenges	

2	Biocatalysis using natural enzymes, microorganisms (bacteria, yeast, fungi), eukaryotic cells (insect, CHO, algae, plant cell cultures), plants, and recombinant enzymes engineered for specific applications, Catalytic activity of biomolecules – enzymes and ribozymes; Enzyme applications: Hydrolase enzymes – lipases, esterases, proteases etc. with specific examples and mechanism, Lyases – e.g. Aspartase, tyrosine-phenol lyase; Isomerases – e.g. glucose isomerise; Transferases –e.g. aminotransferases, PLP as cofactor; Ligases; Oxidoreductases – dehydrogenases, oxidases, oxygenases, peroxidases.
3	Enzyme structure-function relationships, Thermodynamics of protein folding and substrate binding, , Enzyme kinetics and modes of inhibition; Regulation mechanisms; Mechanism of enzyme action; Multienzyme systems; Selection and screening of biocatalysts for activity, stability and substrate or product selectivity; Extremozymes – biocatalysts at extremes of temperature, pressure and pH.
4	Biocatalysis versus chemical catalysis; Understanding when to use a biocatalyst for a chemical problem; Advantages/disadvantages of biocatalysts compared to traditional chemical reactions and heterogeneous/ homogeneous catalysis; Mild reaction conditions, excellent stereo- chemo- and regio- selectivity versus substrate specificity, product inhibition, lack of catalysts robustness, cofactor recycling; Isolated enzyme systems and whole cell systems. Free and immobilized enzymes for biocatalysis. Water versus organic solvent; Reactor and process technology: types, mass balances and their modes of operation; Biocatalyst recycling and recovery; Enzyme immobilization.
5	Enzymes in organic synthesis, Enzymes in novel media, Green chemistry, Oxidation catalysis, Catalysis in water, Homogeneous catalysis, Heterogeneous catalysis, Asymmetric catalysis
6	Modern branches of biotechnology as the workhorse for biocatalysts design improvement, Synthetic biology and biocatalyst engineering, Enzyme discovery and metagenomics, enzyme engineering strategies Chassis selection and host cell engineering, Practical enzyme characterisation, Industrial availability in enzyme production, Scale-up challenges, Process economics and sustainability

	Course Code:	Course Title: (Elective I) (Marks 50)	
	Semester: I	Total contact hours: 30	
Candidat subjects	Elective-I (from the list appended) Candidate will have to choose one of the elective subjects offered for that semester from the elective		
subjects. A consolidated list of all the elective subjects is given at the end.			

	Course Code:	Course Title: (Elective II) (Marks 50)
	Semester: I	Total contact hours: 30

Elective-II (from the list appended) Candidate will have to choose one of the elective subjects offered for that semester from the elective subjects. A consolidated list of all the elective subjects is given at the end.

	Course Code:	Course Title: BSP 2101
	BSP 2101	Bioprocess Engineering Laboratory
		(Marks 50)
	Semester: I	Total contact hours: 30
Module	C	ourse Contents (Topics and subtopics)
1	Flow through pipes, coils and fittings. Flow meters, orifice, venturi, rotameter and turbine meter. Flow through packed beds. Two phase flow. Sedimentation. Fluidization. Solid-liquid separation. Mixing. Evaporators. Absorption in a packed column. Adsorption isotherms. Drying characteristics. Study of spray nozzles, impellers, tower packings, dryers, filters, evaporators. Demonstration of some phenomena, particularly in mixing, fluid mechanics, etc.	
2	Absorption with and without chemical reactions in packed columns. Distillation in packed and/or plate column. Spray, packed and mechanically agitated extraction columns. Absorption/ion exchange in fixed beds. Separation by membranes. Flow of non-Newtonian fluids. Dynamics of feedback control systems. Level and pH control. Demonstration of some important phenomena in bioprocess Engineering, notably coalescence, foaming, internal circulations in drops and bubbles, two and three phase fluidization, aggregative and particulate fluidization, mixing, crystallization etc.	
3	Suitable number of experiments from the above list will be performed. In addition to these experiments, students will also undertake demonstration experiments related to advanced analytical instruments such as GC, HPLC, GC-MS, LC-MS, SEM, FTIR, UV-Vis Spectrophotometry, NMR, TEM, ICP, particle size analyzer etc. In this student will work in groups on these instruments to make a report on theory, working principle, standard operating procedure and one case study as well as live demonstration at the end of laboratory session.	

Course Code: BSP 2102	Course Title: BSP 2102 Seminar and Critical Review (Marks 50)
Semester: I	Total contact hours:

The	e Seminar work is concerned with a detailed and critical review of an area of interest to Bioprocess Technology (Upstream and Downstream processing, Product characterization, Molecular biology etc). Typically, the report should contain and will be evaluated based on the following points:
(a)	Introduction: 2 pages maximum,
(b)	Exhaustive review of literature (including figures): $10 - 12$ pages: 50% Weightage
(c)	Critical analysis of the literature and comments on the analysis Critical analysis should also contain quantitative comparison of observations, results, and conclusion amongst the various papers.
2.	Two typed copies of the report on thesis size bond paper (297 mm x 210 mm) are to be submitted to <u>Coordinator</u> on <u>time to be decided by the coordinator</u> . The detailed timetable for the presentation would be communicated.
3.	The report should be prepared using the Times Roman font (size 12) using 1 1/2 spacing leaving 1-inch margin on all sides producing approximately 29 lines per page. The report should be typed on one side of the paper and need not be bound in a hard cover binding. Figures and tables should be shown as a part of the running text. Each figure should be drawn inside a rectangular box of 12 cm width and 10 cm height. The figures must be sufficiently clear and hand drawn figures will be acceptable. Particular care must be taken if a figure is photocopied from source. Each figure must have a sequence number and caption below. Each table must have a sequence number and title at the top.
4.	Name of the student, title of the problem and year of examination must be indicated on the top cover. THE NAME OF THE SUPERVISOR (ONLY INITIALS) MUST APPEAR ON THE BOTTOM RIGHT CORNER OF THE TOP COVER.
5.	The report must be precise. All important aspects of the topic should be considered and reported. The total number of pages, including tables, figures, and references should not exceed 30. Chapters or subsections need not be started on new pages, while getting the report typed.
6.	Typographical errors in the report must be corrected by the student. The student will be discredited for any omission in the report. All the symbols used in the text should be arranged in an alphabetical order and given separately after conclusions.
7.	The list of references should be arranged in alphabetical order of the names of authors. In the text, the reference should be cited with author's name and year. (author – date style) For example:
(i)	The flow pattern in gas-liquid-solid fluidized bed has been reported in the published literature (Murooka et al., 1982). OR
(ii)	Murooka et al. (1982) have measured flow patterns in gas-liquid-solid fluidized beds. The title of the article should also be included. The references must be given in the following standard format.
(a)	Format for listing references of articles from periodicals: Murooka S., Uchida K. And Kato Y., Recirculation Turbulent Flow of Liquid in Gas-Liquid-Solid Fluidised Bed", J. Chem. Engg. Japan, 15, 29-34 (1982).
(b)	Constant R.F.,"Crystallization, Academic Press, New York, pp. 89-90, 1968.
(c)	Format for listing Thesis: Niranjan K., "Hydrodynamic and Mass Transfer Characteristics of Packed

	Columns", Ph.D. (Tech.) Thesis, University of Mumbai, 1983.		
	(d) Format for listing references of Patents in Chemical Abstracts:		
	Cananaush R.M., U.S.Patent 2,647,141, Cf. C.A. 48, 82636 (1954).		
	(e) Format for listing Handbooks, Tables, Symposia etc.:		
	Kumar R and Kuloor N.R., "Formation of Drops and Bubbles", in Advances in Chemical Engineering, Vol.8, T.B. Drew et.al. (Eds.) New York, Academic Press, pp.256-364 (1970).		
((f) Format for listing Private Communications and other categories: Sharma, M.M., Private Communication (1984).		
5	8. Consistency of units should be maintained in the written report. SI systems should be used. [For SI system – Ref: Ind. Chem. Engr., 24, 32, 3 (1983)]. Units used in the literature (if not SI) should be correctly converted.		
9	9. The time allotted for the oral presentation of seminar is 20 minutes: additional 10 minutes are provided for questions and answers.		
1	 INCOMPLETE AND CARELESSLY WRITTEN REPORT IS LIABLE TO BE REJECTED. 		
1	11. The last date for submission will NOT be extended on any grounds whatsoever.		
1	12. There must not be any acknowledgment about the guidance by the faculty in the Seminar.		
]	13. The Seminar will be evaluated on the basis of (i) rational approach to the problem, ii) correctness and completeness of the written text and iii) performance in the oral presentation.		
1	14. Word-to-word copying from the published article is not permitted. Flowery language is not to be used.		
r.	The submitted report will be evaluated by the research guide and an external examiner		
	from the Department/Industry based on the presentation made by the candidate. A		
	suitable combination of the marks for report and presentation will be considered for		
t	the final evaluation.		

Course Code: BSP 2103	Course Title: BSP 2103 Research Project -I (Marks 100)
Semester: I	Total contact hours:
area in consultation with the experimental/analytical/simu the outcomes of the candidat provided for BSP2102 above examiner from the Departme	ncerned with detailed literature review of the assigned research research supervisor, developing an alation protocol and initiate the actual research work. Based on te is expected to submit a report as per similar guidelines e which will be evaluated by the research guide and an external ent/Industry based on the presentation made by the candidate. A marks for report and presentation will be considered for the

SEMESTER II

	Course Code: BST 2104 (Core subject) Semester: II	Course Title: BST 2104 Bioprocess and Biosystem Engineering (Marks 50) Total contact hours: 30
		ourse Contents (Topics and subtopics)
Module		
1	Thermodynamics of Biosystems	
2	Principles of Cellular Metabolism and Principles of Metabolic flux analysis.	
3	Biochemical pathway engineering, Rational manipulation of biosystems through metabolic and genetic engineering techniques to provide new biocatalysts/bioproducts/value added products.	
4		gn of cellular systems: Integration of recombinant sign, as well as bioinformatics and process systems

	Course Code: BST 2105 (Core subject)	Course Title: BST 2105 Bioprocess Equipment Design and Industrial process automation Automation (Marks 50)
	Semester: II	Total contact hours: 30
Module	C	Course Contents (Topics and subtopics)
1	Background of bioreactors, Modeling and Design of bioreactors: batch, fedbatch, and continuous flow types (Airlift bioreactors, Airlift pressure cycle bioreactors, Loop bioreactor, Stirred tank bioreactors, Fluidized bed bioreactor, Packed-bed reactors, Trickle bed bioreactor, Bubble column fermenter, Multiphase bioreactors, Disposable bioreactors and Wave bioreactor).	
2	e e	mpellers. Design, development and scale up of n of antibiotics, enzymes, vaccines, therapeutic

	Course Code: BST 2112	Course Title: BST 2112 Adsorptive, Chromatographic and Membrane
	(Core subject)	Separations (Marks 50)
	Semester: II	Total contact hours: 30
Module	С	course Contents (Topics and subtopics)
1	Introduction, Theory and chemistry of adsorption. Chromatographic Fundamentals: Retention, Band Spreading, Resolution; Dynamics of Chromatography: Basic mass transfer equations, Method of moments, Linear	
	dispersion model, Linear staged models for chromatography; Instrument Requirements for Chromatography: System design, Column packing techniques; Fundamentals of Adsorption: Gibbs adsorption isotherm, Adsorption isotherm models, Local equilibrium theory and solute movement plots;	
2	Preparative Chromatography: Preparative elution, Frontal, Gradient, Displacement chromatography, Optimization; Hydrodynamic design of adsorbent: Particle size, pore size, surface area and pore volume etc. Thermodynamic design of adsorbent: Ligand design through Molecular modeling, retention mechanisms;	
3	exchange and Ion exclusion	y: Reversed phase and hydrophobic interaction, Ion on, Size-exclusion, Group specific and biospecific ical fluid chromatography; Isocratic and Gradient atography.
4	phenomena of species, me not, membranes. Membran Ultrafiltration, Microfiltra	e separation, Membrane Materials, Transport olecular and ionic, in porous or dense, charged or ne separation processes: Reverse Osmosis, tion, Nanofiltration, Dialysis, Electrodialysis, Gas n, Liquid membranes, Membrane modules and

Course Code: BST 2110 (Core	Course Title: BST 2110 Industrial biocatalysis (Marks 50)	
subject) Semester: II	Total contact hours: 30	
Modules	Course Contents (Topics and subtopics)	
1	Protein and enzyme engineering & thermodynamics	
2	Structure activity relationship of enzyme and their	
	modelling/visualization	
3	Modelling of enzyme using MM, QM & hybrid techniques	
4	Molecular stimulations as a tool for rational enzyme design	
5	Chemometrics & QSAR for prediction of enzyme selecytivity	
6	Enzyme in organic synthesis	

7	Enzyme in novel media
8	Green chemistry
9	Catalysis: Oxidation, catalysis in water, homogenous, heterogenous, asymmetric
10	Biocatalysis versus chemical catalysis; Understanding when to use a biocatalyst for a chemical problem
11	Advantage and disadvantage of biocatalyst compared to traditional chemical reactions, mild reaction conditions
12	Substrate specificity versus Excellent stero-chemo-regio selectivity, Product inhibition, Lack of catalyst robustness, co-factor.
13	Recycling of an enzyme
14	Isolated enzyme system & whole cell system
15	Free and immobilized enzymes for biocatalysis
16	Water versus organic solvent
17	Reactor and process technology: types, mass balances and their modes of operation
18	Biocatalyst recycling and recovery
19	Enzyme immobilization

	Course Code:	Course Title: (Elective III) (Marks 50)	
	Semester: I	Total contact hours: 30	
Elective-I (from the list appended) Candidate will have to choose one of the elective subjects offered for that semester from the elective subjects. A consolidated list of all the elective subjects is given at the end.			

	Course Code:	Course Title: (Elective IV) (Marks 50)	
	Semester: I	Total contact hours: 30	
Elective-II (from the list appended) Candidate will have to choose one of the elective subjects offered for that semester from the elective subjects. A consolidated list of all the elective subjects is given at the end.			

	Course Code:	Course Title: BSP 2104
	BSP 2104	Biosciences and Bioprocess Technology
		Laboratory
		(Marks 50)
	Semester: I	Total contact hours: 30
Module	Course Contents (Topics and subtopics)	
1	Technical Microbiology pertaining to strain isolation for pure culture and its maintenance	
2	Technical Biochemistry pertaining to sugars, enzyme activity and kinetics, nucleic acid isolation.	
3	Fermentation and Bioreactions: fermentation of primary and secondary metabolite on shake flask and at fermentor level with control parameters	
5	Downstream processing cons	sisting column packing, column qualification.

Course Code: BSP 2105	Course Title: BSP 2105 Research Project -II (Marks 150)
Semester: I	Total contact hours:
research area in consultation experimental/analytical/simu Based on the outcomes of the guidelines provided for BSP2 guide and an external examin	cerned with detailed literature review of the assigned with the research supervisor, developing an lation protocol and initiate the actual research work. candidate is expected to submit a report as per similar 2102 above which will be evaluated by the research her from the Department/Industry based on the didate. A suitable combination of the marks for report

FOLLOWING IS THE LIST OF ELECTIVE SUBJECTS

	Course Code: BST 2104 (Elective Subject)	Course Title: BST 2104 Analytical Techniques in Bioprocessing (Marks 50)
	Semester: I	Total contact hours: 30
Module	Course Contents (Topics and subtopics)	

1	Qualitative and quantitative analysis of proteins, nucleic acids, polysaccharides and small molecules such as antibiotics, vitamins, natural products etc. Development and application of modern analytical instrumentation.
2	Electrophoretic techniques; Capillary electrophoresis, Gel electrophoresis, PAGE; native, SDS, 2D PAGE, TGGE, DGGE, PULSE, Isoelectric focusing

	Course Code: BST 2112 (Elective)	Course Title: BST 2112 Applied Molecular and Synthetic Biology (Marks 50)
	Semester: I	Total contact hours: 30
Module	(Course Contents (Topics and subtopics)
1	Concept and History of S	Synthetic Biology
2	Basic expression and regu2.1 Gene expression and2.2 Metabolic pathways a	0
3	Natural and advanced Genetic Tool Box for Manipulation of Pathways and Gene Expression3. 1 Extra chromosomal tool: Plasmids, Cosmids3.2 Genomic tools: Homologous recombination, CRISPR-Cas systems3.3 Synthetic expression elements: Promoter, ribosome binding sites3.4 Advance tools for assembly of genetic elements: Gibson assembly 3.5Genome engineering and synthetic cells	
4	 Synthetic Biology approach for production of Biopharmaceuticals 4.1 Anti-malarial Drug : artemisinin production in <i>Saccharomyces cerevisiae</i> 4.2 Glycoengineered microbial strain for glycosylated proteins 4.3 Production of recombinant proteins : Single protein production (SPP) system in <i>Escherichia coli</i> 4.4 Production of sugar based biotechnological molecules: Xylitol, GOS (galacto-oligosaccharides) 	
5		ch for production of Biochemical and Biocatalyst hthetic Pathway for Biotransformation) Microbial Engineering
6	6.1 The Internation (iGEM) foundation and	scientific literature in the area nal Genetically Engineered Machine iGEM competition

	Course Code: BST 2115 (Elective)	Course Title: BST 2115 Introduction to Biopharmaceutical Manufacturing (Marks 50)	
	Semester: I	Total contact hours: 30	
Module	C	ourse Contents (Topics and subtopics)	
1	Introduction to Biopharma	ceutical Manufacturing process	
2	Upstream Operation: Cell	Culture	
3	Upstream Operation: Bioreactors		
4	Upstream Operation: Critical Parameters for batch release		
5	Downstream Operation: Cell Biomass Clarification, Ultrafiltration and microfiltration		
6	Downstream Operation: Virus Inactivation and Chromatography for purifications		
7	C-GMP and Regulatory Control (USFDA, EMEA)		
8	Process Analytical Tools for Batch release		
9	Packaging and filling	Packaging and filling	
10	Data Integrity		

	Course Code: BST 2109 (Elective)	Course Title: BST 2109 Fermentation and Cell Culture Engineering (Marks 50)	
	Semester: II	Total contact hours: 30	
Module	C	ourse Contents (Topics and subtopics)	
1	Overview of cell culture te	chnology and engineering	
2	Cell biology for bioprocess	sing	
3	Cell physiology for process engineering		
4	Medium design for cell culture processing		
5	Cell line development		
6	Stoichiometry, kinetics and data analysis		
7	Cell culture bioreactors, oxygen transfer in bioreactors		
8	Fed batch culture and dynamic nutrient feeding		
9	Cell retention and perfusio	Cell retention and perfusion	
10	Scaling up and scaling dov	vn for cell culture bioreactors	

Course	Course Title: BST 2111
Code:	Patents, Trademarks and IPR (Marks 50)
BST 2111	Tatents, Trademarks and Tr K (Warks 50)
(Core	
subject)	
Semester:	Total contact hours: 30
II	Total contact nours. 50
Modules	Course Contents (Topics and subtopics)
wiodules	Course Contents (Topics and Subtopics)
1	OVERVIEW OF INTELLECTUAL PROPERTY
2	PATENTS: Patent document, How to protect your inventions, Granting
	of patent, Rights of a patent, How extensive is patent protection, Why
	protect inventions by patents, Searching & Drafting of a patent, Filing
	of a patent, The different layers of the international patent system
3	COPYRIGHT : What is copyright, What is covered by copyright, How
	long does copyright last, Why protect copyright,
4	RELATED RIGHTS : What are related rights, Distinction between
-	related rights and copyright, Rights covered by copyright.
5	TRADEMARKS : What is a trademark, Rights of trademark, What
5	kind of signs can be used as trademarks, types of trademark function
	does a trademark perform How is a trademark protected, How is a
	trademark registered, How long is a registered trademark protected for,
	How extensive is trademark protection, What are well-known marks
	and how are they protected
6	GEOGRAPHICAL INDICATIONS: What is a geographical indication,
	How is a geographical indication protected, Why protect geographical
	indications
7	INDUSTRIAL DESIGNS: What is an industrial design, How can
	industrial designs be protected, What kind of protection is provided by
	industrial designs, How long does the protection last, Why protect
	industrial designs
8	NEW PLANT VARIETIES : Why protect new varieties of plants, How
	can new plants be protected, What protection does the breeder get, How
	long do the breeder's rights last, How extensive is plant variety
	protection
9	ENFORCEMENT OF INTELLECTUAL PROPERTY RIGHTS :
	Infringement of intellectual property rights ,Enforcement Measure
10	INTELLECTUAL PROPERTY : Overview of Biotechnology and
	Intellectual Property Biotechnology Research and Intellectual Property
	Rights
11	Case studies of patents in other areas
	r